Stochastic Gene Expression in Systems Biology

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On the menu...

Today

- Overview of Stochastic Gene Expression (Examples from the Literature)
- Stochastic Chemical Kinetics
- Solutions for Simple Stochastic Processes (Transcription)
- ▶ Importance of Population Size
- ▶ Break??
- Moment Computations for Linear Propensities
- Linear Noise Approximation

• Thursday (8:40-10:25)

- Monte Carlo Simulation Techniques
 - * Gillespie (SSA), Tau leaping, Chemical Langevin (SDEs), Slow Scale SSA.
- Density Computations with Finite State Projection Techniques
- Switch and Trajectory Analyses
- Examples

On the menu...

- Friday, July 24--Nikolai Sinitsyn
 - Methods of generating functional in stochastic kinetics: enzymatic reactions and molecular motors
- Monday, July 27--Ilya Nemenman
 - Signal processing in biochemical networks: Fourier transforms, central limit theorem, linear feedback, and all that, Part I
- Tuesday, July 28--Anton Zilman
 - Stochastic view of transport through biological channels
- Wednesday, July 29--Ilya Nemenman
 - Signal processing in biochemical networks: Information theory, noise suppression, form and function, and all that. Part II
- Thursday, July 30--William Ryu
 - Bacterial chemo- and thermo-taxis as examples of stochastic signal processing systems
- Tuesday, August 4--William Ryu
 - Bacterial chemo- and thermo-taxis as examples of stochastic signal processing systems

Markov Chain

Propensity Function (stochastic reaction rate)

Stoichiometry (reaction path)

Master Equation

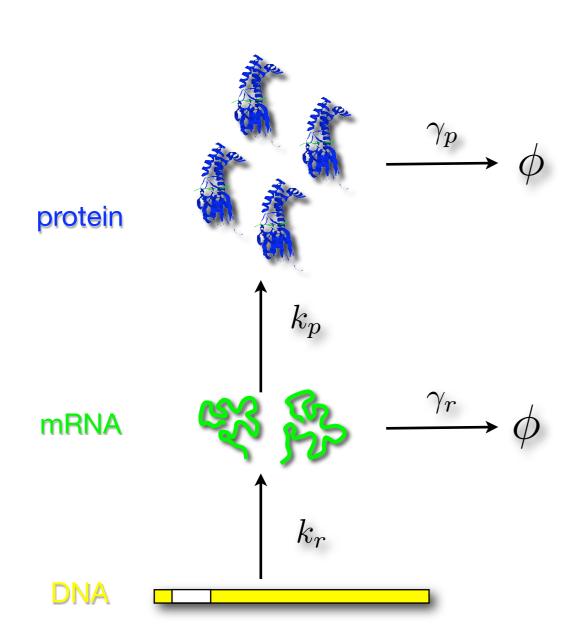
Slides will be made available online as soon as possible

Stochastic Gene Expression: An Overview

Why Are Stochastic Models Needed?

- Much of the mathematical modeling of gene networks represents gene expression deterministically
- Why worry about stochastic models?
 - Randomness
 - Quantization
 - Low copy number
- Experimental evidence indicates that stochastic fluctuations are present
- There are many examples when deterministic models are not adequate

The Central Dogma of Molecular Biology: Modeling Gene Expression



Deterministic model

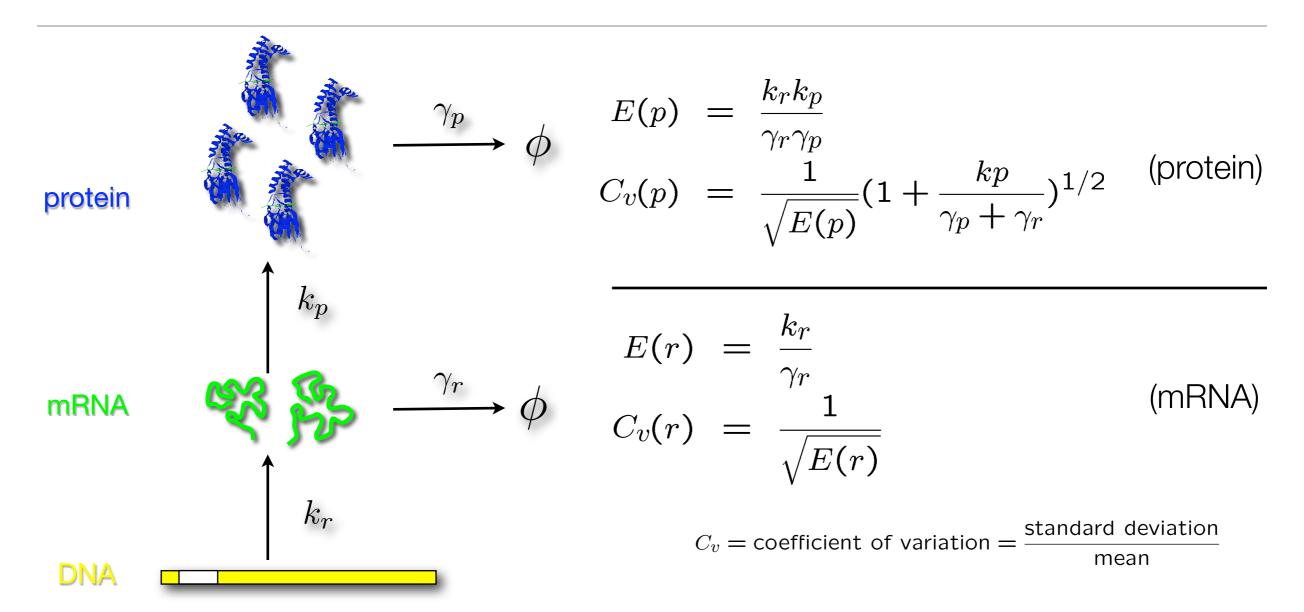
$$\frac{d[mRNA]}{dt} = -\gamma_r[mRNA] + k_r$$

$$\frac{d[protein]}{dt} = -\gamma_p[protein] + k_p[mRNA]$$

Stochastic model

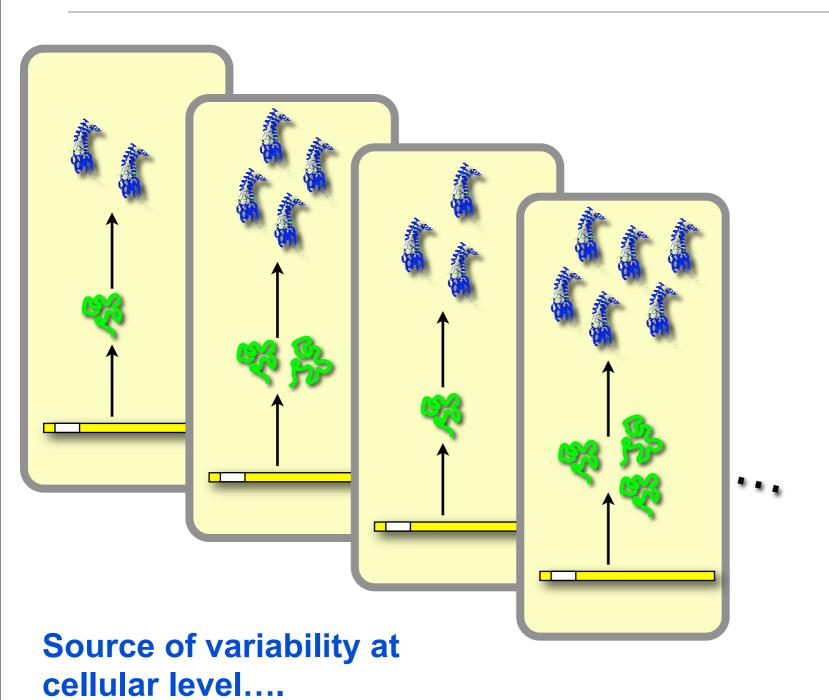
- Probability a single mRNA is transcribed in time dt is $k_r dt$.
- Probability a single mRNA is degraded in time dt is $(\#mRNA) \cdot \gamma_r dt$

Fluctuations at Small Copy Numbers



- ★ Deterministic steady-state equals stochastic mean
- ★ Coefficient of variation goes as 1/√mean
- ★ When mean is large, the coefficient of variation is (relatively) small

Intrinsic Variability in Gene Expression



- Small # of molecules
- Random events

"Intrinsic noise"

Impact of variability

- Noise propagates through the network
- Its amount depends on
 - # of molecules
 - stoichiometry
 - regulation
 - **)**
- Sometimes it is suppressed; other times it is exploited
- Deterministic models are not adequate

Stochastic Influences on Phenotype





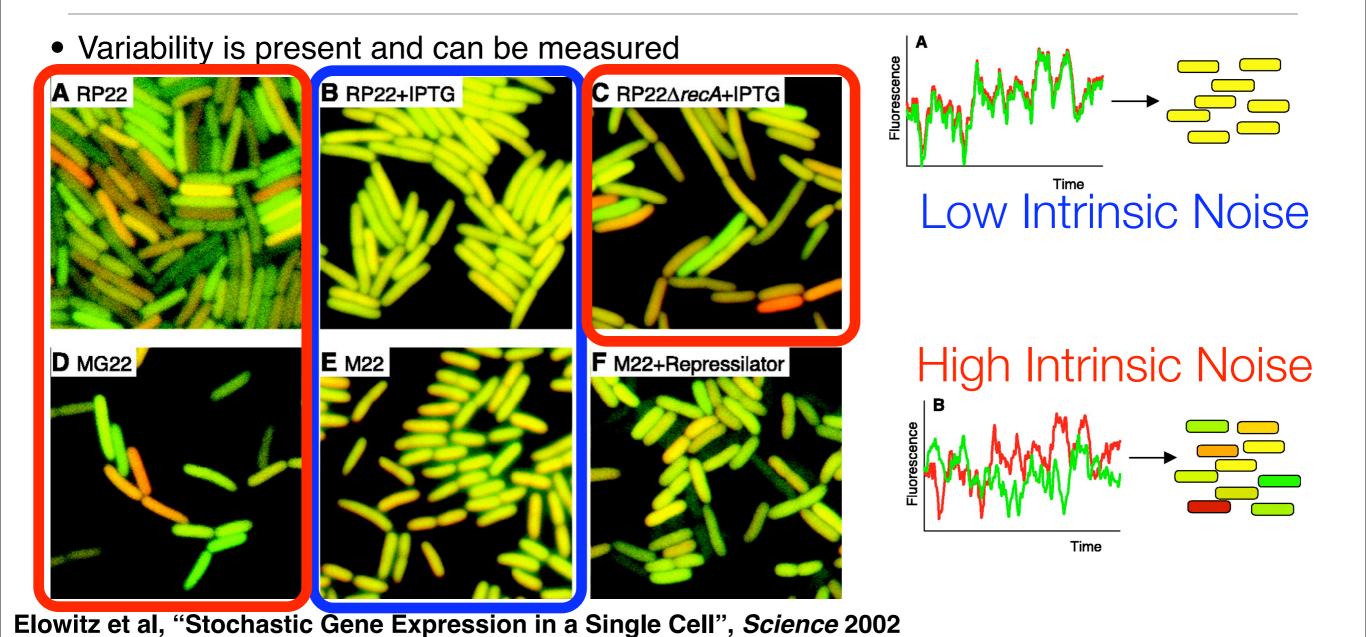


Fingerprints of identical twins

Cc, the first cloned cat and her genetic mother, Rainbow

J. Raser and E. O'Shea, "Noise in Gene Expression: Origins, Consequences, and Control", Science, 2005

We Are Starting to See the "Noise"!

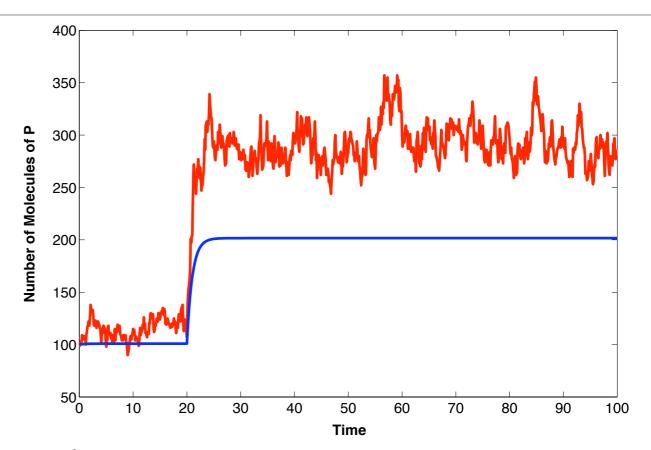


- Inserted two reporter genes on the chromosome (cfp, yfp)
- Each was controlled by the same promoter
- Expression of cfp shown in green, yfp in red

Deterministic Model Fails to Capture Mean

$$\phi \quad \stackrel{k}{\underset{k_a S}{\rightleftharpoons}} \quad I \stackrel{k_p}{\rightarrow} P \stackrel{1}{\rightarrow} \phi$$

$$\phi \quad \stackrel{k_s}{\underset{k_d}{\rightleftharpoons}} \quad S$$

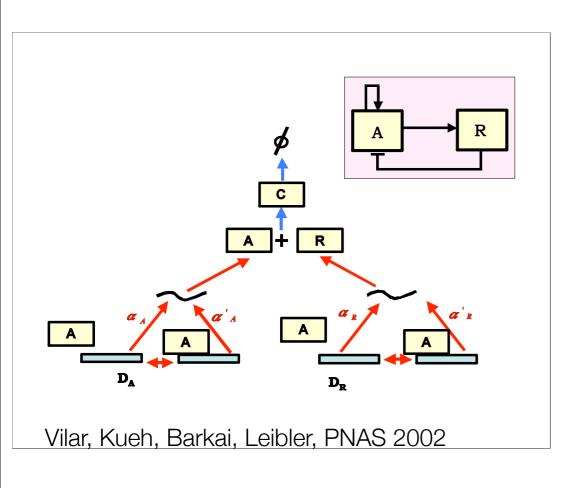


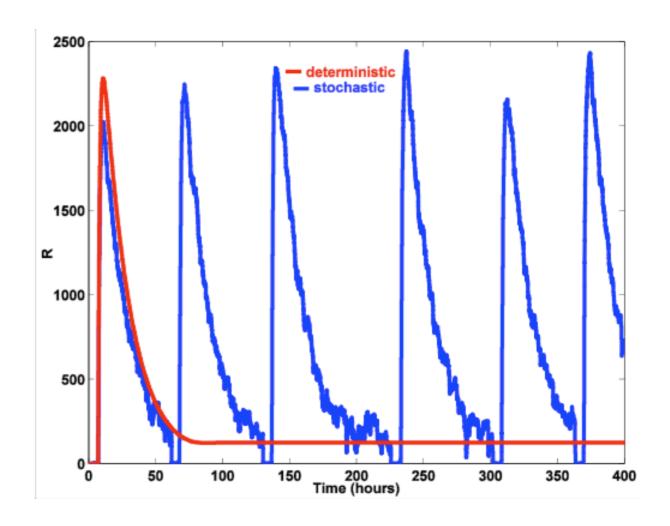
Johan Paulsson, Otto G. Berg, and Måns Ehrenberg, "Stochastic Focusing: Fluctuation-enhansed sensitivity of intracellular regulation" PNAS 2000

- Stochastic mean value different from deterministic steady state
- Noise enhances signal!

Noise Induced Oscillations

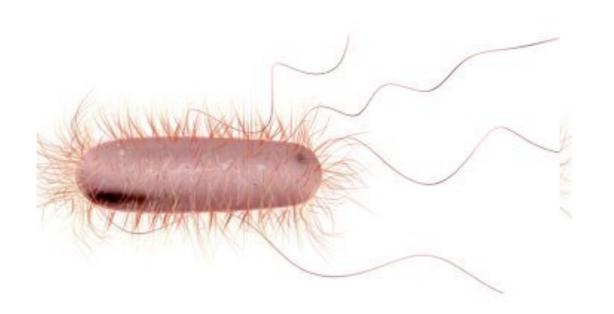
Circadian rhythm

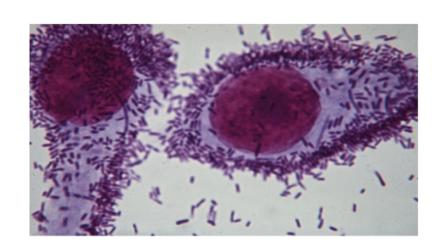




- Oscillations disappear from deterministic model after a small reduction in deg. of repressor
- (Coherence resonance) Regularity of noise induced oscillations can be manipulated by tuning the level of noise [*El-Samad, Khammash*]

The Pap Pili Stochastic Switch



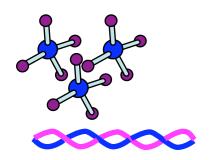


- Pili enable uropathogenic E. coli to attach to epithelial cell receptors
 - Plays an essential role in the pathogenesis of urinary tract infections
- E. coli expresses two states ON (piliated) or OFF (unpiliated)
- Piliation is controlled by a stochastic switch that involves random molecular events

The Importance of Stochasticity.

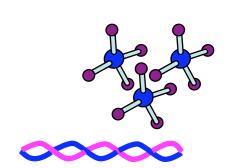
Stochastic Switching: Identical genotypes and identical environments can produce different phenotypes.

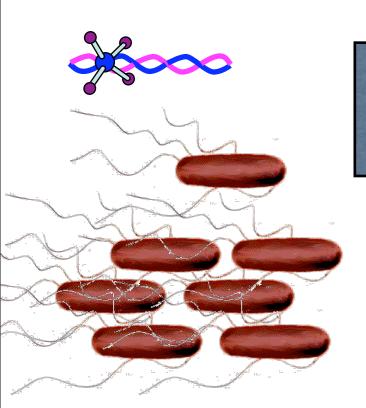




Same chemical environment.

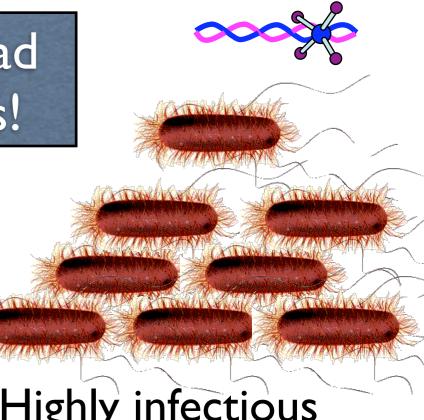
Same genetic code.





Harmless phenotype.

Random reactions can lead to vastly different results!



Highly infectious phenotype.

The Importance of Stochasticity.

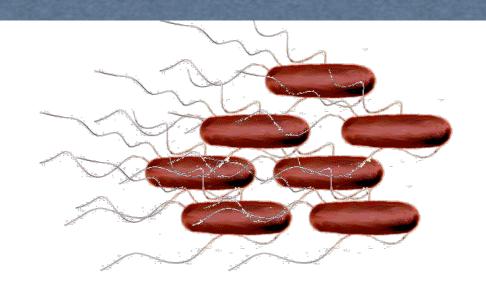
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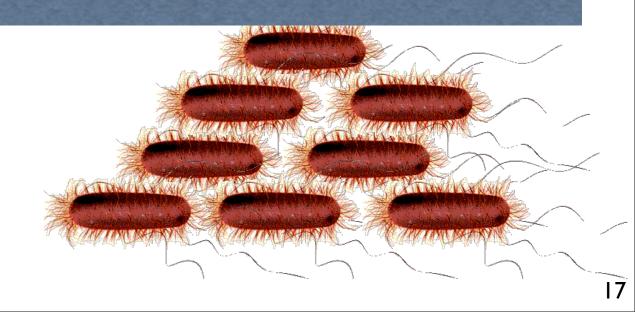


For these systems, we need analytical models to answer:

- ★ What will happen?
- ★ How frequently?
- ★ Why does it happen?
- ★ Under what conditions?

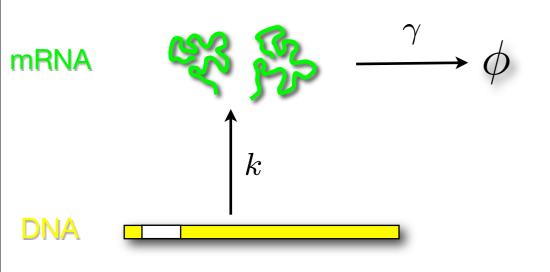
- ★ What advantages does it provide?
- ★ How can we prevent it?
- ★ How can we cause it?







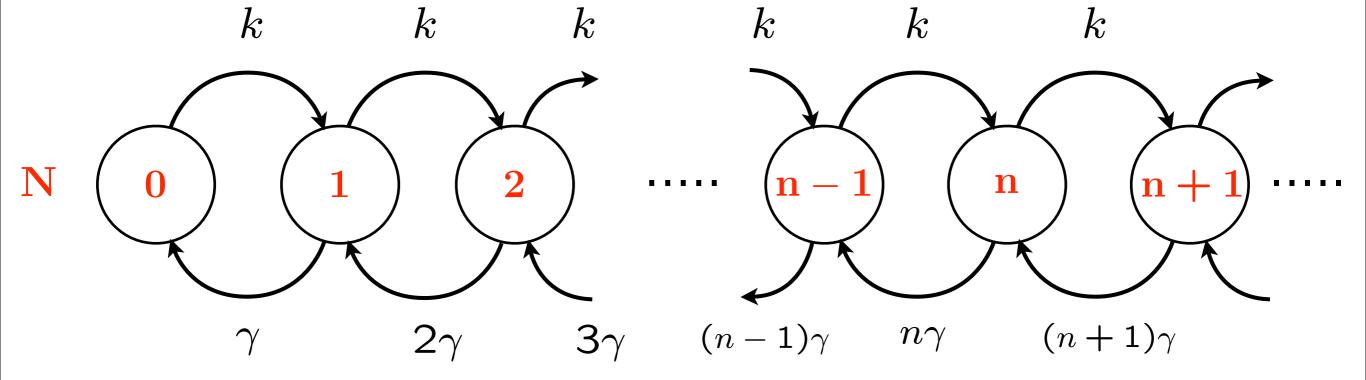
RNA Copy Number as a Random Variable



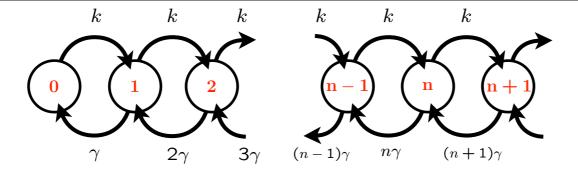
mRNA copy number N(t) is a random variable

Transcription: Probability a single mRNA is transcribed in time dt is k dt

Degradation: Probability a single mRNA is degraded in time dt is $n\gamma dt$



Key Question:



Find p(n,t), the probability that N(t) = n.

$$P(n,t+dt) = P(n-1,t) \cdot kdt \qquad \text{Prob.} \{N(t) = n-1 \text{ and mRNA created in } [t,t+dt)\}$$

$$+ P(n+1,t) \cdot (n+1)\gamma dt \qquad \text{Prob.} \{N(t) = n+1 \text{ and mRNA degraded in } [t,t+dt)\}$$

$$+ P(n,t) \cdot (1-kdt)(1-n\gamma dt) \quad \text{Prob.} \{N(t) = n \text{ and mRNA degraded in } [t,t+dt)\}$$

$$+ P(n,t) \cdot (1-kdt)(1-n\gamma dt) \quad \text{Prob.} \{N(t) = n \text{ and mRNA degraded in } [t,t+dt)\}$$

$$P(n, t + dt) - P(n, t) = P(n - 1, t)kdt + P(n + 1, t)(n + 1)\gamma dt - P(n, t)(k + n\gamma)dt + O(dt^{2})$$

Dividing by dt and taking the limit as $dt \rightarrow 0$

The Chemical Master Equation

$$\frac{d}{dt}P(\mathbf{n},t) = kP(\mathbf{n}-\mathbf{1},t) + (n+1)\gamma P(\mathbf{n}+\mathbf{1},t) - (k+n\gamma)P(\mathbf{n},t)$$

mRNA Stationary Distribution

We look for the stationary distribution $P(n,t) = p(n) \ \forall t$

The stationary solution satisfies: $\frac{d}{dt}P(n,t) = 0$

From the Master Equation ...

$$(k+n\gamma)p(n) = kp(n-1) + (n+1)\gamma p(n+1)$$

$$n = 0 kp(0) = \gamma p(1)$$

$$n = 1 \qquad kp(1) = 2\gamma p(2)$$

$$n = 2$$
 $kp(2) = 3\gamma p(3)$

.

$$kp(n-1) = n\gamma \ p(n)$$

 $kp(n-1) = n\gamma \ p(n)$ We can express p(n) as a function of p(0):

$$p(n) = \frac{k}{\gamma} \frac{1}{n} p(n-1)$$

$$= \left(\frac{k}{\gamma}\right)^2 \frac{1}{n} \frac{1}{n-1} p(n-2)$$

$$\vdots$$

$$= \left(\frac{k}{\gamma}\right)^n \frac{1}{n!} p(0)$$

We can solve for p(0) using the fact $\sum_{n=0}^{\infty} p(n) = 1$

$$1 = \sum_{n=0}^{\infty} \left(\frac{k}{\gamma}\right)^n \frac{1}{n!} p(0)$$

$$= e^{k/\gamma} p(0) \implies p(0) = e^{-k/\gamma}$$

$$p(n) = e^{-a} \frac{a^n}{n!} \qquad a = \frac{k}{\gamma}$$

Poisson Distribution

We can compute the mean and variance of the Poisson RV \bar{N} with density $p(n) = e^{-a} \frac{a^n}{n!}$:

$$\mu = E[\bar{N}] = \sum_{n=0}^{\infty} np(n) = e^{-a} \sum_{n=0}^{\infty} n \frac{a^n}{n!} = a$$

The second moment

$$E[\bar{N}^2] = \sum_{n=0}^{\infty} n^2 p(n) = a^2 + a$$

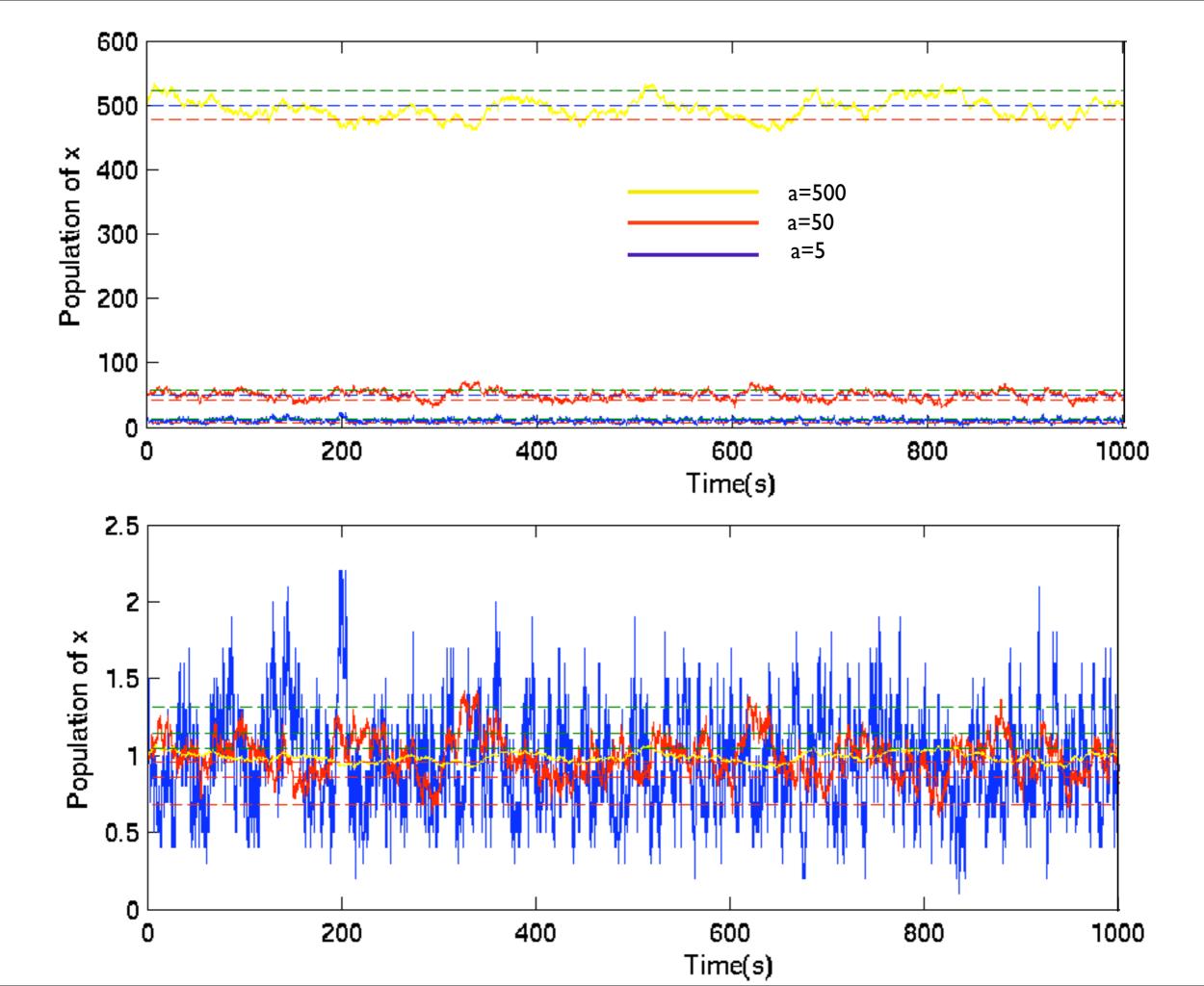
Therefore,

$$\sigma^2 = E[\bar{N}^2] - E[\bar{N}]^2 = a$$

mean = variance = a

The coefficient of variation $C_v = \sigma/\mu$ is

$$C_v = \frac{1}{\sqrt{a}} = \frac{1}{\sqrt{\mu}}$$

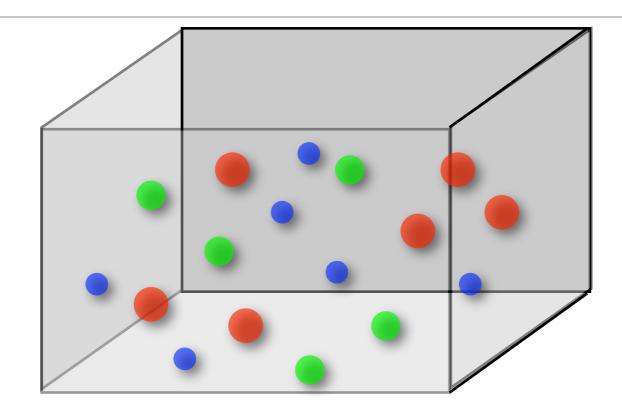


Stochastic Chemical Kinetics

Formulation of Stochastic Chemical Kinetics

Gillespie, Physical A, 1992

Reaction volume= Ω



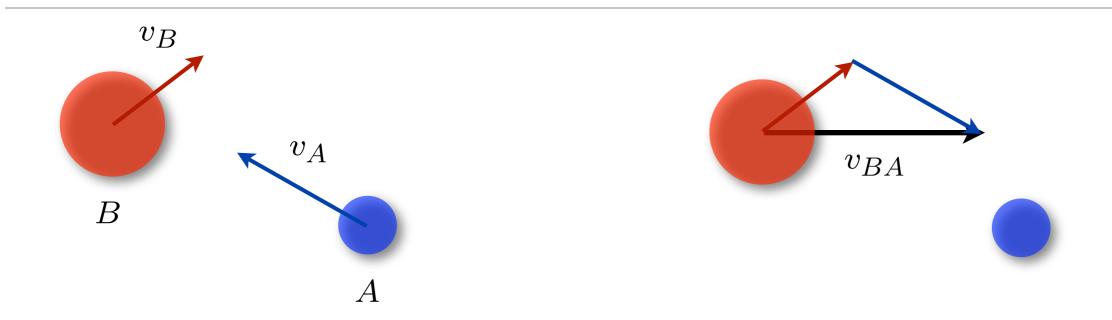
Key Assumptions

(**Well-Mixed**) The probability of finding any molecule in a region $d\Omega$ is given by $\frac{d\Omega}{\Omega}$.

(**Thermal Equilibrium**) The molecules move due to the thermal energy. The reaction volume is at a constant temperature T. The velocity of a molecule is determined according to a Boltzman distribution:

$$f_{v_x}(v) = f_{v_y}(v) = f_{v_z}(v) = \sqrt{\frac{m}{2\pi k_B T}} e^{-\frac{m}{2k_B T}v^2}$$

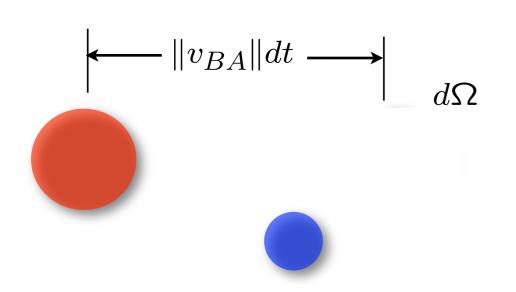
Probability of Collision: Two Specific Molecules



Given:

- ullet Two spheres A and B with velocities v_A and v_B , and radii r_A and r_B .
- The probability that the center of either sphere lies in a volume $d\Omega$ is given by $\frac{d\Omega}{\Omega}$.

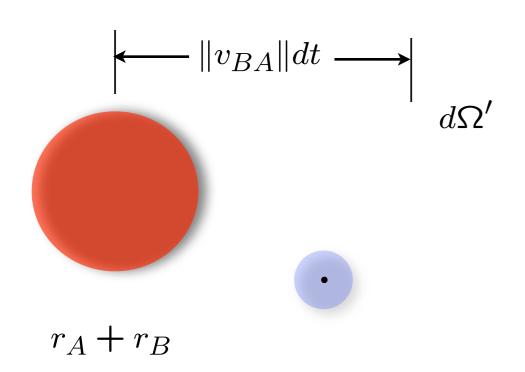
What is the probability that A and B will collide in the time [t, t + dt]?



In the time [t,t+dt] molecule A sweeps a volume of $d\Omega=\pi r_B^2 \ \|v_{BA}\| \ dt$

Collision takes place if any part of A lies in the region $d\Omega$.

Equivalently ...



During [t, t+dt] a molecule with radius r_A+r_B sweeps a volume of $d\Omega'=\pi(r_A+r_B)^2 \|v_{BA}\| dt$

Collision takes place if the center of A lies in the region $d\Omega'$.

The probability of A and B colliding during [t, t + dt] is

$$\frac{1}{\Omega}\pi(r_A + r_B)^2 ||v_{BA}|| \ dt$$

Note:

- The probability of A and B colliding was computed for a given a relative velocity of v_{BA} (conditional probability)
- The relative velocity is a random variable, and we must average over all velocities.

If we denote by $f_{BA}(\cdot)$ the probability density of the random variable V_{BA} we have

Collision Probability in [t,t+dt]
$$= \int_{\mathbb{R}^3} P(\text{collision in } [t,t+dt] \mid V_{BA} = v) \ f_{BA}(v) dv$$

$$= \int_{\mathbb{R}^3} \frac{1}{\Omega} \pi (r_A + r_B)^2 ||v|| dt \ f_{BA}(v) dv$$

$$= \frac{1}{\Omega} \pi (r_A + r_B)^2 dt \int_{\mathbb{R}^3} ||v|| f_{BA}(v) dv$$
 mean relative speed

The probability density function of $f_{BA}(\cdot)$ can be easily computed from the Boltzman distribution of the velocity and the independence of V_x , V_y , and V_z .

$$f_{BA}(v) = \left(\frac{\hat{m}}{2\pi k_B T}\right)^{3/2} e^{-\frac{\hat{m}}{2k_B T}||v||^2}, \quad \text{where } \hat{m} = \frac{m_A + m_B}{2}$$

Hence

Mean relative speed
$$= \int_{\mathbb{R}^3} \|v\| f_{BA}(v) dv$$

$$= \int_{\mathbb{R}^3} \|v\| \left(\frac{\hat{m}}{2\pi k_B T}\right)^{3/2} e^{-\frac{\hat{m}}{2k_B T} \|v\|^2} dv$$

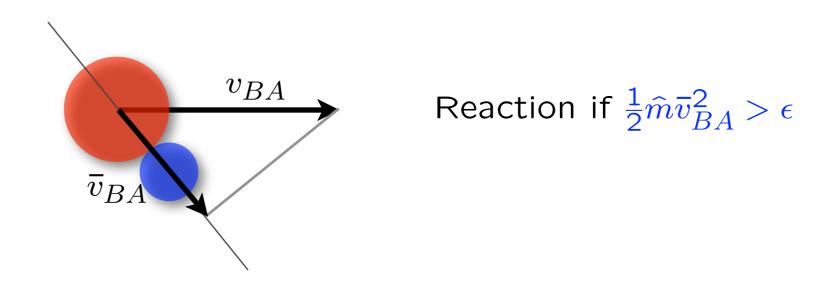
$$= \sqrt{\frac{8k_B T}{\pi \hat{m}}}$$

Probability of A-B collision within [t,t+dt]:

$$\frac{1}{\Omega}\pi(r_A+r_B)^2dt\sqrt{\frac{8k_BT}{\pi\hat{m}}}$$

Not all collisions lead to reactions. One can factor in the "reaction energy".

Assumption: An A-B collision leads to a reaction only if the kinetic energy associated with the component of the velocity along the line of contact is greater than a critical energy ϵ .



It can be shown that:

Probability (A-B reaction | A-B collision) = $e^{-\frac{\epsilon}{k_BT}}$

Probability of A-B reaction within [t,t+dt]:

$$\frac{1}{\Omega}\pi(r_A+r_B)^2\sqrt{\frac{8k_BT}{\pi\hat{m}}}e^{-\frac{\epsilon}{k_BT}}dt$$

Given N species: S_1, \ldots, S_N with populations x_1, \ldots, x_N at time t.

Consider the bimolecular reaction channel (with distinct species):

$$R: \mathcal{S}_i + \mathcal{S}_j \rightarrow \text{products}$$

The number of distinct $S_i - S_j$ pairs that can react is: $x_i \cdot x_j$. Therefore,

Probability of an R reaction within [t,t+dt]:

$$x_i x_j \frac{1}{\Omega} \pi (r_i + r_j)^2 \sqrt{\frac{8k_B T}{\pi \hat{m}}} e^{-\frac{\epsilon}{k_B T}} dt = w(x)$$

 $w(\cdot)$ is called the propensity function.

Consider the bimolecular reaction channel (with same species):

$$R': \mathcal{S}_i + \mathcal{S}_i \to \text{products}$$

The number of distinct $S_i - S_i$ pairs that can react is: $\frac{x_i(x_i-1)}{2}$. Therefore,

Probability of an R' reaction within [t,t+dt]:

$$\frac{x_i(x_i-1)}{2} \frac{1}{\Omega} \pi r_i^2 \sqrt{\frac{8k_B T}{\pi \hat{m}}} e^{-\frac{\epsilon}{k_B T}} dt = w(x) dt$$

Reactions and Propensity Functions

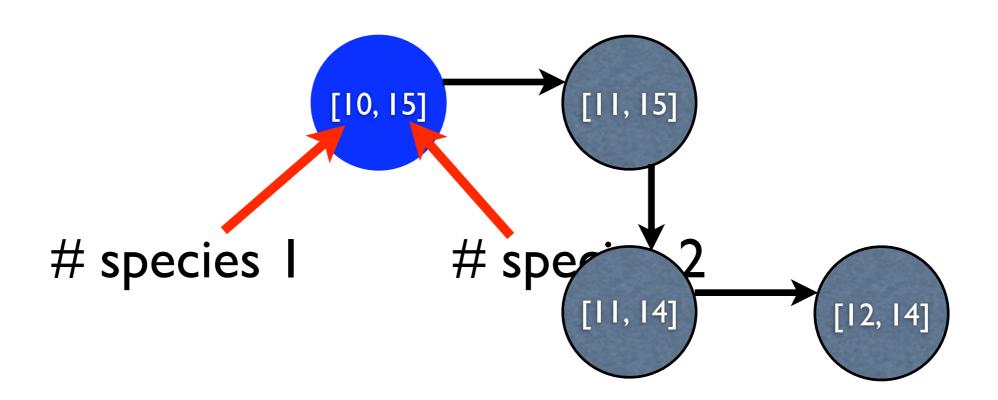
Reaction	Propensity $w(x)$	Rate
$\phi \xrightarrow{c} Products$	c	
$\mathcal{S}_i \stackrel{c}{ o} Products$	$c \cdot x_i$	
$S_i + S_j \xrightarrow{c} Products$	$c \cdot x_i x_j$	$\frac{1}{\Omega}\pi(r_i+r_j)^2 \sqrt{\frac{8k_BT}{\pi\hat{m}}}e^{-\frac{\epsilon}{k_BT}}$
$S_i + S_i \xrightarrow{c} $ Products	$c \cdot rac{x_i(x_i-1)}{2}$	$\frac{4}{\Omega}\pi r_i^2 \sqrt{\frac{8k_BT}{\pi \hat{m}}}e^{-\frac{\epsilon}{k_BT}}$

For a monomolecular reaction: c is numerically equal to the reaction rate constant k of conventional deterministic chemical kinetics

For a bimolecular reaction: c is numerically equal to k/Ω , where k is the reaction rate constant of conventional deterministic chemical kinetics

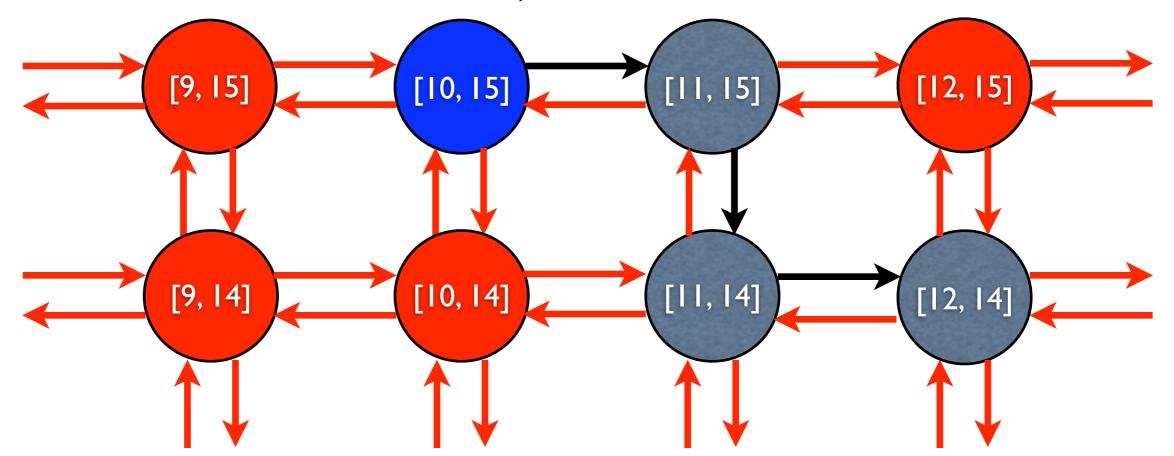
A Jump-Markov description of chemical kinetics

- At any time, the state of the system is defined by its integer population vector: $\mathbf{x} \in \mathbb{Z}^N$
- Reactions are transitions from one state to another:

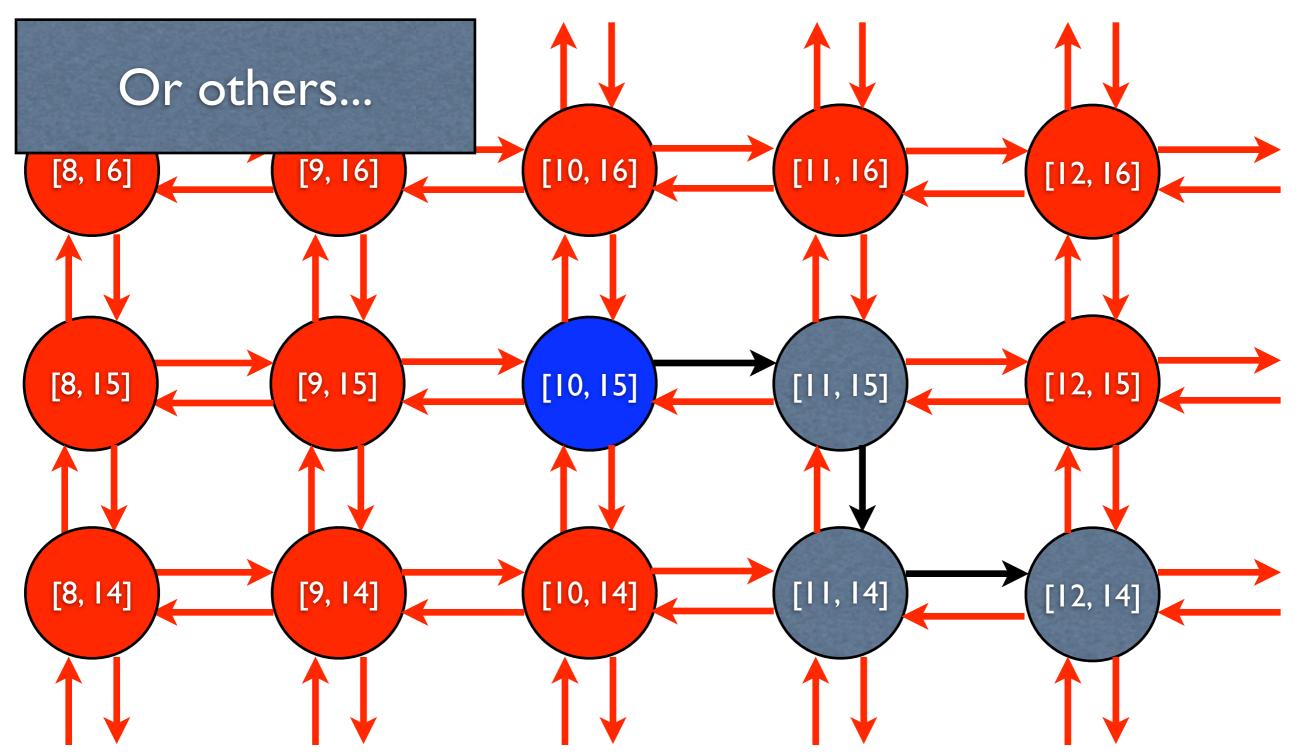


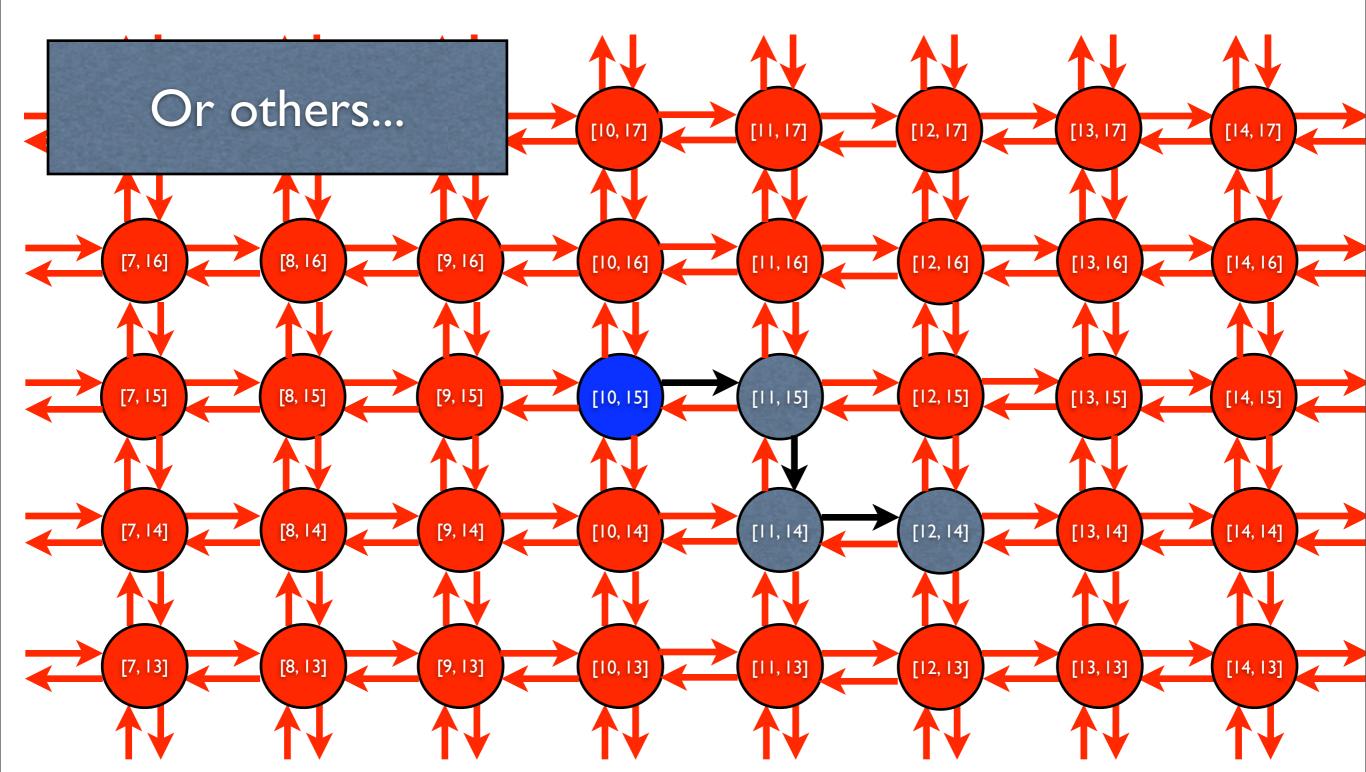
A Jump-Markov description of chemical kinetics

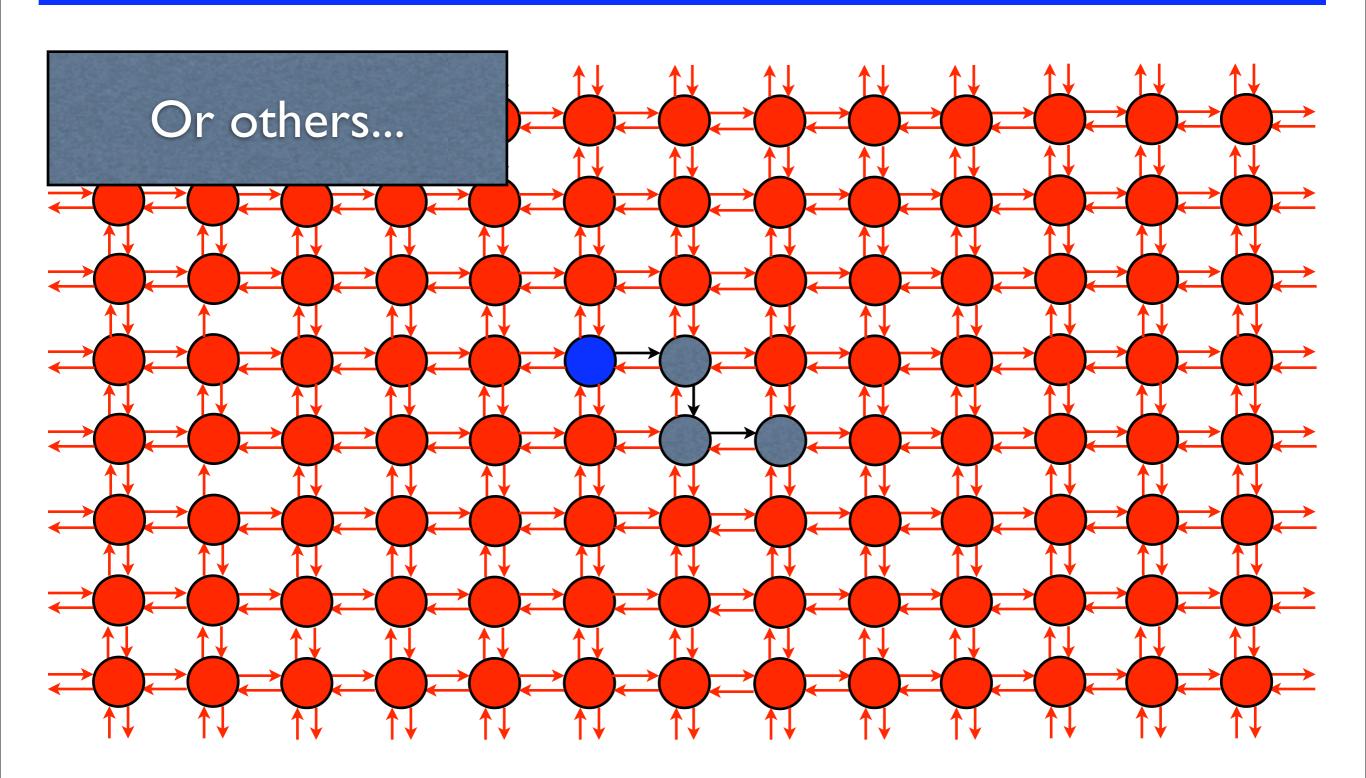
- At any time, the state of the system is defined by its integer population vector: $\mathbf{x} \in \mathbb{Z}^N$
- Reactions are transitions from one state to another:
- These reactions are random, others could have occurred:

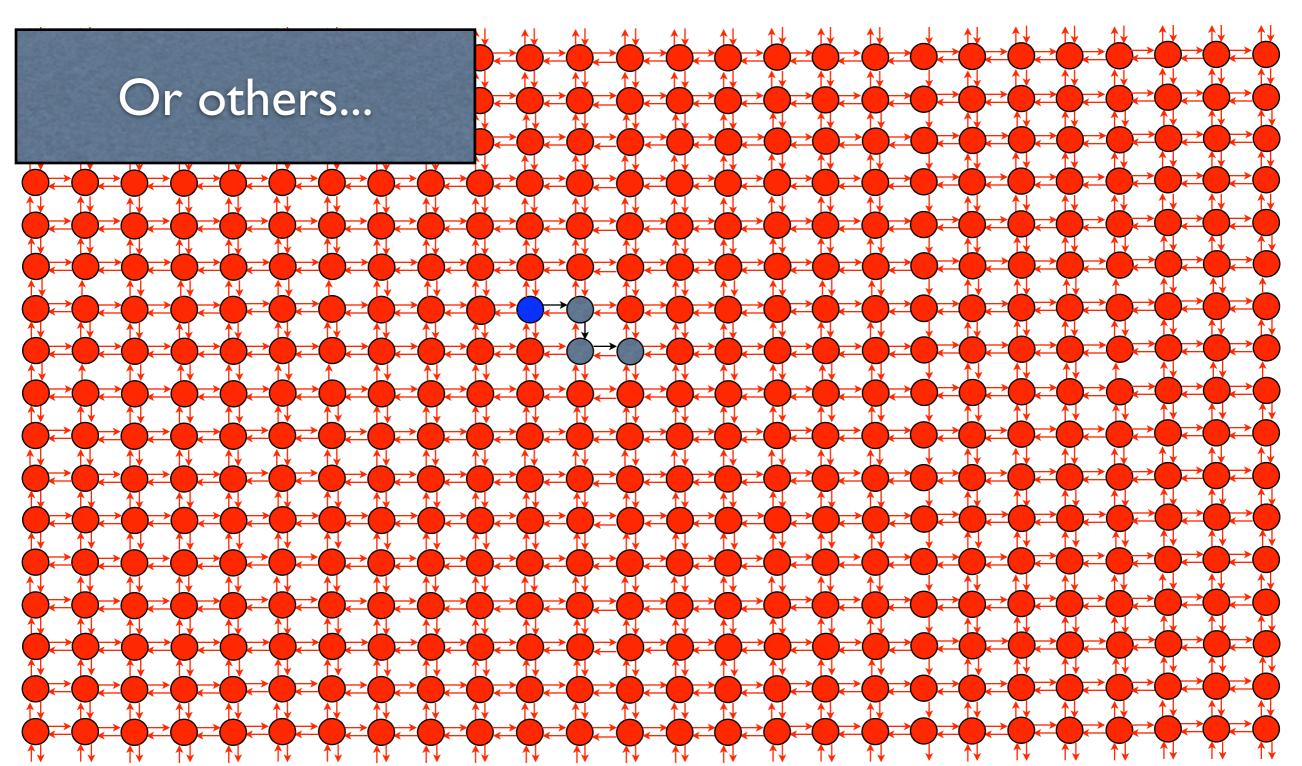


A Jump-Markov description of chemical kinetics









Reaction Stoichiometry (review)

- The Stoichiometric vector, **s**, refers to the relative change in the population vector after a reaction.
- There may be many different reactions for a given stoichiometry.

$$\mathbf{s}_1 = [1, 0]^T$$

$$\mathcal{S}_1 \to \mathcal{S}_1 + \mathcal{S}_1$$

$$\mathcal{S}_2 \to \mathcal{S}_2 + \mathcal{S}_1$$

$$\emptyset \to \mathcal{S}_1$$

$$\mathbf{s}_2 = [-1, 0]^T$$

$$\mathcal{S}_1 + \mathcal{S}_1 \to \mathcal{S}_1$$

$$\mathcal{S}_1 + \mathcal{S}_2 \to \mathcal{S}_2$$

$$\mathcal{S}_1 \to \emptyset$$

$$\mathbf{s}_3 = [0, 1]^T$$

$$\mathcal{S}_2 \to \mathcal{S}_2 + \mathcal{S}_2$$

$$\mathcal{S}_1 \to \mathcal{S}_1 + \mathcal{S}_2$$

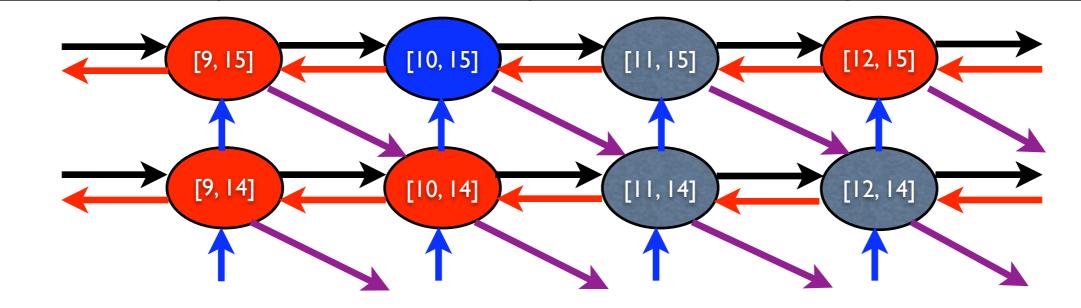
$$\emptyset \to \mathcal{S}_2$$

$$\mathbf{s}_4 = [1, -1]^T$$

$$\mathcal{S}_2 \to \mathcal{S}_1$$

$$\mathcal{S}_1 + \mathcal{S}_2 \to \mathcal{S}_1 + \mathcal{S}_1$$

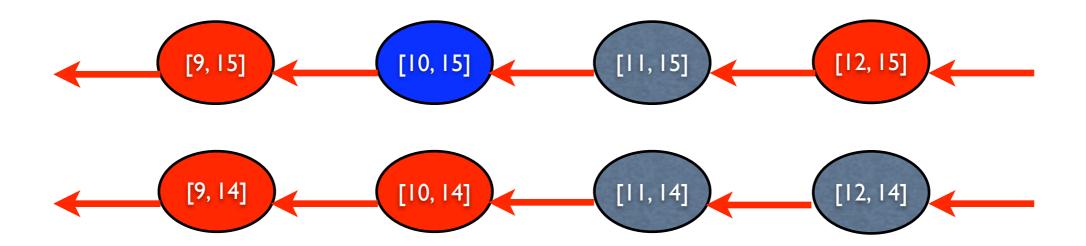
$$\mathcal{S}_2 + \mathcal{S}_2 \to \mathcal{S}_1 + \mathcal{S}_2$$



Reaction Propensities (review)

- The propensity, w, of a reaction is its rate.
- $\mathbf{w}_{\mu}dt$ is the probability that the μ^{th} reaction will occur in a time step of length dt .
- Typically, propensities depend only upon reactant populations.

$\mathbf{s}_2 = [-1, 0]^T$	$w_2(x_1, x_2)$
$\mathcal{S}_1 + \mathcal{S}_1 o \mathcal{S}_1$	$k_1 x_2 (x_1 - 1)/2$
$\mathcal{S}_1 + \mathcal{S}_2 ightarrow \mathcal{S}_2$	$k_{2}x_{1}x_{2}$
$\mathcal{S}_1 o \emptyset$	k_3x_1



The Chemical Master Equation

Prob. that no reactions fire in $[t, t+dt] = 1 - \sum_k w_k(x)dt + \mathcal{O}(dt^2)$

Prob. that reaction R_k fires once in $[t, t+dt] = w_k(x)dt + \mathcal{O}(dt^2)$

Prob. that more than one reaction fires in $[t, t + dt] = \mathcal{O}(dt^2)$

$$p(x,t+dt) = p(x,t) \left(1 - \sum_k w_k(x)dt + \mathcal{O}(dt^2)\right) \\ + \sum_k p(x-s_k,t) \left(\sum_k w_k(x)dt + \mathcal{O}(dt^2)\right) + \mathcal{O}(dt^2) \\ R_k \text{ reaction away from } x \\ R_k \text{ fires once}$$
 more than one reaction in dt

$$p(x,t+dt) - p(x,t) = -p(x,t) \sum_{k} w_{k}(x)dt + \sum_{k} p(x-s_{k},t)w_{k}(x)dt + \mathcal{O}(dt^{2})$$

The Chemical Master Equation

$$\frac{dp(x,t)}{dt} = -p(x,t)\sum_{k} w_k(x) + \sum_{k} p(x-s_k,t)w_k(x-s_k)$$

Relationship of Stochastic and Deterministic Descriptions

Given N species X_1, \ldots, X_N and M elementary reactions. Let $\Phi_i := [X_i]$.

A deterministic description can be obtained from mass-action kinetics:

$$\frac{d\Phi}{dt} = Sf(\Phi)$$

where $f(\cdot)$ is at most a second order monomial. It depends on the type of reactions and their rates.

Example:

$$A + B \xrightarrow{k_1} C$$

$$A \xrightarrow{k_2} B$$

$$\frac{d\Phi_A}{dt} = -k_1 \Phi_A \Phi_B - k_2 \Phi_A$$

$$\frac{d\Phi}{dt} = Sf(\Phi) \text{ where}$$

$$\frac{d\Phi}{dt} = Sf(\Phi) = Sf(\Phi)$$

$$\frac{d\Phi}{dt} = Sf(\Phi) = Sf(\Phi)$$
or
$$S = \begin{bmatrix} -1 & -1 \\ -1 & 1 \\ 1 & 0 \end{bmatrix}, f(\Phi) = \begin{bmatrix} k_1 \Phi_A \Phi_B \\ k_2 \Phi_A \end{bmatrix}$$

Relationship of Stochastic and Deterministic Descriptions

Define
$$X^{\Omega}(t) = \frac{X(t)}{\Omega}$$
.

Question: How does $X^{\Omega}(t)$ relate to $\Phi(t)$?

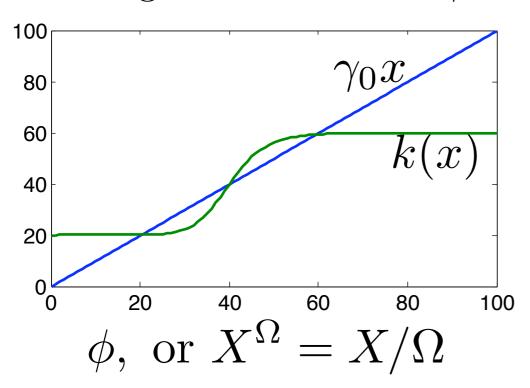
Fact: Let $\Phi(t)$ be the deterministic solution to the reaction rate equations

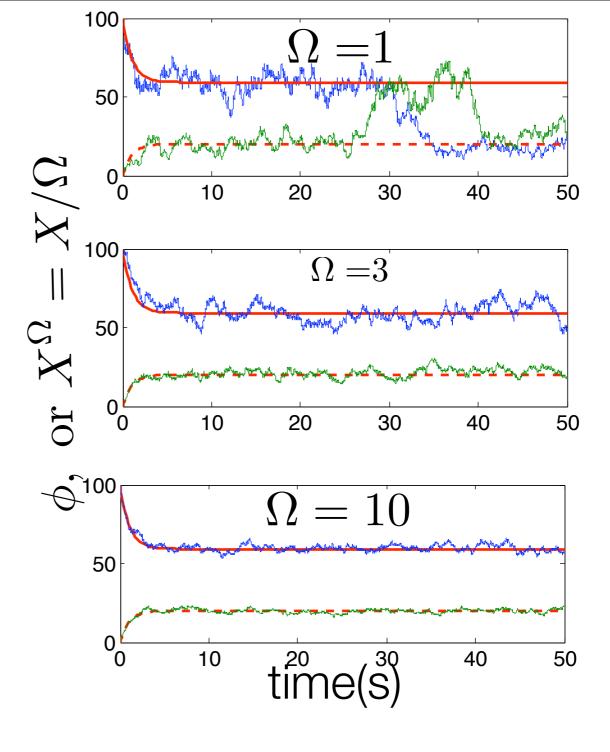
$$\frac{d\Phi}{dt} = Sf(\Phi), \ \Phi(0) = \Phi_0.$$

Let $X^{\Omega}(t)$ be the stochastic representation of the same chemical systems with $X^{\Omega}(0) = \Phi_0$. Then for every $t \geq 0$:

$$\lim_{\Omega \to \infty} \sup_{s < t} |X^{\Omega}(s) - \Phi(s)| = 0 \ a.s.$$

x produced with rate k(x) and degraded with rate $\gamma_0 x$.





$$w_1(\phi)=\gamma \phi$$
 $w_2(\phi)=\left(20+40rac{\phi^{10}}{40^{10}+\phi^{10}}
ight)$ Deterministic

$$w_1(X)=\Omega\gamma_0X/\Omega=\gamma_0X$$

$$w_2(X)=\Omega\left(20+40\frac{(X/\Omega)^{10}}{40^{10}+(X/\Omega)^{10}}\right)$$
 Stochastic

Moment Computations

- Affine Propensity
- Linear Noise Approximation

Moment Computations

For the first moment $E[X_i]$, multiply the CME by x_i and sum over all $(x_1,\ldots,x_N)\in\mathbb{N}^N$

For the second moment $E[X_iX_j]$, multiply the CME by x_ix_j and sum over all $(x_1,\ldots,x_N)\in\mathbb{N}^N$

$$\frac{dE[X_i]}{dt} = \sum_{k=1}^{M} s_{ik} E[w_k(X)]$$

$$\frac{dE[X_i X_j]}{dt} = \sum_{k=1}^{M} (s_{ik} E[X_j w_k(X)] + E[X_i w_k(X)] s_{jk} + s_{ik} s_{jk} E[w_k(X)])$$
Let $w(x) = [w_1(x), \dots, w_M(x)]^T$

In matrix notation:

$$\frac{dE[X]}{dt} = SE[w(X)]$$

$$\frac{dE[XX^T]}{dt} = SE[w(X)X^T] + E[w(X)X^T]^T S^T + S\{diagE[w(X)]\}S^T$$

Affine Propensity

Suppose the propensity function is affine:

$$w(x) = Wx + w_0,$$
 (W is $N \times N$, w_0 is $N \times 1$)

Then $E[w(X)] = WE[X] + w_0$, and $E[w(X)X^T] = WE[XX^T] + w_0E[X^T]$.

This gives us the moment equations:

$$\frac{d}{dt}E[X] = SWE[X] + Sw_0$$
 First Moment
$$\frac{d}{dt}E[XX^T] = SWE[XX^T] + E[XX^T]W^TS^T + S \operatorname{diag}(WE[X] + w_0)S^T + Sw_0E[X^T] + E[X]w_0^TS^T$$
 Second Moment

These are linear ordinary differential equations and can be easily solved!

Affine Propensity (cont.)

Define the covariance matrix $\Sigma = E[(X - E[X])(X - E(X))^T]$.

We can also compute covariance equations:

$$\frac{d}{dt}\Sigma = SW\Sigma + \Sigma W^T S^T + S \operatorname{diag}(WE[X] + w_0)S^T$$

Steady-state Case

The steady-state moments and covariances can be obtained by solving linear algebraic equations:

Let
$$\bar{X} = \lim_{t \to \infty} E[X(t)]$$
 and $\bar{\Sigma} = \lim_{t \to \infty} \Sigma(t)$.

Then

$$SW\bar{X} = -Sw_0$$

$$SW\bar{\Sigma} + \bar{\Sigma}W^TS^T + S \operatorname{diag}(W\bar{X} + w_0)S^T = 0$$

Fluctuations Arise from Noise Driven Dynamics

Define A = SW, and $B = S\sqrt{diag(W\bar{X} + w_0)}$.

The steady-state covariances equation

$$SW\bar{\Sigma} + \bar{\Sigma}W^TS^T + S \operatorname{diag}(W\bar{X} + w_0)S^T = 0$$

becomes

$$A\bar{\Sigma} + \bar{\Sigma}A^T + BB^T = 0$$
 Lyapunov Equation

Moment Computations

- Affine Propensity
- Linear Noise Approximation

Linear Noise Approximation (LNA)

Let
$$X^{\Omega}(t) := \frac{X(t)}{\Omega}$$

Write $X^{\Omega} = \Phi_0(t) + \frac{1}{\sqrt{\Omega}}V^{\Omega}$ where $\Phi_0(t)$ solves the deterministic RRE

$$\frac{d\Phi}{dt} = Sf(\Phi)$$

Linear Noise Approximation

$$V^{\Omega}(t) \to V(t)$$
 as $\Omega \to \infty$, where $dV(t) = A(t)V(t)dt + B(t)dW_t$

$$A(t) = \frac{d[Sf(\Phi)]}{d\Phi} \qquad B(t) := S\sqrt{diag[f(\Phi_0(t))]}$$

Linear Noise Approximation: $X^{\Omega}(t) \approx \Phi(t) + \frac{1}{\sqrt{\Omega}}V(t)$

Multiplying
$$X^{\Omega}(t) \approx \bar{\Phi} + \frac{1}{\sqrt{\Omega}}V(t)$$
 by Ω , we get

population

$$X(t) \approx \Omega \bar{\Phi} + \sqrt{\Omega} V(t)$$

deterministic zero mean concentration stochastic

$$E[X(t)] = \Omega \bar{\Phi}$$

Let $\overline{\Sigma}$ be the covariance matrix of $\sqrt{\Omega} \cdot V(t)$. Then

$$\frac{d}{dt}\bar{\Sigma}(t) = A(t)\bar{\Sigma}(t) + \bar{\Sigma}(t)A^{T}(t) + \Omega B(t)B(t)^{T}$$

$$A(t) = \frac{d[Sf(\Phi)]}{d\Phi}$$

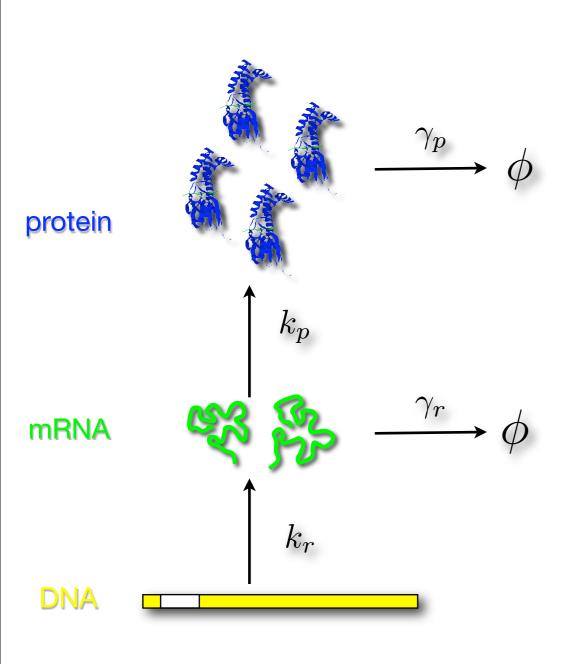
$$B(t) := S\sqrt{diag[f(\Phi_0(t))]}$$

At stationary distribution, we have the same Lyapunov equation as in the affine linear case:

$$A = SW B = S\sqrt{diag(W\bar{X} + w_0)}$$

Application to Gene Expression

Application to Gene Expression



Reactants

 $X_1(t)$ is # of mRNA; $X_2(t)$ is # of protein

Reactions

 $\xrightarrow{\gamma_p} \phi \qquad R_1 : \phi \xrightarrow{k_r} mRNA$

 $R_2: mRNA \xrightarrow{\gamma_r} \phi$

 $R_3: mRNA \xrightarrow{k_p} protein + mRNA$

 $R_4: protein \xrightarrow{\gamma_p} \phi$

Stoichiometry and Propensity

$$S = \begin{bmatrix} 1 & -1 & 0 & 0 \\ 0 & 0 & 1 & -1 \end{bmatrix}$$

$$w(X) = \begin{bmatrix} k_r \\ \gamma_r X_1 \\ k_p X_1 \\ \gamma_p X_2 \end{bmatrix} = \begin{bmatrix} 0 & 0 \\ \gamma_r & 0 \\ k_p & 0 \\ 0 & \gamma_p \end{bmatrix} \begin{bmatrix} X_1 \\ X_2 \end{bmatrix} + \begin{bmatrix} k_r \\ 0 \\ 0 \\ 0 \end{bmatrix}$$

$$W$$

Steady-State Moments

$$A = SW = \begin{bmatrix} -\gamma_r & 0 \\ k_p & -\gamma_p \end{bmatrix}, \qquad Sw_0 = \begin{bmatrix} k_r \\ 0 \end{bmatrix}$$

$$\bar{X} = -A^{-1}Sw_0 = \begin{bmatrix} \frac{k_r}{\gamma_r} \\ \frac{k_p k_r}{\gamma_p \gamma_r} \end{bmatrix}$$

Steady-State Covariance

$$BB^{T} = S \operatorname{diag}(W\bar{X} + w_{0})S^{T} = \begin{bmatrix} 2k_{r} & 0\\ 0 & \frac{2k_{p}k_{r}}{\gamma_{r}} \end{bmatrix}$$

The steady-state covariances equation

$$A\bar{\Sigma} + \bar{\Sigma}A^T + BB^T = 0$$
 Lyapunov Equation

can be solved algebraically for $\bar{\Sigma}$.

$$ar{\Sigma} = egin{bmatrix} rac{k_{p}k_{r}}{\gamma_{r}} & rac{k_{p}k_{r}}{\gamma_{r}(\gamma_{r}+\gamma_{p})} \ rac{k_{p}k_{r}}{\gamma_{r}(\gamma_{r}+\gamma_{p})} & rac{k_{p}k_{r}}{\gamma_{p}\gamma_{r}}(1+rac{k_{p}}{\gamma_{r}+\gamma_{p}}) \end{bmatrix}$$

Coefficients of Variation

$$C_{vr}^2 = \frac{1}{\frac{k_r}{\gamma_r}} = \frac{1}{\bar{X}_1}$$

$$C_{vp}^{2} = \frac{1}{\frac{k_r k_p}{\gamma_r \gamma_p}} \left(1 + \frac{k_p}{\gamma_r + \gamma_p} \right) = \frac{1}{\bar{X}_2} \left(1 + \frac{k_p}{\gamma_r + \gamma_p} \right)$$

Question: Does a large \bar{X}_2 imply a small C_{vp} ?

$$C_{vp}^{2} = \frac{1}{\frac{k_{r}k_{p}}{\gamma_{r}\gamma_{p}}} \left(1 + \frac{k_{p}}{\gamma_{r} + \gamma_{p}} \right)$$

$$\geq \frac{1}{\frac{k_{r}k_{p}}{\gamma_{r}\gamma_{p}}} \left(\frac{k_{p}}{\gamma_{r} + \gamma_{p}} \right) = \frac{\gamma_{r}\gamma_{p}}{k_{r}} \cdot \frac{1}{\gamma_{r} + \gamma_{p}}$$

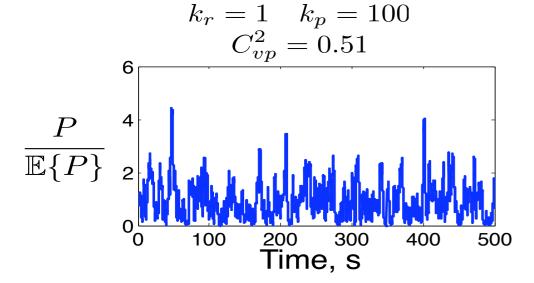
 $\bar{X}_2 = \frac{k_r k_p}{\gamma_r \gamma_p}$, which can be chosen *independently* from C_{vp} .

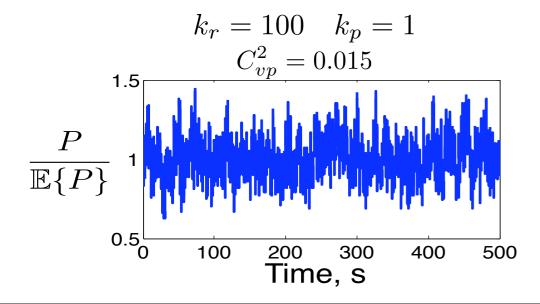
Large mean does not imply small fluctuations!

$$\mathbb{E}\{P\} = 100, \quad \gamma_r = \gamma_p = 1$$

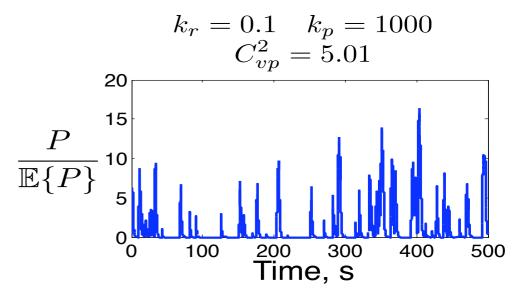
$$k_r = 0.01 \quad k_p = 10,000 \\ C_{vp}^2 = 50.01$$

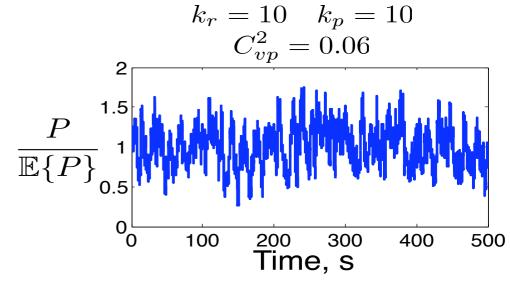
$$\frac{P}{\mathbb{E}\{P\}}_{20}$$
0 100 200 300 400 500 Time, s

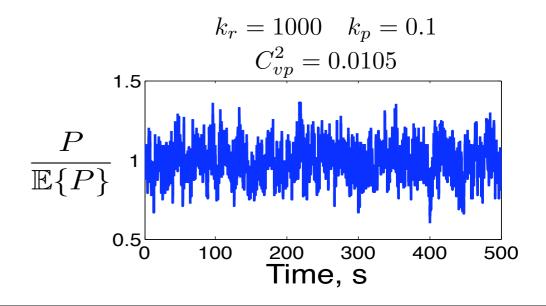






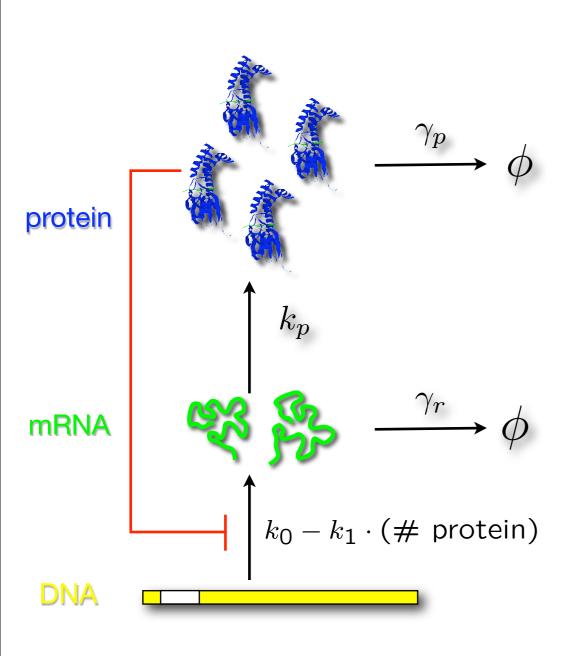






Noise Suppression and Exploitation

Noise Attenuation through Negative Feedback



Reactants

 $X_1(t)$ is # of mRNA; $X_2(t)$ is # of protein

Reactions

$$\xrightarrow{\gamma_p} \phi \qquad R_1: \phi \xrightarrow{k_r} mRNA \qquad k_r = k_0 - k_1 \cdot (\# \text{ protein})$$

$$R_2: mRNA \xrightarrow{\gamma_r} \phi$$

$$R_3: mRNA \xrightarrow{k_p} protein + mRNA$$

$$R_4: protein \xrightarrow{\gamma_p} \phi$$

Stoichiometry and Propensity

$$S = \begin{bmatrix} 1 & -1 & 0 & 0 \\ 0 & 0 & 1 & -1 \end{bmatrix}$$

$$w(X) = \begin{bmatrix} k_0 - k_1 X_2 \\ \gamma_r X_1 \\ k_p X_1 \\ \gamma_p X_2 \end{bmatrix} = \begin{bmatrix} 0 & -k_1 \\ \gamma_r & 0 \\ k_p & 0 \\ 0 & \gamma_p \end{bmatrix} \begin{bmatrix} X_1 \\ X_2 \end{bmatrix} + \begin{bmatrix} k_0 \\ 0 \\ 0 \\ 0 \end{bmatrix}$$

Steady-State Moments

$$A = SW = \begin{bmatrix} -\gamma_r & -k_1 \\ k_p & -\gamma_p \end{bmatrix}, \qquad Sw_0 = \begin{bmatrix} k_0 \\ 0 \end{bmatrix}$$

$$\bar{X} = -A^{-1}Sw_0 = \begin{bmatrix} \frac{\frac{k_0}{\gamma_r}}{1 + \frac{k_1 k_p}{\gamma_p \gamma_r}} \\ \frac{\frac{k_0 k_p}{\gamma_r \gamma_p}}{1 + \frac{k_1 k_p}{\gamma_p \gamma_r}} \end{bmatrix} =: \begin{bmatrix} \mu_r \\ \mu_p \end{bmatrix}$$

Steady-State Covariance

$$BB^T = S \operatorname{diag}(W\bar{X} + w_0)S^T = \begin{bmatrix} k_0 + \gamma_r \mu_r - k_1 \mu_p & 0\\ 0 & k_p \mu_r + \gamma_p \mu_p \end{bmatrix}$$

The steady-state covariances equation

$$A\bar{\Sigma} + \bar{\Sigma}A^T + BB^T = 0$$
 Lyapunov Equation

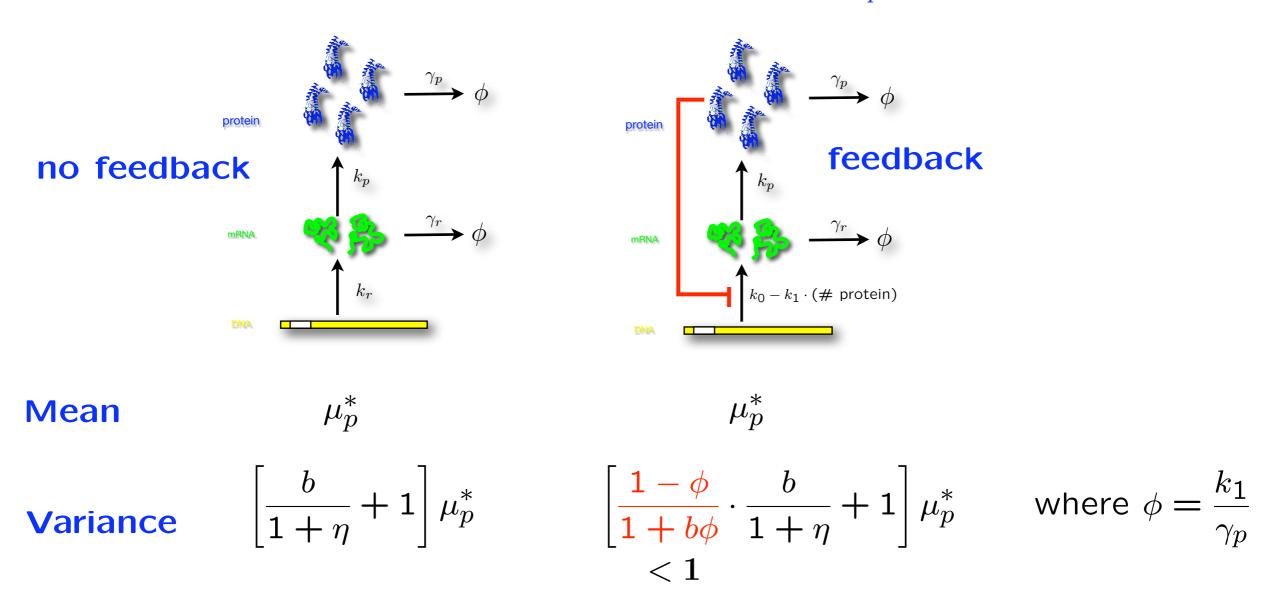
can be solved algebraically for $\bar{\Sigma}$.

$$\bar{\Sigma}_{22} = \sigma_p^2 = \left[\frac{1-\phi}{1+b\phi} \cdot \frac{b}{1+\eta} + 1\right] \mu_p \qquad \text{where } \phi = \frac{k_1}{\gamma_p}, \ b = \frac{k_p}{\gamma_r}, \ \eta = \frac{\gamma_p}{\gamma_r}$$

Feedback vs. No Feedback

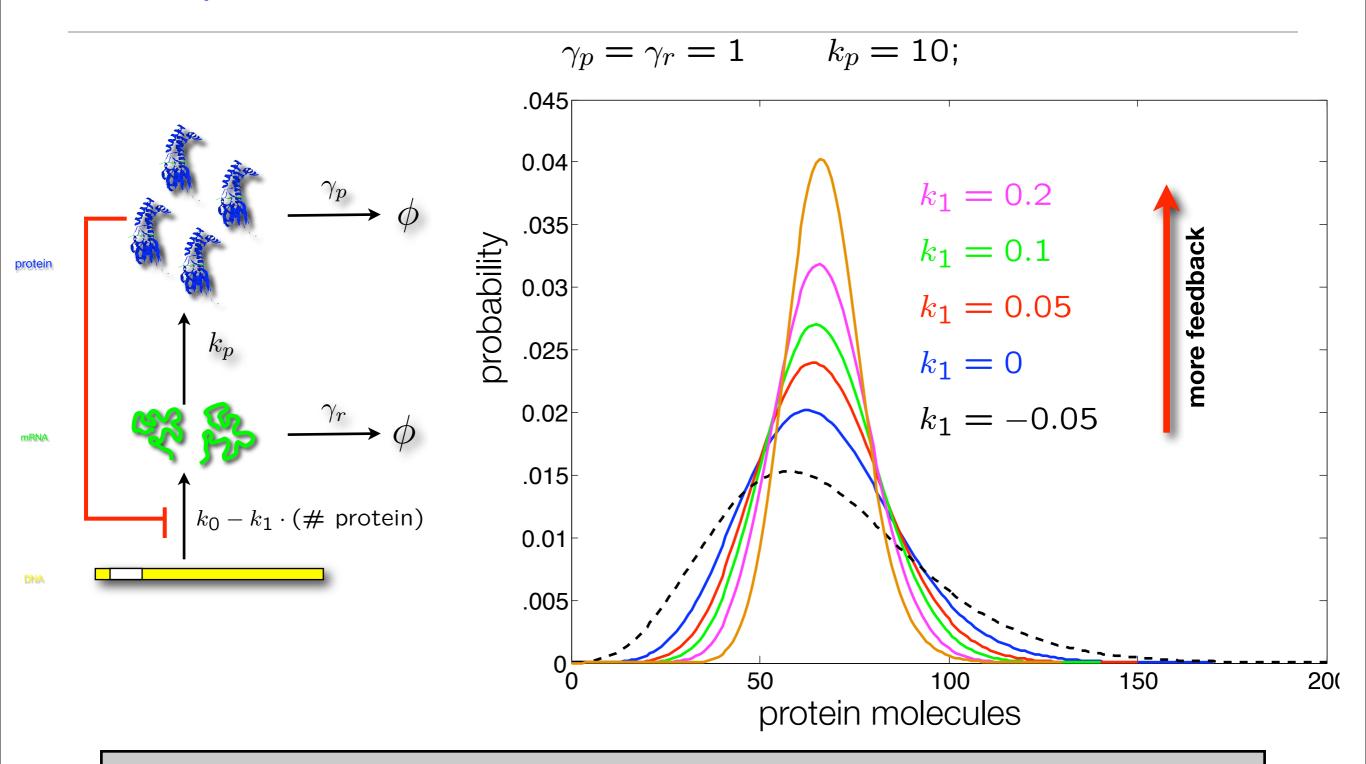
In order to compare the noise in the two cases, we must ensure that both configuations have the same mean!

Impose the constraint: $\mu_p^{FB} = \mu_p^{NFB} =: \mu_p^*$ This may be achieved by choosing $k_0 = k_r + k_1 \mu_p^{NFB}$.



Protein variance is always smaller with negative feedback!

Example



Note that these distributions are NOT Gaussian.

Exploiting the Noise: Failure of the linear noise approximation

$$\phi \quad \stackrel{k}{\underset{k_a S}{\rightleftharpoons}} \quad I \stackrel{k_p}{\rightarrow} P \stackrel{1}{\rightarrow} \phi$$

$$\phi \quad \stackrel{k_s}{\underset{k_d}{\rightleftharpoons}} \quad S$$

may be approximated by

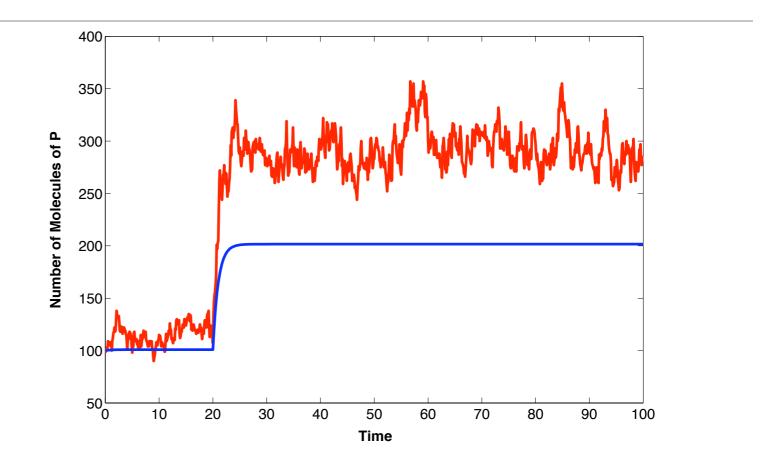
$$\phi \xrightarrow{kq} P \xrightarrow{1} \phi$$

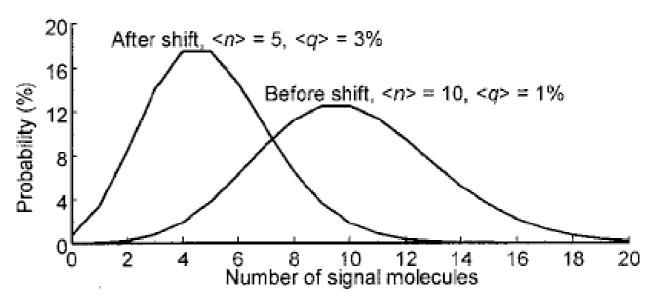
$$q = \frac{1}{1 + \frac{n}{\Omega K}} \qquad K = k_p/k_a$$
 n is #S
$$\text{CONVEX}$$

From Jensen's Inequality:

$$E[q] = E\left[\frac{1}{1 + \frac{n}{\Omega K}}\right] \ge \frac{1}{1 + \frac{E[n]}{\Omega K}}$$

Noise enhances signal!





Stochastic Gene Expression in Systems Biology (Part 2)

Brian Munsky

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On the menu...

Yesterday

- Overview of Stochastic Gene Expression
- Stochastic Chemical Kinetics
- Solutions for Simple Stochastic Processes (Transcription)
- Moment Computations for Linear Propensities
- Linear Noise Approximation

Today

- Monte Carlo Simulation Techniques
 - * Gillespie (SSA), Tau leaping, Chemical Langevin (SDEs), Slow Scale SSA.
- Density Computations with Finite State Projection Techniques
- Switch and Trajectory Analyses

Relationship of Stochastic and Deterministic Descriptions

Define
$$X^{\Omega}(t) = \frac{X(t)}{\Omega}$$
.

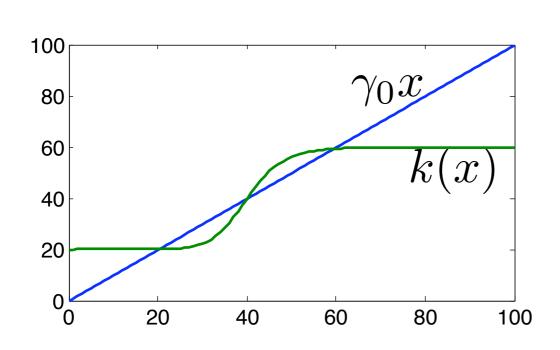
Question: How does $X^{\Omega}(t)$ relate to $\Phi(t)$?

Fact: Let $\Phi(t)$ be the deterministic solution to the reaction rate equations

$$\frac{d\Phi}{dt} = Sf(\Phi), \ \Phi(0) = \Phi_0.$$

Let $X^{\Omega}(t)$ be the stochastic representation of the same chemical systems with $X^{\Omega}(0) = \Phi_0$. Then for every $t \geq 0$:

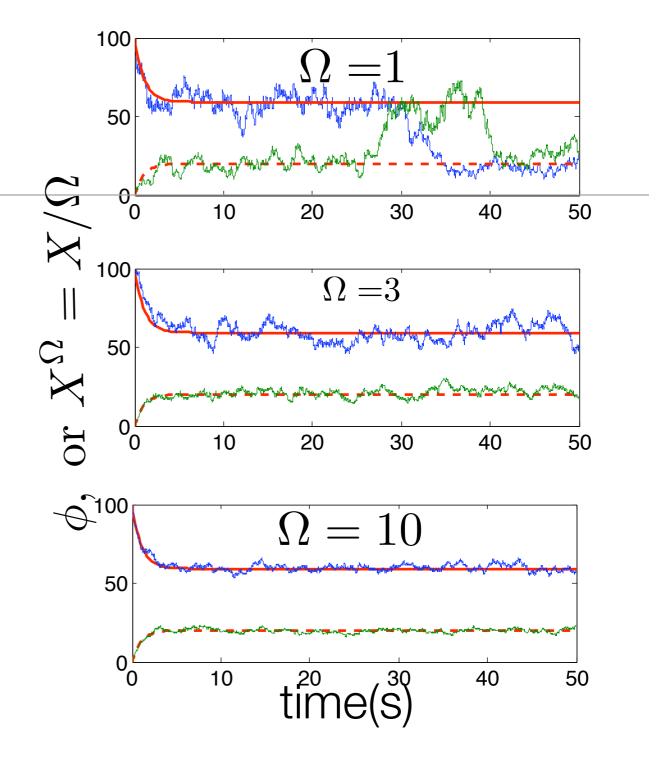
$$\lim_{\Omega \to \infty} \sup_{s < t} |X^{\Omega}(s) - \Phi(s)| = 0 \ a.s.$$



$$\phi$$
, or $X^{\Omega} = X/\Omega$

$$w_1(\phi) = \gamma_0 x$$

$$w_2(\phi) = \left(20 + 40 \frac{\phi^{10}}{40^{10} + \phi^{10}}\right)$$



$$w_1(X) = \Omega \gamma_0 X / \Omega = \gamma_0 X$$

$$w_2(X) = \Omega \left(20 + 40 \frac{(X/\Omega)^{10}}{40^{10} + (X/\Omega)^{10}} \right)$$

Monte Carlo Methods

Kinetic Monte-Carlo Simulation Methods



Stochastic Simulation Algorithm

- •D.T. Gillespie, J. Phys. Chem. A 81, 2340 (1977)
- •M. Gibson and J. Bruck, J. Phys. Chem. **104**, 1876 (2000)

τ leaping

- •D. Gillespie, J. Chem. Phys. **115**, 1716 (2001); **119**, 8229 (2003)
- •M. Rathinam et al., J. Chem. Phys. 119, 12784 (2003)
- •T. Tian and K. Burrage, J. Chem. Phys. **121**, 10356 (2004)
- •A. Chatterjee, et al. J. Chem. Phys. 122, 054104 (2005)
- •Y. Cao, D. Gillespie and L. Petzold, J. Chem. Phys. 123, 054104 (2005)

Chemical Langevin Equations

•D. Gillespie, J. Chem. Phys. 113, 1716 (2000)

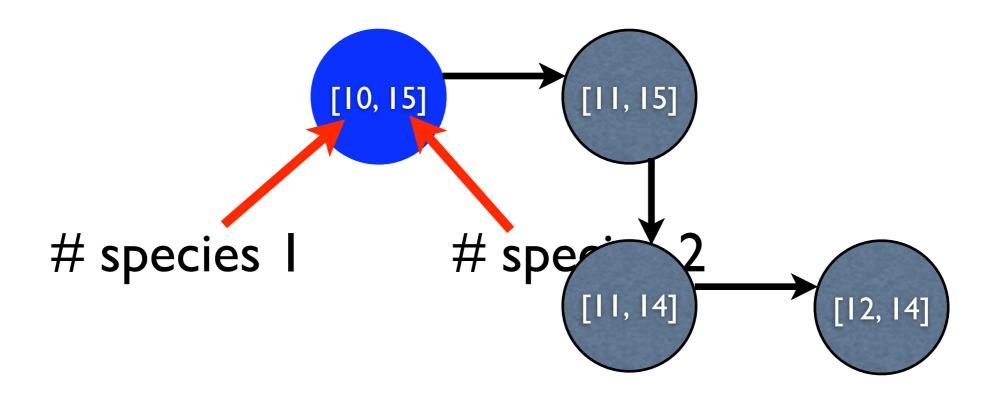
System Partitioning Methods

- •C. Rao and A. Arkin, J. Chem. Phys. 118, 4999 (2003)
- •Y. Cao et al., J. Chem. Phys. 122, 014116 (2005)

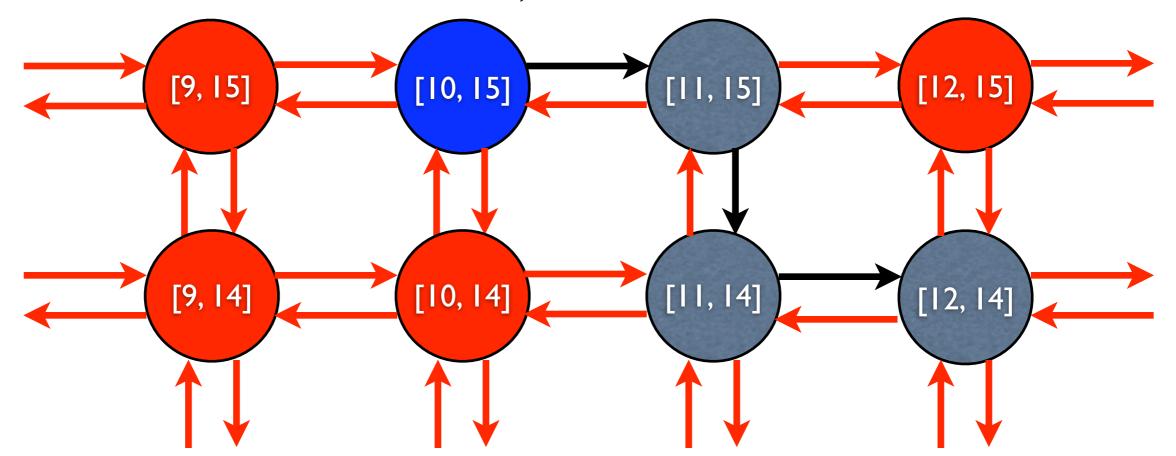
Hybrid Methods

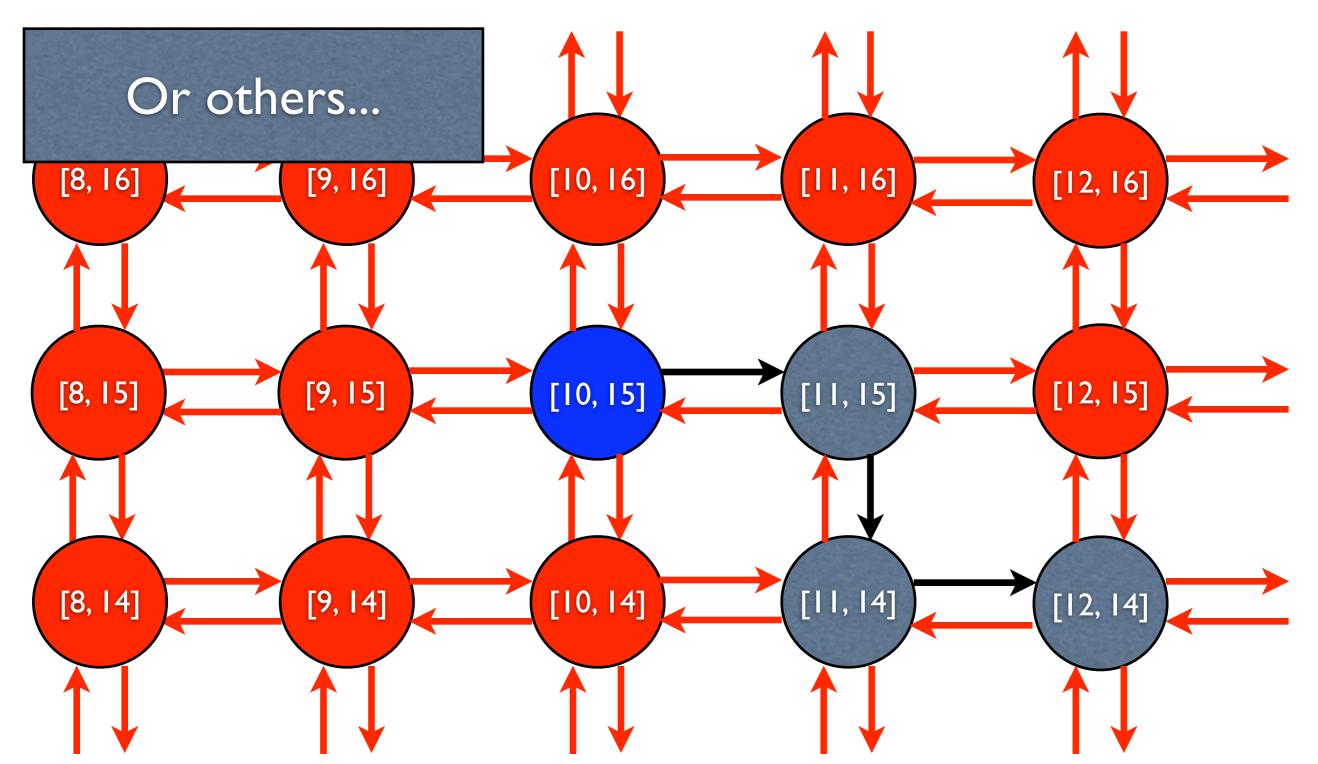
- •E. Haseltine and J. Rawlings, J. Chem. Phys. **117**, 6959 (2002)
- •H. Salis and Y. Kaznessis, J. Chem. Phys. **122**, 054103 (2005)

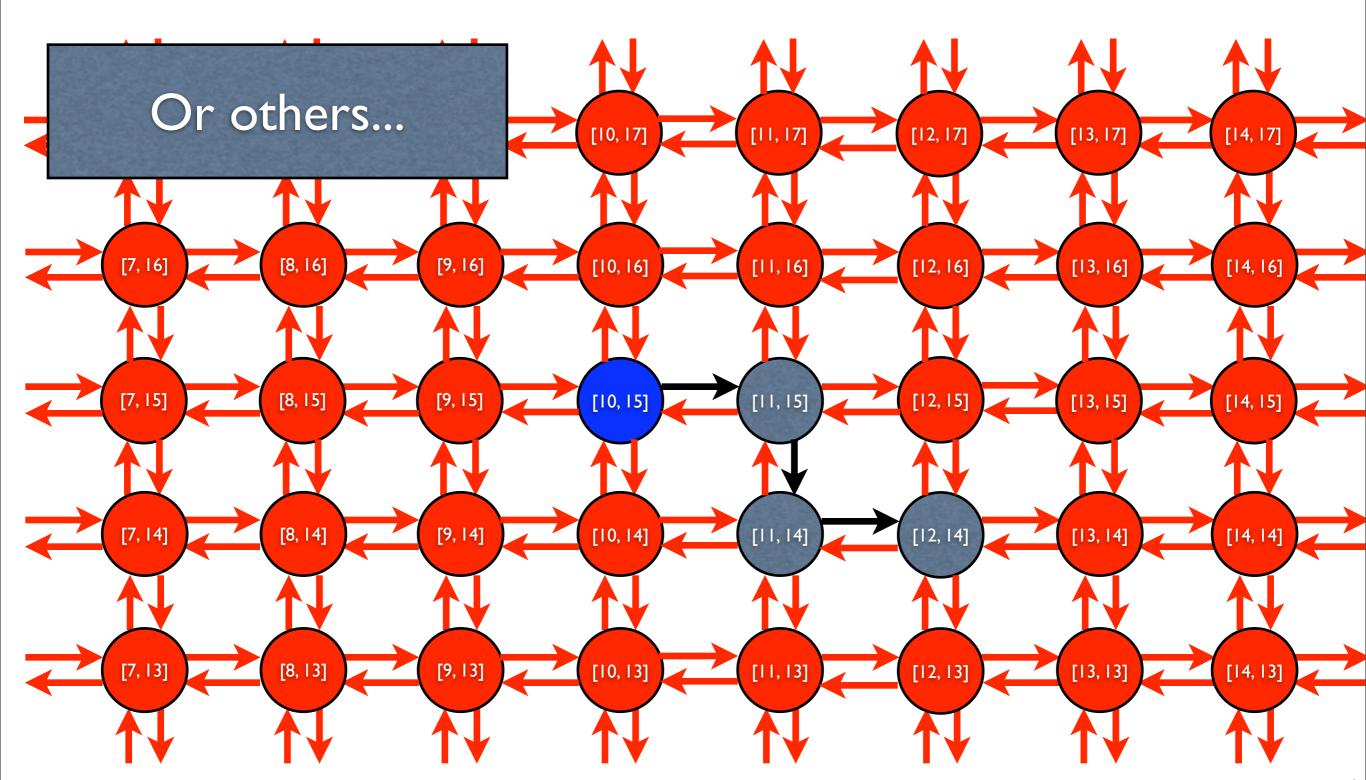
- At any time, the state of the system is defined by its integer population vector: $\mathbf{x} \in \mathbb{Z}^N$
- Reactions are transitions from one state to another:

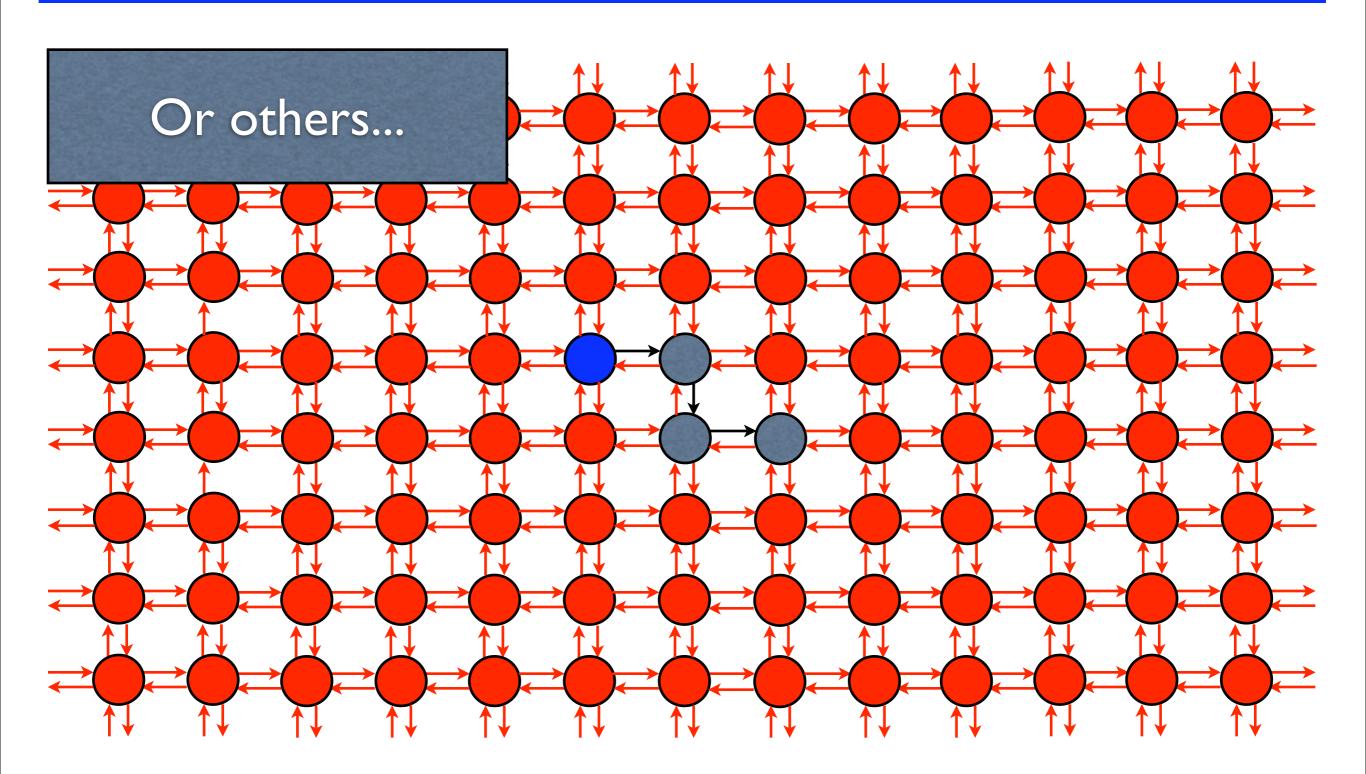


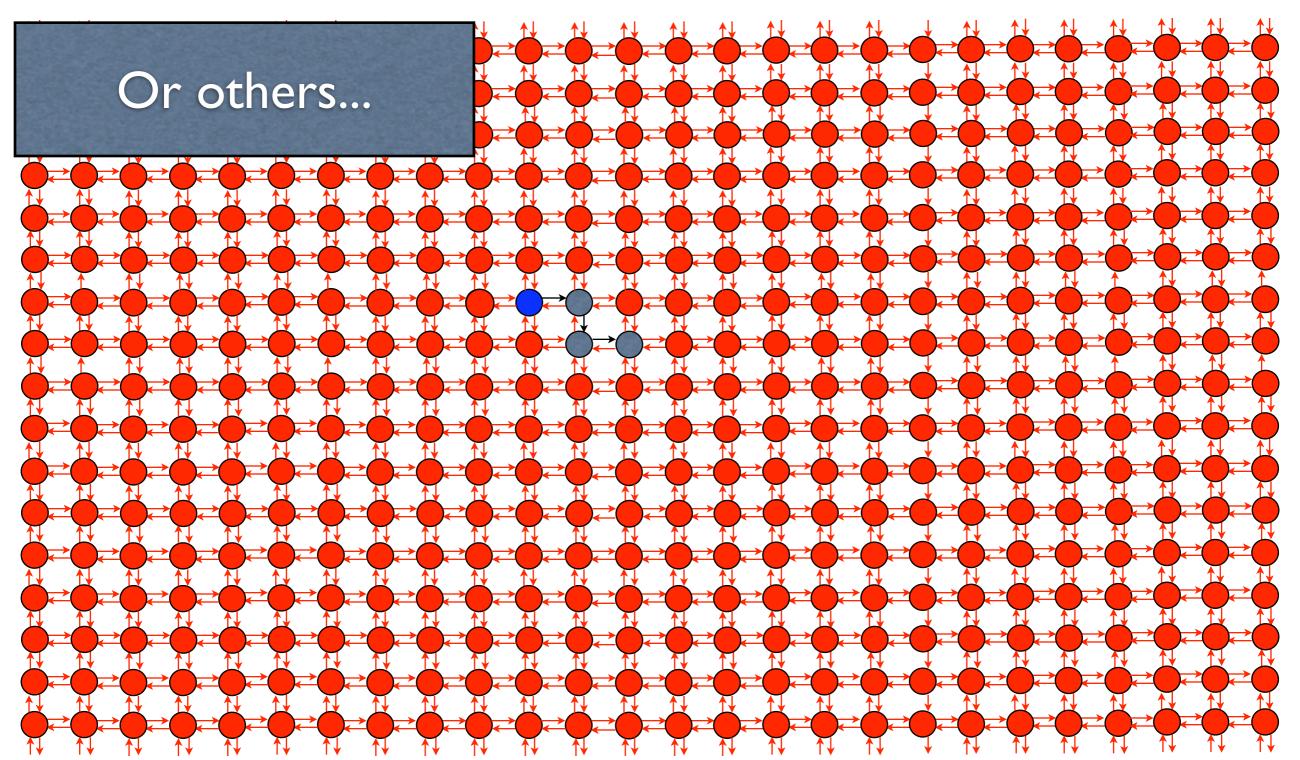
- At any time, the state of the system is defined by its integer population vector: $\mathbf{x} \in \mathbb{Z}^N$
- Reactions are transitions from one state to another:
- These reactions are random, others could have occurred:

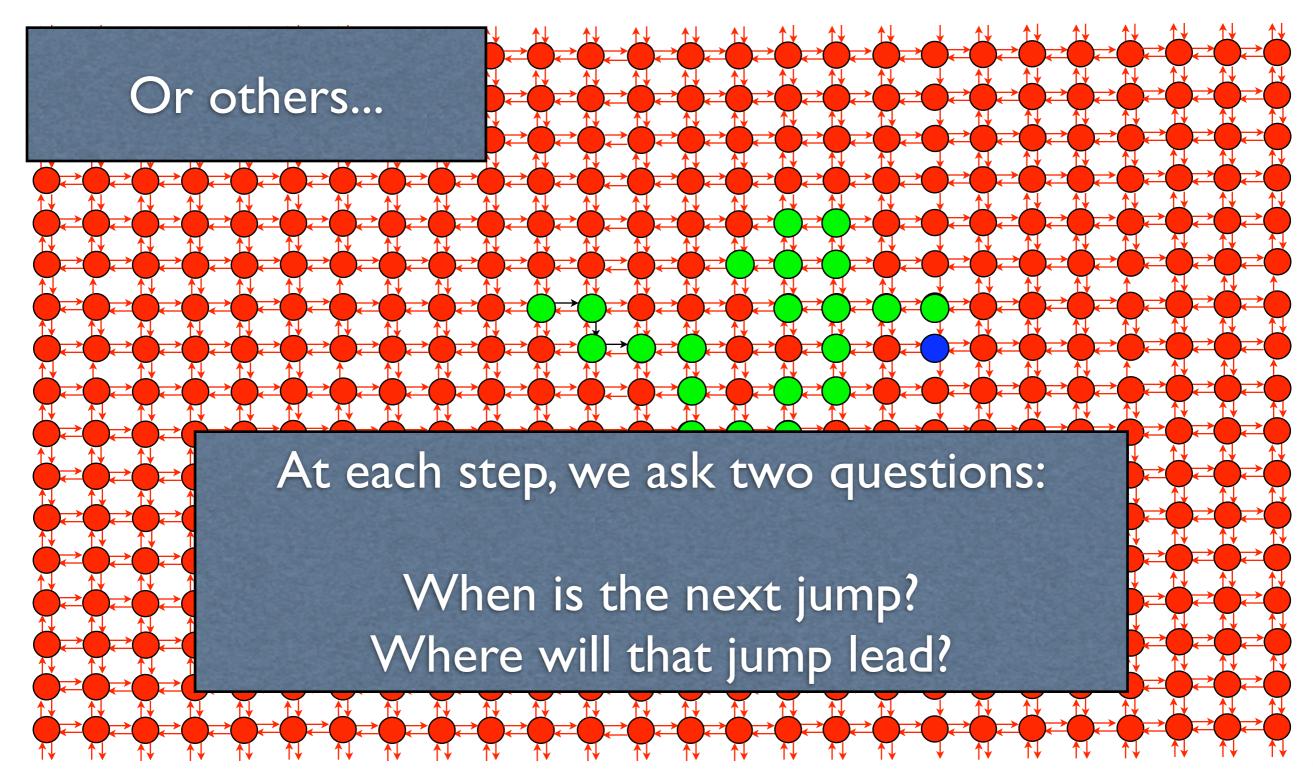












Reaction Stoichiometry (review)

- The Stoichiometric vector, **s**, refers to the relative change in the population vector after a reaction.
- There may be many different reactions for a given stoichiometry.

$$\mathbf{s}_1 = [1, 0]^T$$

$$\mathcal{S}_1 \to \mathcal{S}_1 + \mathcal{S}_1$$

$$\mathcal{S}_2 \to \mathcal{S}_2 + \mathcal{S}_1$$

$$\emptyset \to \mathcal{S}_1$$

$$\mathbf{s}_2 = [-1, 0]^T$$

$$\mathcal{S}_1 + \mathcal{S}_1 \to \mathcal{S}_1$$

$$\mathcal{S}_1 + \mathcal{S}_2 \to \mathcal{S}_2$$

$$\mathcal{S}_1 \to \emptyset$$

$$\mathbf{s}_3 = [0, 1]^T$$

$$\mathcal{S}_2 \to \mathcal{S}_2 + \mathcal{S}_2$$

$$\mathcal{S}_1 \to \mathcal{S}_1 + \mathcal{S}_2$$

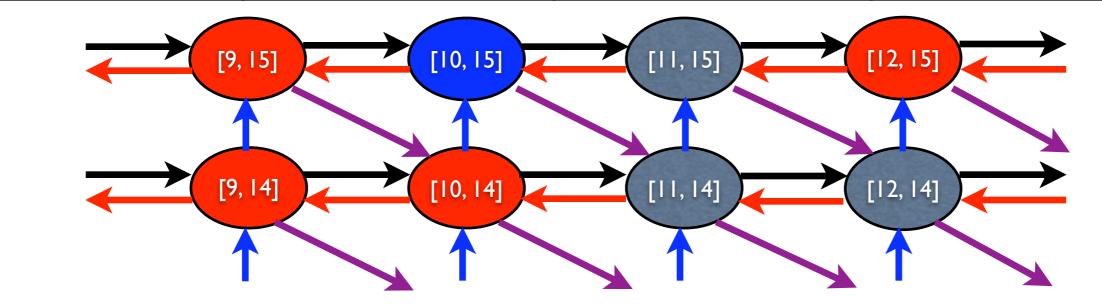
$$\emptyset \to \mathcal{S}_2$$

$$\mathbf{s}_4 = [1, -1]^T$$

$$\mathcal{S}_2 \to \mathcal{S}_1$$

$$\mathcal{S}_1 + \mathcal{S}_2 \to \mathcal{S}_1 + \mathcal{S}_1$$

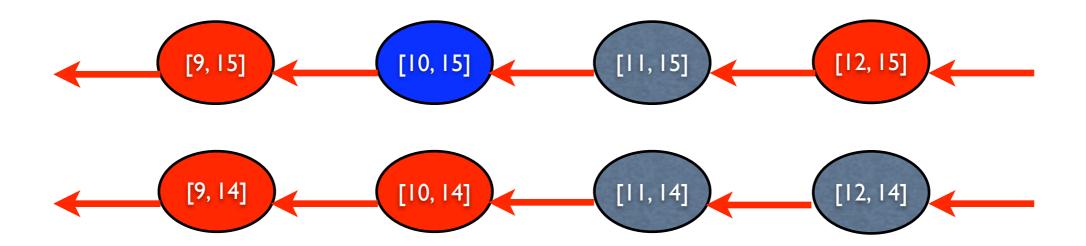
$$\mathcal{S}_2 + \mathcal{S}_2 \to \mathcal{S}_1 + \mathcal{S}_2$$



Reaction Propensities (review)

- The propensity, w, of a reaction is its rate.
- $\mathbf{w}_{\mu}dt$ is the probability that the μ^{th} reaction will occur in a time step of length dt .
- Typically, propensities depend only upon reactant populations.

$\mathbf{s}_2 = [-1, 0]^T$	$w_2(x_1, x_2)$
$\mathcal{S}_1 + \mathcal{S}_1 o \mathcal{S}_1$	$k_1 x_2 (x_1 - 1)/2$
$\mathcal{S}_1 + \mathcal{S}_2 ightarrow \mathcal{S}_2$	$k_{2}x_{1}x_{2}$
$\mathcal{S}_1 o \emptyset$	k_3x_1



Exponential Waiting Times

Probability reaction will occur in $[t, t + \Delta t]$:

$$w\Delta t + \mathcal{O}(\Delta t)^2$$

Probability reaction will not occur in $[t,t+\Delta t]$: $1-w\Delta t+\mathcal{O}(\Delta t)^2$

Probability a reaction will not occur in two such time

intervals
$$[t, t + 2\Delta t]$$
: $(1 - w\Delta t + \mathcal{O}(\Delta t)^2)^2 = 1 - 2w\Delta t + \mathcal{O}(\Delta t)^2$

Suppose that, $\tau = K\Delta t$, then the probability that no reaction will occur in the interval [t,t+ au) is

$$\left(1 - w\frac{\tau}{K} + \mathcal{O}(K^{-2})\right)^K$$

Taking the limit as K goes to infinity yields that the probability that no reaction will occur in the interval $[t,t_{\tau}+\tau)$ is

$$\lim_{k \to \infty} \left(1 - w \frac{\tau}{K} + \mathcal{O}(K^{-2}) \right)^{K} = \exp(-w\tau)$$

Exponential Random Variables

The probability that a reaction will occur in the interval $[t, t + \tau]$ is $F_T(\tau) = 1 - \exp(-w\tau)$. This is a cumulative distribution.

The density (derivative) of the random number, T, is:

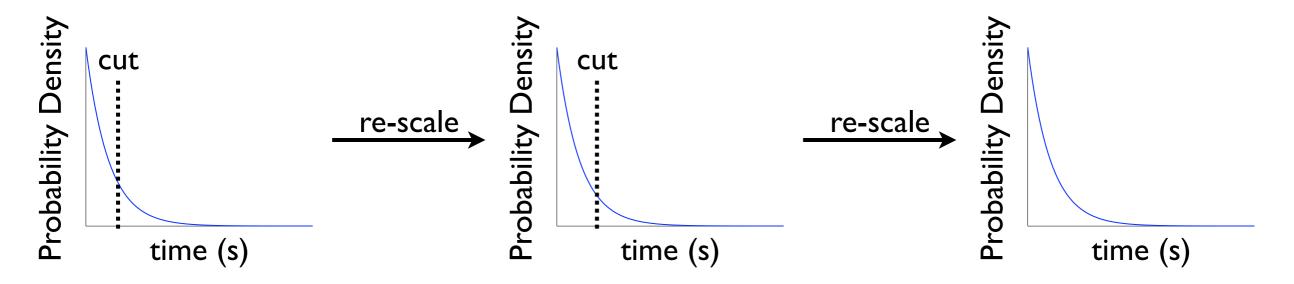
$$f_T(\tau) = \frac{1}{w} \exp(-w\tau)$$

Such a random number is known as an exponentially distributed random number.

Notation: $T \in \mathrm{EXP}(\lambda) \to T$ is an exponentially distributed r.v. with parameter: λ .

Exponential Waiting Times

- We have assumed that the system is fully described by the population vectors.
- If no reaction occurs, then nothing will have changed.
- Waiting times must be memoryless random variables.



 No matter where we cut and scale the distribution, it must always looks the same.

The exponential is the *only* continuous r.v. with this property.

Generating Waiting Times

- To generate an exponentially distributed random number, all we need is a uniform random number generator.
- Find the cumulative distribution,

$$F(t) = 1 - \exp(-\lambda t)$$

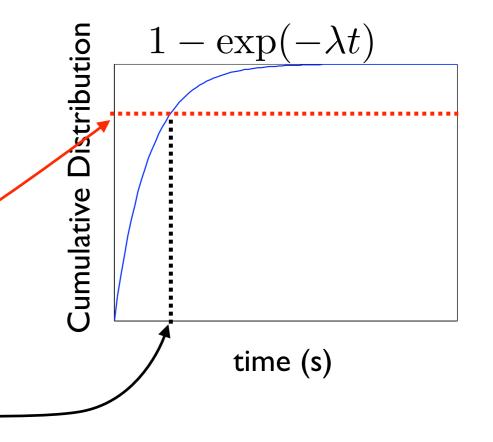
• Generate uniform random number,

$$r \in \mathrm{U}[0,1]^-$$

• Find intersection where F(t) = r:

$$\tau = \frac{1}{\lambda} \log \frac{1}{1 - r}$$

This is the time of the next reaction.

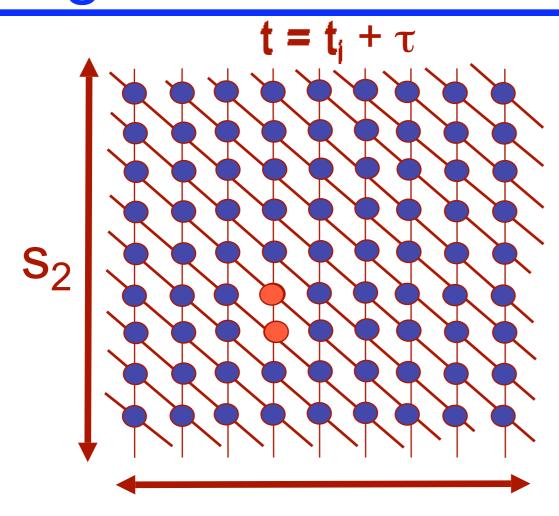


Monte-Carlo Simulation Methods

The Jump Markov Process

- Stochastic Simulation Algorithm
 - •D.T. Gillespie, J. Phys. Chem. A 81, 2340 (1977)
 - •M. Gibson and J. Bruck, J. Phys. Chem. **104**, 1876 (2000)

Stochastic Simulation Algorithm



Step 1. Generate the time of the next reaction.

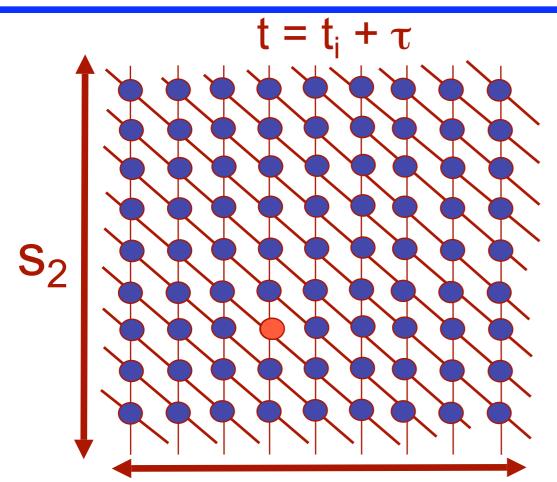
Step 2. Decide which reaction has occurred.

Step 3. Update current Time $(t=t+\tau)$ and State $(\mathbf{x}=\mathbf{x}+\mathbf{s_k})$.

Monte-Carlo Simulation Methods

- Stochastic Simulation Algorithm
 - •D.T. Gillespie, J. Phys. Chem. A 81, 2340 (1977)
 - •M. Gibson and J. Bruck, J. Phys. Chem. **104**, 1876 (2000)
- Possible SSA methods:
 - First Reaction Method (Gillespie '77)
 - Next Reaction Method (Gibson and Bruck '00)
 - Direct Method (Gillespie '77)

The First Reaction Method (FRM)



Step 1. Generate the time of the next reaction of each type.

The time until the next reaction is a random variable of exponential distribution:

$$P_{\tau_{\mu}}(t) = w_{\mu}(\mathbf{x})e^{-w_{\mu}(\mathbf{x})t}$$

To generate each next reaction time, generate r_1 from a uniform distribution on (0,1) and use the equation:

 $\tau_{\mu} = \frac{1}{w_{\mu}(\mathbf{x})} \log \frac{1}{r_{\mu}}$

Step 2. Decide which reaction has occurred.

This is simply the reaction with the smallest τ_{μ} :

$$k = \arg\left\{\min_{\mu \in \{0, \dots, M\}} \tau_{\mu}\right\}$$

Step 3. Update current Time (t=t+ τ_k) and State ($\mathbf{x} = \mathbf{x} + \mathbf{s_k}$).

The First Reaction Method SSA in Matlab.

```
clear all
t=0; t=0; t=0;
                                                    %%specify initial and final times
                                                    %% Specify initial conditions
x = [0; 0];
S = [1 -1 0 0; 0 0 1 -1];
                                                    %% Specify stoichiometry
w = inline('[10, 1*x(1), 10*x(1), 1*x(2)]', 'x');
                                                    %% Specify Propensity functions
while t<tstop
    tpos = 1./w(x).*log(1./rand(4,1));
                                                   % possible times until first reaction
    [tpos,i]=min(tpos);
                                                   % find which is first reaction
    t=t+tpos;
    if t<=t_stop</pre>
        x = x+S(:,i);
                                                   % update the configuration
    end
end
```

The Next Reaction Method (NRM)

- In the FRM, we generate times, $\{\tau_{\mu}\}$, for all M reactions and choose the reaction, k, with the smallest time, τ_k .
- Only a few species will change population as a result of this reaction--the rest will remain constant.
- For most reactions, the propensity functions will remain constant.
 - For these, the times can be reused in the subsequent step to find the next reaction: $\{\tau_{\mu}\} \rightarrow \{\tau_{\mu} \tau_{k}\}$.
- When there are many different species and reactions, this NRM approach can be done with far fewer random number than the FRM.
- Particularly useful for compartmental or Reaction-Diffusion processes.

Monte-Carlo Simulation Methods

- Stochastic Simulation Algorithm
 - •D.T. Gillespie, J. Phys. Chem. A 81, 2340 (1977)
 - •M. Gibson and J. Bruck, J. Phys. Chem. **104**, 1876 (2000)
- Possible SSA methods:
 - First Reaction Method (Gillespie '77)
 - Next Reaction Method (Gibson and Bruck '00)
 - Direct Method (Gillespie '77)

Minimum of two Exponential Random Variables

Let $\{\tau_1, \tau_2, \dots, \tau_M\}$ be a set of exponentially distributed random variables: $\tau_{\mu} \in \text{EXP}(w_{\mu})$

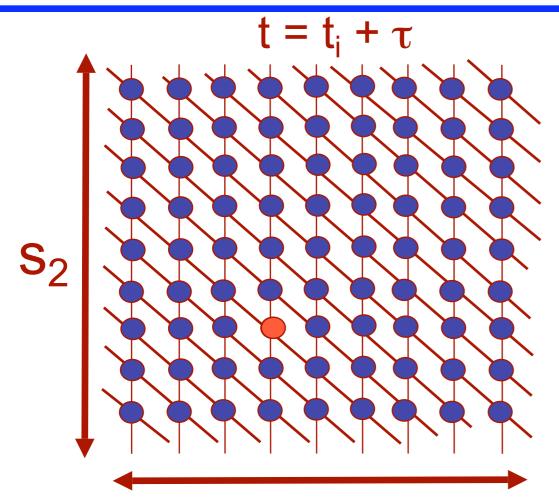
The minimum of $\{\tau_{\mu}\}$ is an exponentially distributed random variable given by:

$$\min_{\mu \in \{0,...,M\}} \tau_{\mu} \in \text{EXP}\left(|\mathbf{w}|_{1}\right)$$

The argument, k, of this distribution is also a random variable with distribution:

$$P(k=\mu) = \frac{w_{\mu}}{|\mathbf{w}|_{1}}$$

The Direct Method (DM)



Step 1. Generate the time of the next reaction.

The time until the next reaction is a random variable of exponential distribution:

$$P_{\tau}(t) = |\mathbf{w}(\mathbf{x})|_{1} e^{-|\mathbf{w}(\mathbf{x})|_{1}t}$$

To generate the next reaction time, generate r_1 from a uniform distribution on (0,1) and use the equation:

 $\tau = \frac{1}{|\mathbf{w}|_1} \log \frac{1}{r_1}$

Step 2. Decide which reaction has occurred.

To obtain a realization of which reaction will occur, generate a second uniform random number, r_2 , and find the smallest k such that:

$$\sum_{\mu=1}^{\kappa-1} w_{\mu}(\mathbf{x}) \le r_2 |\mathbf{w}|_1 \le \sum_{\mu=1}^{\kappa} w_{\mu}(\mathbf{x})$$

Step 3. Update current Time (t=t+ τ) and State ($\mathbf{x} = \mathbf{x} + \mathbf{s_k}$).

The Direct Method SSA in Matlab.

```
clear all
t=0; t=0; t=0;
                                                    %%specify initial and final times
x = [0; 0];
                                                    %% Specify initial conditions
S = [1 -1 0 0; 0 0 1 -1];
                                                    %% Specify stoichiometry
w = inline('[10, 1*x(1), 10*x(1), 1*x(2)]', 'x');
                                                   %% Specify Propensity functions
while t<tstop
    w0 = sum(w(x));
                                                   % compute the sum of the prop. functions
    t = t+1/w0*log(1/rand);
                                                   % update time of next reaction
    if t<=t_stop
    r2w0=rand*w0;
                                % generate second random number and multiply by prop. sum
                                                   % initialize reaction counter
    i=1;
    while sum(w(1:i)) < r2w0
                                        % increment counter until sum(w(1:i)) exceeds r2w0
      i=i+1;
    end
    x = x+S(:,i);
                                                   % update the configuration
  end
end
```

Limitations on the SSA

- The SSA is an "exact" simulation of the system.
- But...
 - Stepping through every reaction can take a lot of time.
 - A statistical representation of the system dynamics may require many realizations (10⁴ to 10⁶).
- Faster approximations are available for some problems.

Monte-Carlo Simulation Methods

- Stochastic Simulation Algorithm (SSA).
- τ-leaping
 - •D. Gillespie, J. Chem. Phys. **115**, 1716 (2001)
 - •D. Gillespie, L. Petzold, J. Chem. Phys. **119**, 8229 (2003)
 - •M. Rathinam et al., J. Chem. Phys. 119, 12784 (2003)
 - •T. Tian and K. Burrage, J. Chem. Phys. **121**, 10356 (2004)
 - •Y. Cao, D. Gillespie and L. Petzold, J. Chem. Phys. **123**, 054104 (2005)

τ Leaping

Step 0. Specify length of each time step, τ.

Assume that all propensity functions are constant over the time interval $(t,t+\tau)$.

 $\mu=1$

The number of times each reaction will fire is a Poisson* random number with mean $\mathbf{w}_{\mu}\tau$:

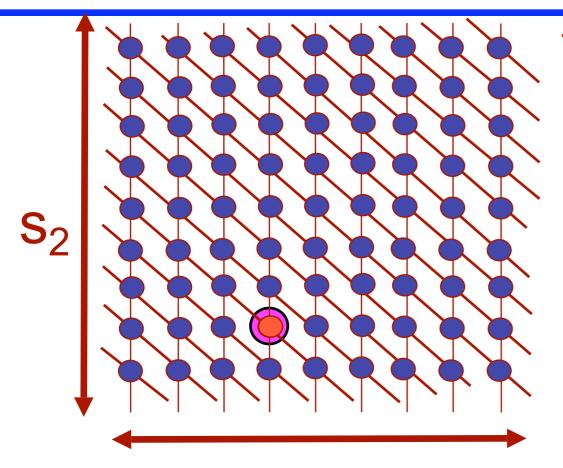
Poisson* random number with mean
$$\mathbf{w}_{\mu} \mathbf{\tau}$$
:
$$P_{k_{\mu}}(n) = \frac{[w_{\mu}(\mathbf{x})\tau]^n}{n!} \mathrm{e}^{w_{\mu}(\mathbf{x})\tau}$$

Step 1. For each μ , generate k_{μ} .

Step 2. Update the time:
$$t=t+ au_M$$
 Update the state: $\mathbf{x}=\mathbf{x}+\sum k_{\mu}\mathbf{s}_{\mu}$

*For some recent studies, binomial RV's are used (T. Tian and K. Burrage, 2004)

τ Leaping



$t = t_i + \tau$ Update Time

$$k_1 = 4$$
; $\mathbf{s}_1 = [0, 1]^T$
 $k_2 = 2$; $\mathbf{s}_1 = [-1, 1]^T$
 $k_3 = 3$; $\mathbf{s}_1 = [0, -1]^T$
 $k_4 = 4$; $\mathbf{s}_1 = [1, -1]^T$

The number of times each reaction will fire is a Poisson random number with mean $\mathbf{W}_{\mu}\mathbf{\tau}$: $P_{k_{\mu}}(n) = \frac{[w_{\mu}(\mathbf{x})\tau]^n}{n!} e^{w_{\mu}(\mathbf{x})\tau}$

Step 1. For each μ , generate k_{μ} . M

Step 2. Update the state: $\mathbf{x} = \mathbf{x} + \sum_{\mu=1}^{\infty} k_{\mu} \mathbf{s}_{\mu}$

Update the time: $t = t + \tau$

Limitations of τ leaping

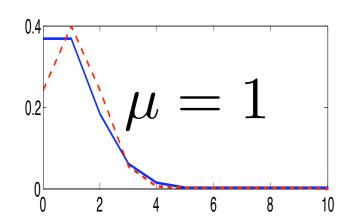
- For many situations τ leaping significantly speeds up the Monte Carlo simulation, but:
 - Poisson r.v.'s are unbounded
 - Propensity functions may change dramatically over small time intervals.
 - May result in negative populations.

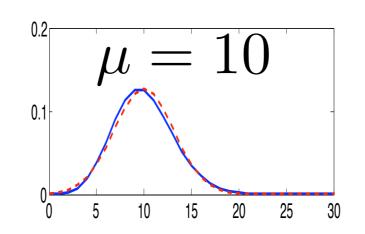
Note that these concerns are most important when the population of some species are very small.

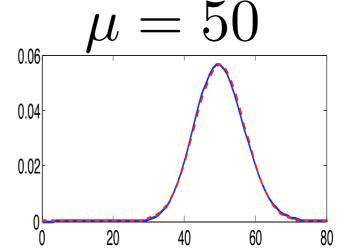
Precisely the circumstance where stochastic models are most important!

Chemical Langevin Equation

Comparison of Poisson and Gaussian random variables.





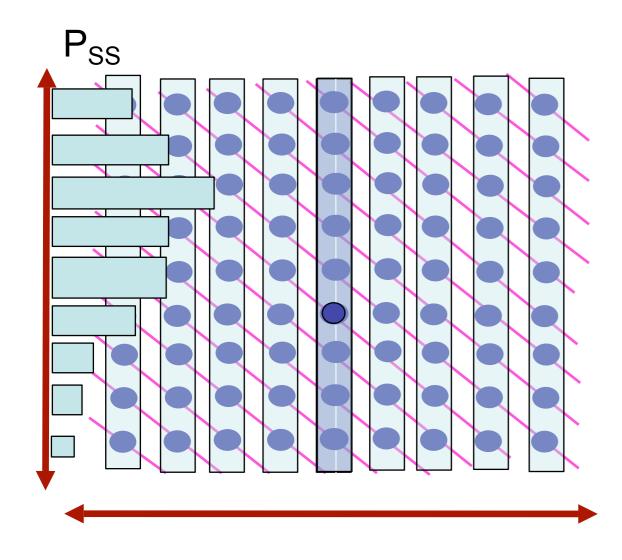


- For small numbers of reaction steps, tau leaping doesn't give much help.
- For large numbers of reactions, replace the Poisson distribution with a normal distribution (same mean and variance. These are cheaper to generate.
- This is known as the chemical Langevin equation.

Monte-Carlo Simulation Methods

- Stochastic Simulation Algorithm (SSA).
- τ-leaping
- System Partitioning Methods
 - Fast--Slow Partitions
 - •C. Rao and A. Arkin, J. Chem. Phys. **118**, 4999 (2003)
 - •Y. Cao et al., J. Chem. Phys. 122, 014116 (2005)
 - Continuous--Discrete Partitions
 - •E. Haseltine and J. Rawlings, J. Chem. Phys. **117**, 6959 (2002)
 - •H. Salis and Y. Kaznessis, J. Chem. Phys. **122**, 054103 (2005)

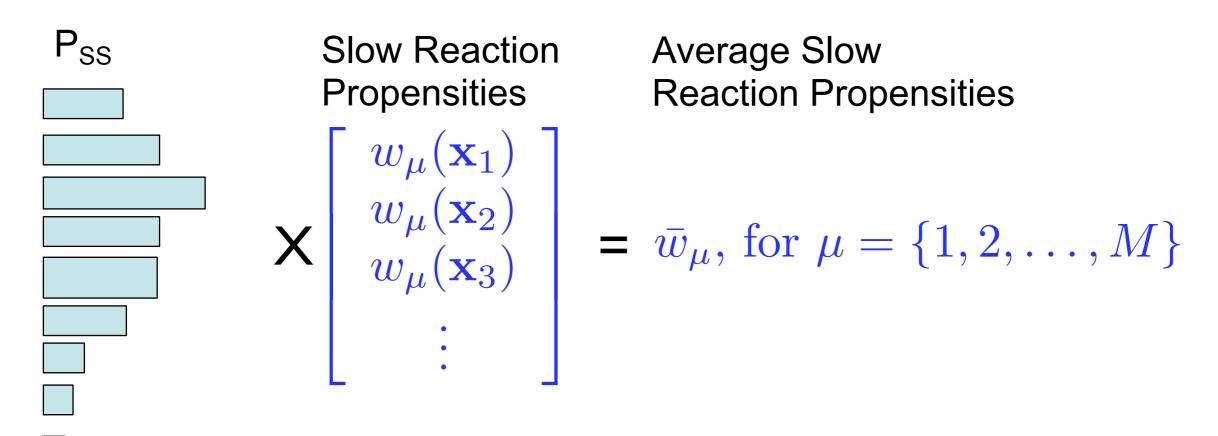
Fast--Slow partitions.



Separate into "fast" and "slow" partitions.

Assume that the "fast" partitions reach probabilistic equilibrium before a slow reaction occurs.

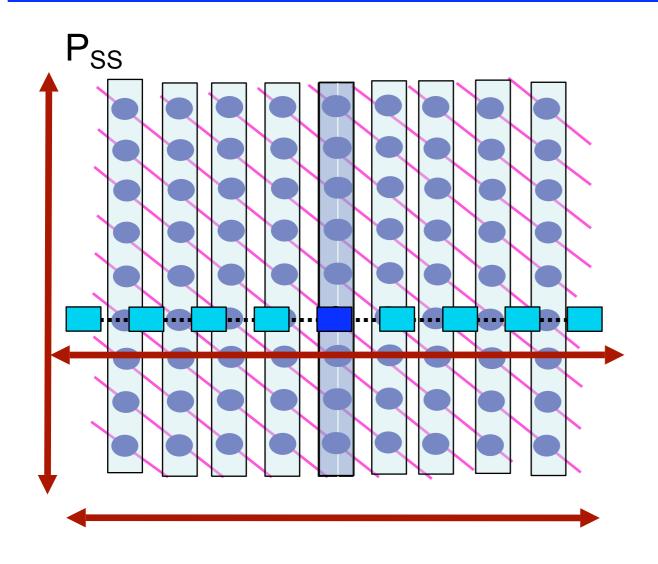
Fast--Slow partitions.



Use the fast sets' steady state probability distributions to scale the propensity functions of the slow reactions.

Results in a vector of average propensity functions, $\overline{\mathbf{w}}$, for the slow reactions.

Fast--Slow partitions.



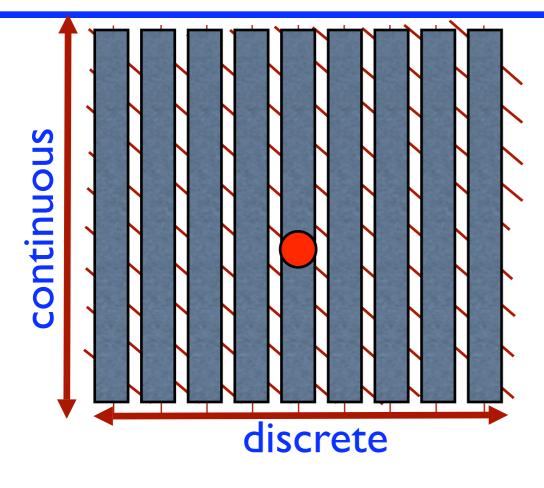
The projection to the slow manifold results in a new lower dimensional Markov chain.

This is simulated with SSA.

Continuous--Discrete partitions.

- In some systems, there are great differences in scale:
 - Large populations (continuous)
 - Small populations (discrete)
- All discrete models take too long.
- All continuous models are inaccurate.
- Hybrid models are necessary.

Separate into "continuous" and "discrete" partitions.



Simulate the continuous part with ordinary or stochastic differential equations.

Choose uniform rv, r.

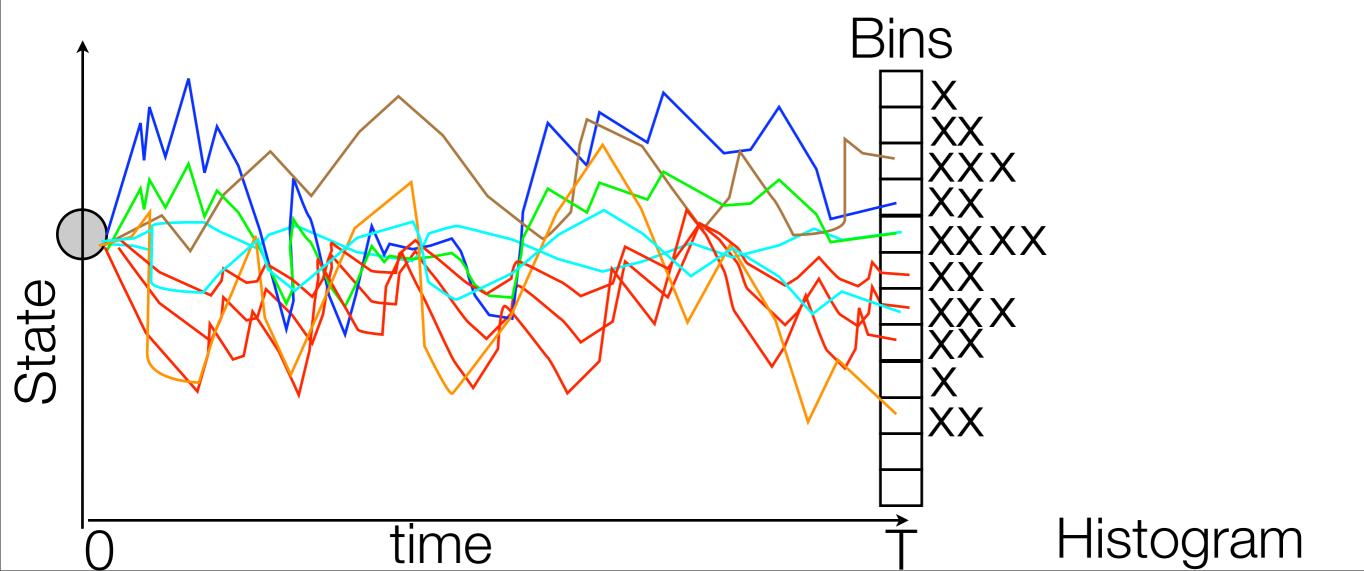
Numerically integrate propensity functions until:

$$\int_{t_0}^{t_0+\tau} \sum_{\mu=1}^{M} w_{\mu}(\mathbf{x}(t))dt = -\log r$$

Choose next discrete reaction.

Using the SSA to Find Distributions

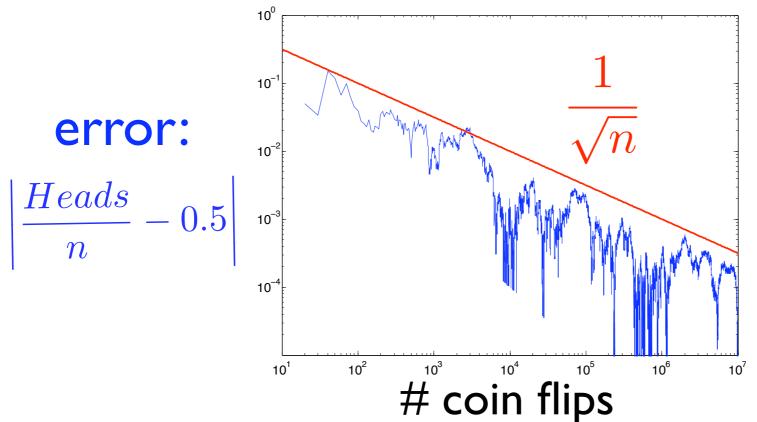
 The SSA does an excellent job of producing possible trajectories.



Convergence of the SSA

- To get more accurate distributions, one needs more SSA runs.
- Unfortunately, the convergence rate of any Monte Carlo algorithm is fundamentally limited: $error = \mathcal{O}(n^{-\frac{1}{2}})$
- If very high precision is required, then MC methods will be very inefficient.

Convergence for Coin Toss



After 10^7 tosses there is still an error of about 3×10^{-4} .

Density Computations

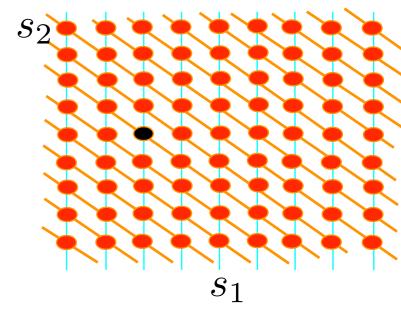
The Finite State Projection (FSP) solution to the Chemical Master Equation.

Reductions to the FSP

Case studies.

The Chemical Master Equation

The probability that the system is in configuration \mathbf{x} at t+dt is equal to the probability that the system is at \mathbf{x} at t, and no reaction occurs between t and t+dt plus the probability that the system is one reaction removed from \mathbf{x} at t and that reaction occurs between t and t+dt.



The CME (McQuarrie '67):

$$\dot{p}(\mathbf{x},t) = -p(\mathbf{x},t) \sum_{k=1}^{M} w_k(\mathbf{x}) + \sum_{k=1}^{M} p(\mathbf{x} - \mathbf{s}_k, t) w_k(\mathbf{x} - \mathbf{s}_k)$$

Define the probability density state

$$\text{vector (pdv):} \quad \mathbf{P}(\mathbf{X},t) := [p(\mathbf{x}_1,t),p(\mathbf{x}_2,t),p(\mathbf{x}_3,t),\ldots]^T$$

 $\mathbf{P}(\mathbf{X},t)$ evolves according to the Linear Time Invariant ODE:

$$\dot{\mathbf{P}}(\mathbf{X},t) = \mathbf{A} \cdot \mathbf{P}(\mathbf{X},t)$$

The matrix CME

The Chemical Master Equation

The solution of the CME is a transfer operator:

$$\mathcal{P}(t_0) \longrightarrow \mathbb{CME} \longrightarrow \mathcal{P}(t_0 + \tau)$$

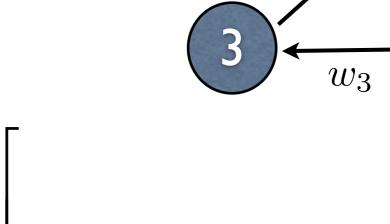
- The dimension of the CME can be INFINITE.
 - Most CME's cannot be solved, so approximations are needed.

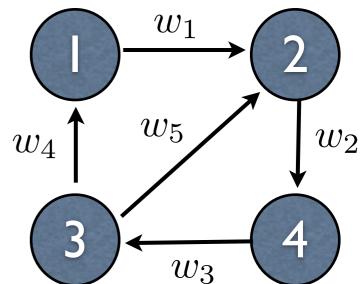
Forming the Generator

A has one row/column for each state.

Each transition, $\mathbf{x}_i \to \mathbf{x}_j$, contributes to \mathbf{A} in two locations:

 $-w_{\mu}(\mathbf{x}_i)$ goes in the diagonal element $A_{i,i}$ $+w_{\mu}(\mathbf{x}_i)$ goes in the off-diagonal element $A_{j,i}$





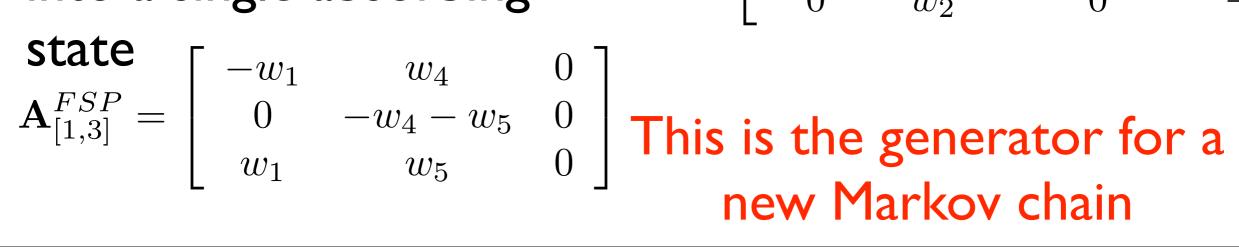
The Finite State Projection

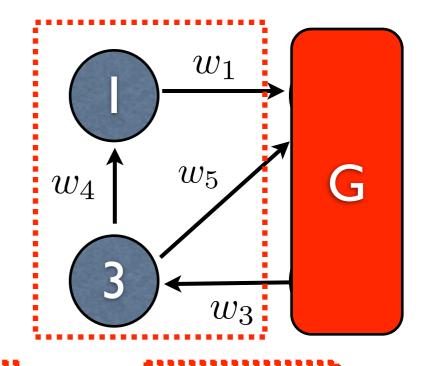
Select the states to keep.

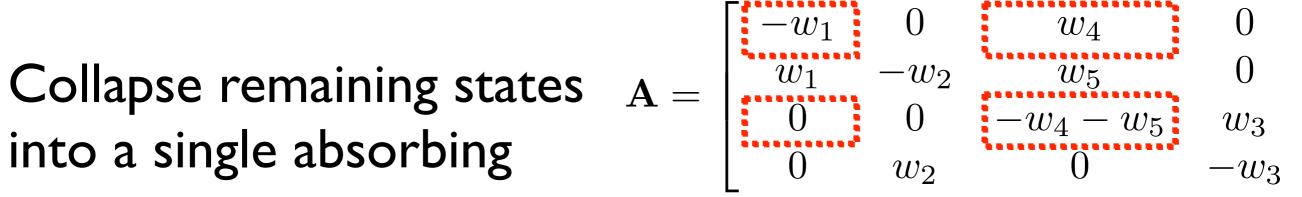
Find the corresponding

projection matrix:
$$\mathbf{A}_{[1,3]} = \begin{bmatrix} -w_1 & w_4 \\ 0 & -w_4 - w_5 \end{bmatrix}$$



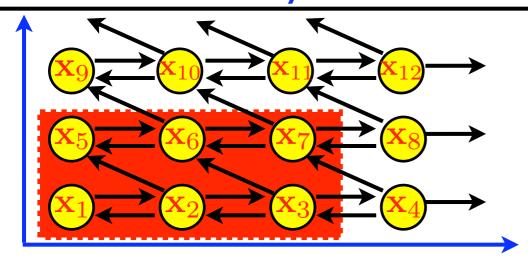






The Finite State **Projection Method**

The Full System

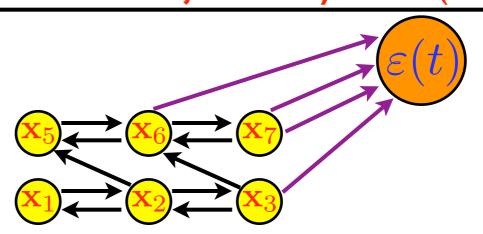


Full Master Equation

$$\begin{bmatrix} \dot{\mathbf{P}}_{J} \\ \dot{\mathbf{P}}_{J'} \end{bmatrix} = \begin{bmatrix} \mathbf{A}_{J} & \mathbf{A}_{JJ'} \\ \mathbf{A}_{J'J} & \mathbf{A}_{J'} \end{bmatrix} \begin{bmatrix} \mathbf{P}_{J}(t) \\ \mathbf{P}_{J'}(t) \end{bmatrix}$$

Dimension = #(J) + #(J') = Infinite

The Projected System (FSP)



FSP Master Equation

$$\begin{bmatrix} \dot{\mathbf{P}}_{J} \\ \dot{\mathbf{P}}_{J'} \end{bmatrix} = \begin{bmatrix} \mathbf{A}_{J} & \mathbf{A}_{JJ'} \\ \mathbf{A}_{J'J} & \mathbf{A}_{J'} \end{bmatrix} \begin{bmatrix} \mathbf{P}_{J}(t) \\ \mathbf{P}_{J'}(t) \end{bmatrix} \begin{bmatrix} \dot{\mathbf{P}}_{J}^{FSP} \\ \dot{\varepsilon} \end{bmatrix} = \begin{bmatrix} \mathbf{A}_{J} & \mathbf{0} \\ -\mathbf{1}^{T} \mathbf{A}_{J} & 0 \end{bmatrix} \begin{bmatrix} \mathbf{P}_{J}^{FSP}(t) \\ \varepsilon(t) \end{bmatrix}$$

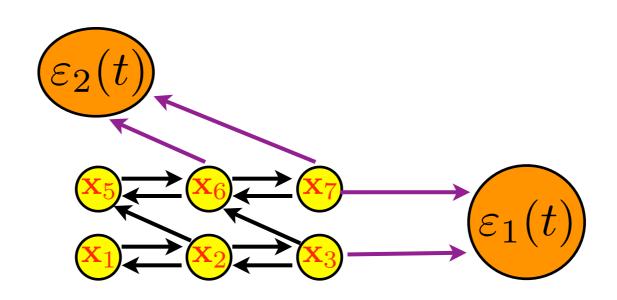
Dimension = #(J) + 1 = 7

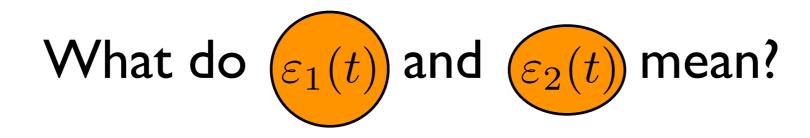
The FSP Theorem

(Munsky/Khammash JCP '06)

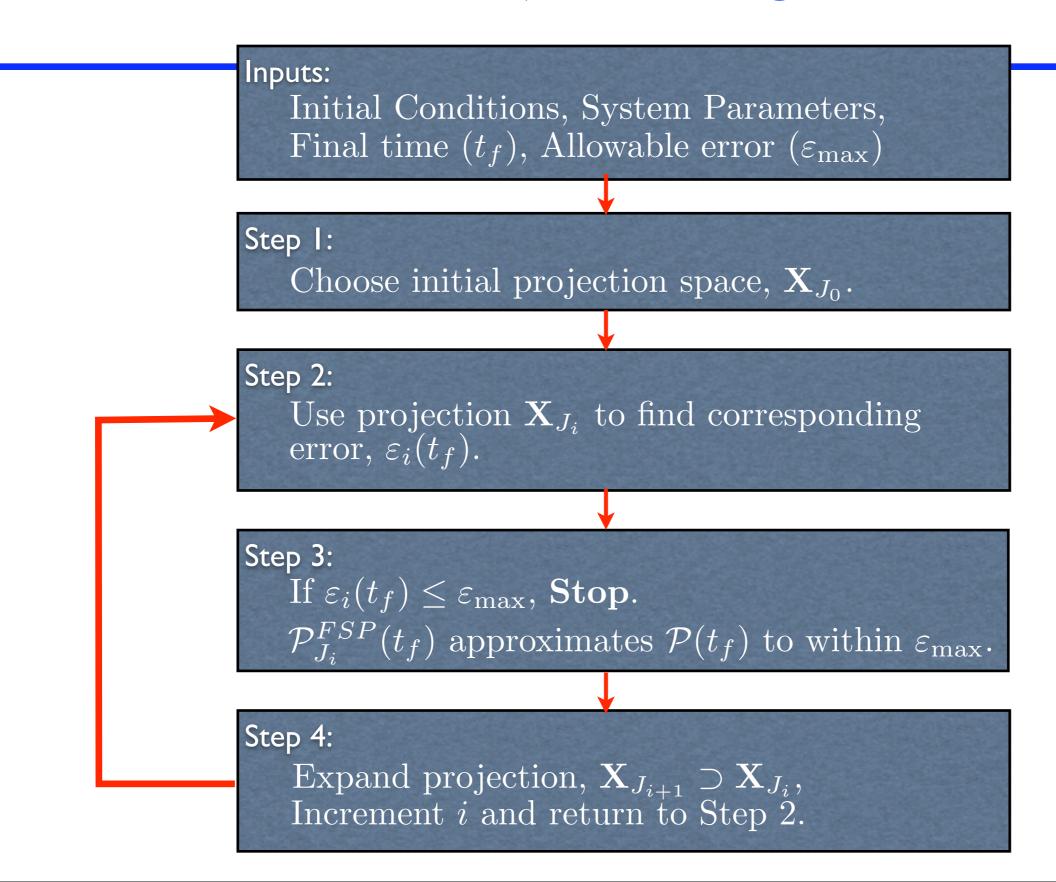
$$\left\| \begin{bmatrix} \mathbf{P}_{J}(t) \geq \mathbf{P}_{J}^{FSP}(t) \text{ and} \\ \begin{bmatrix} \mathbf{P}_{J}(t) \\ \mathbf{P}_{J'} \end{bmatrix} - \begin{bmatrix} \mathbf{P}_{J}^{FSP}(t) \\ \mathbf{0} \end{bmatrix} \right\|_{1} = \varepsilon(t)$$

A Test...

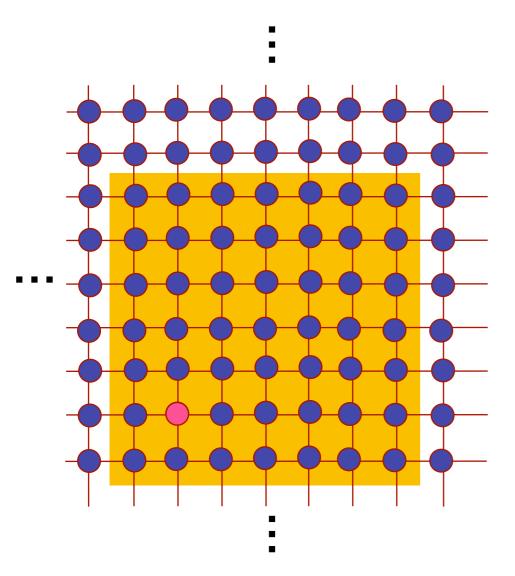




The Finite State Projection Algorithm



The FSP Algorithm



Begin with initial conditions, process parameters, and error tolerance. Choose an initial set: X_{J_0} .

Find
$$\varepsilon_i$$
;
If $\varepsilon_i < \varepsilon_{max}$, STOP.

Otherwise add more configurations to get $X_{J_{i+1}}$.

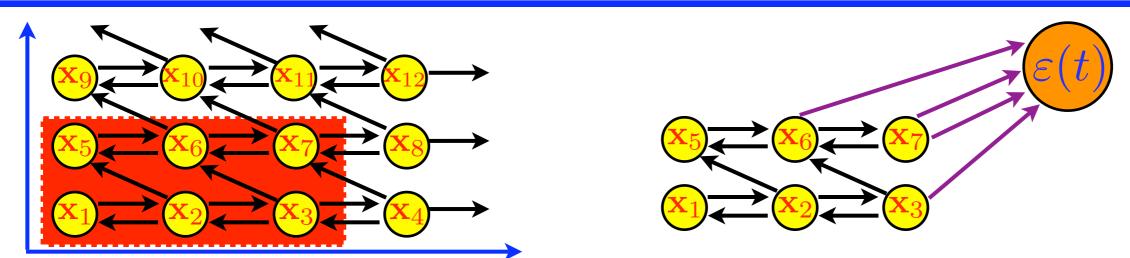
$$\varepsilon_0 > \varepsilon_{\text{max}}$$
 $\varepsilon_2 > \varepsilon_{\text{max}}$ $\varepsilon_4 > \varepsilon_{\text{max}}$

$$\varepsilon_1 > \varepsilon_{\text{max}}$$
 $\varepsilon_3 > \varepsilon_{\text{max}}$ $\varepsilon_5 < \varepsilon_{\text{max}}$ STOP

$$\varepsilon_5 < \varepsilon_{\text{max}}$$



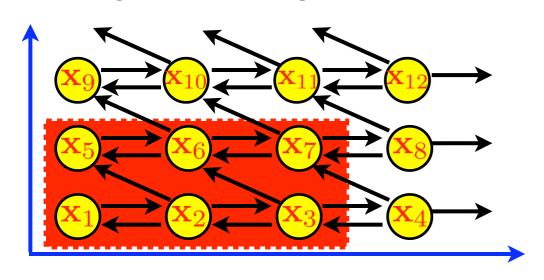
The "error" sink of the FSP to get exit times.

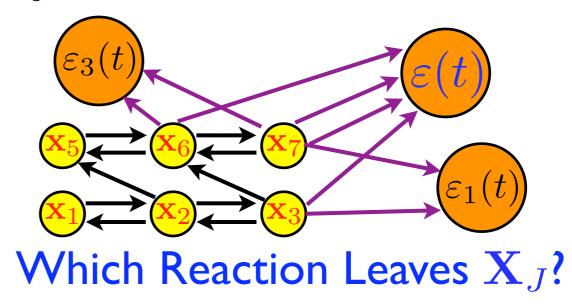


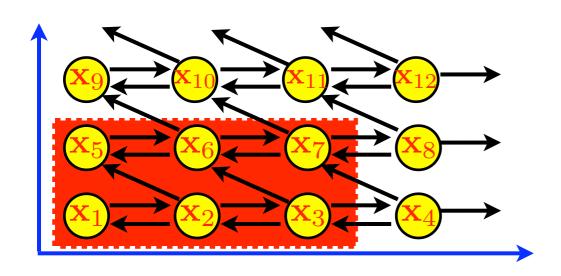
- In the original FSP, $\varepsilon(t)$ is the amount of the probability measure that exits the projection region \mathbf{X}_J .
- Median exit time: $t_{50} = t$, s.t. $\varepsilon(t) = 0.5$
- In this form $\varepsilon(t)$ gives information as to when the system exits \mathbf{X}_J , but not how.

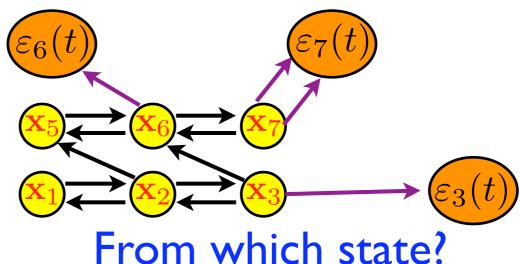
Multiple FSP sinks to get exit directions.

By using multiple sinks, one can determine how the probability measure exits X_{J} .



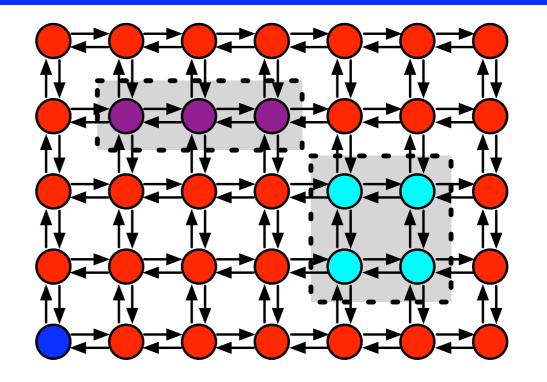


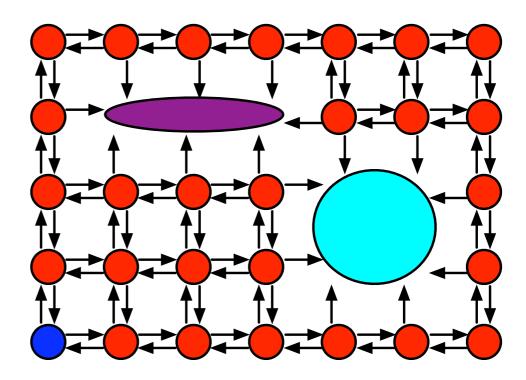




Multiple FSP sinks to analyze switch decisions

Using the FSP to determine initial switch decisions.





Advantages of the FSP.

- Deterministic.
 - * Every run of the FSP yields the same result.
 - * Enables easier comparisons of different systems (sensitivity analysis).
- Provides accuracy guarantees.
 - * Can be made as precise as required.
 - * Allows for analysis of rare events.
- Does not depend upon initial conditions.
- Is open to many subsequent model reductions.

Limitations

- Numerical stiffness may lead to computational inefficiency.
- Systems may become very large as distributions cover large regions of the configuration space.
 - * Compact distributions may drift over time.
 - ★ Dilute distributions may spread over large regions.
 - ★ Dimension grows exponentially with the number of species.
- For these problems, the original FSP may not suffice,
- BUT, with additional model reductions and systematic techniques, many of these problems may be alleviated.

Outline

- Finite State Projection (FSP)
- Reductions to the FSP
 - ★ Aggregating unobservable states
 Munsky/Khammash, CDC, 2006
 - ★ Time interval discretization
 - ★ Slow manifold projection
 - ★ Coarse meshes for the CME

Using Input & Output relations for model reduction.

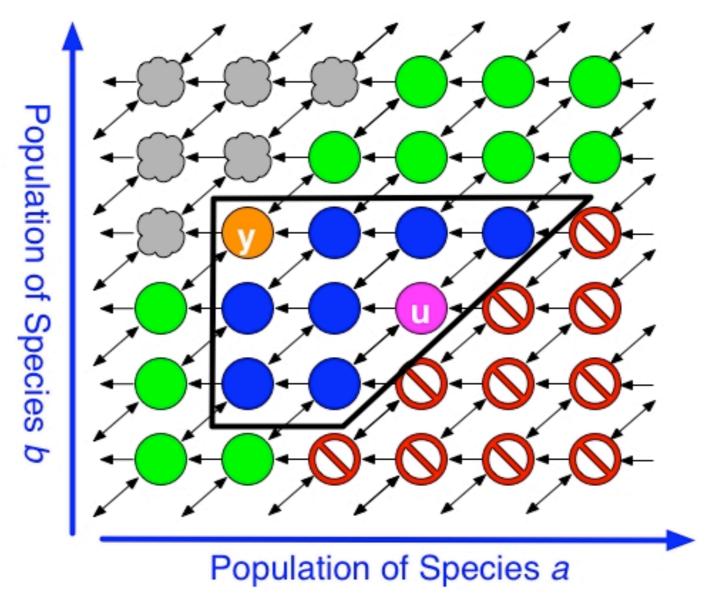
- Often one is not interested in the entire probability distribution.
- Instead one may wish only to estimate:
 - * a statistical summary of the distribution, e.g.
 - means, variances, or higher moments
 - probability of certain traits:
 - → switch rate, extinction, specific trajectories, etc...
- In each of these cases, one can define an output y(t):

$$\dot{\mathbf{P}}(t) = \mathbf{AP}(t)$$

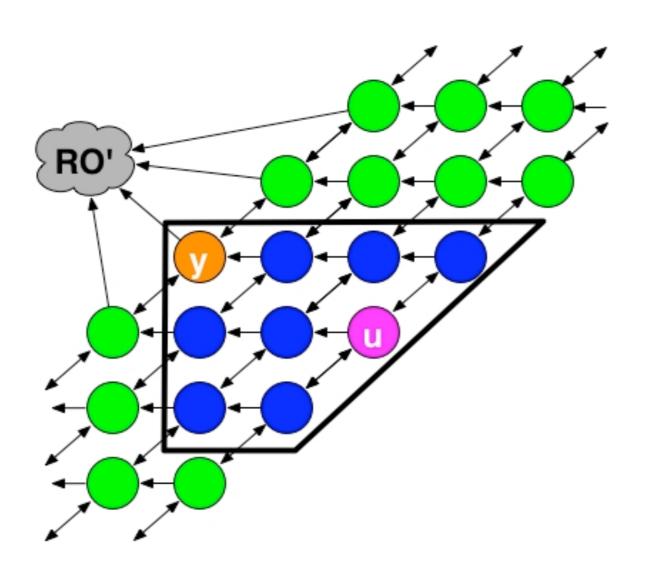
 $\mathbf{y}(t) = \mathbf{CP}(t)$

Begin with a Full Integer Lattice Description of the System States.

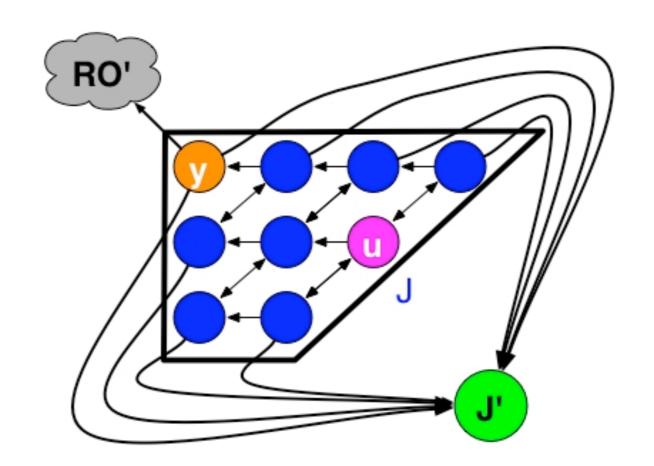
- u Initial State
- Observed State
- O Unreachable States (R')
- Unobservable State {O'}
- Reachable/
 Observable
 States (RO)



Remove Unreachable States and Aggregate the Observable States.



Project the Reachable/Observable States onto a Finite Subspace.

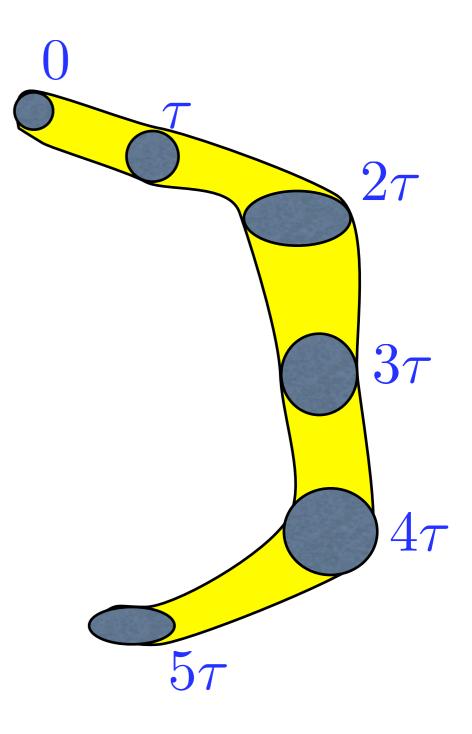


We now have a solvable approximation, for which the FSP gives bounds on the approximation's accuracy.

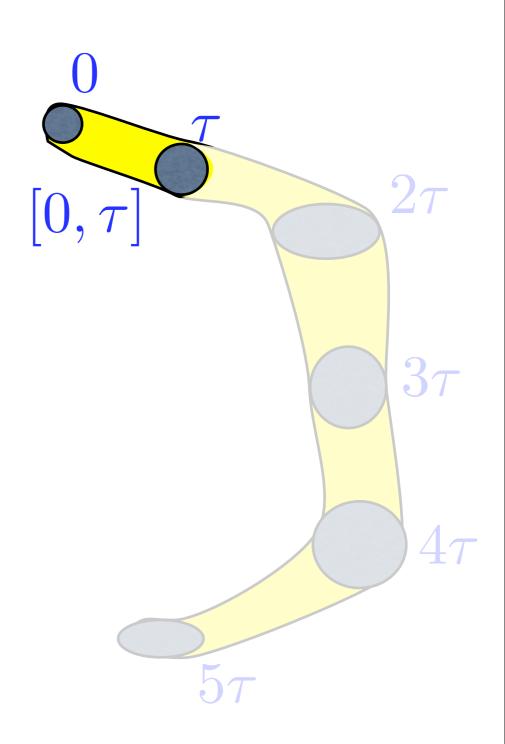
Outline

- Finite State Projection (FSP)
- Reductions to the FSP
 - ★ Aggregating unobservable states
 - ★ Time interval discretization
 Munsky and Khammash, J. Comp. Phys., 2007
 Burrage et al, A.A. Markov 150th Anniv. Meeting, 2006
 - ★ Slow manifold projection
 - ★ Coarse meshes for the CME

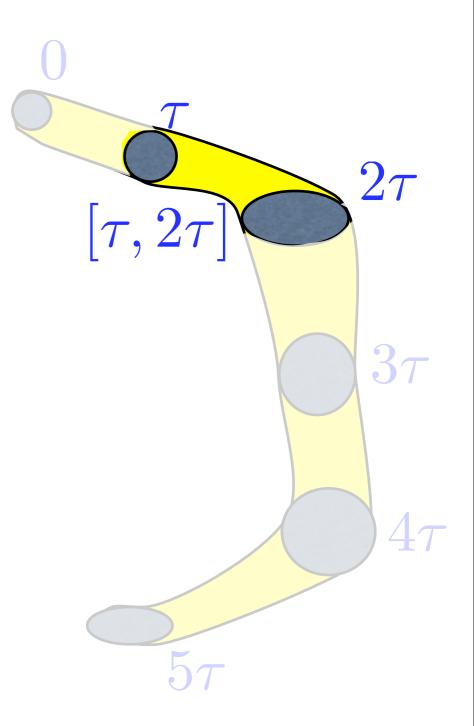
- ★ For many systems, the distribution may drift over time.
- ★ At any one time, the distribution may have a limited support, but...
- ★ The FSP solution must include all intermediate configurations.
- ★ This may lead to an exorbitantly large system of ODEs.



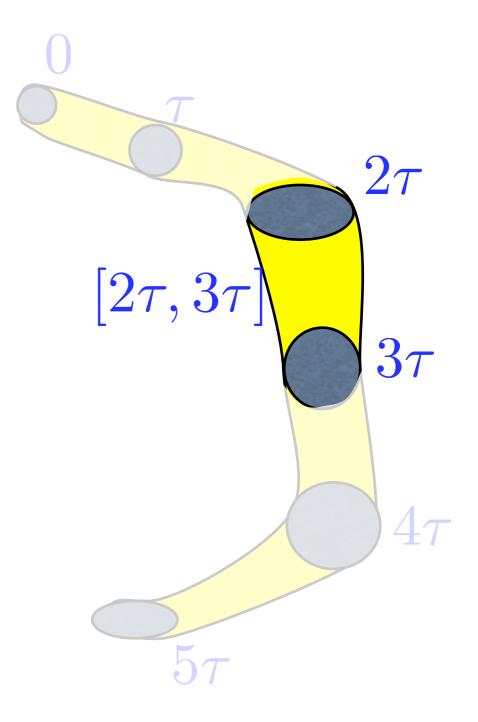
★ Instead:



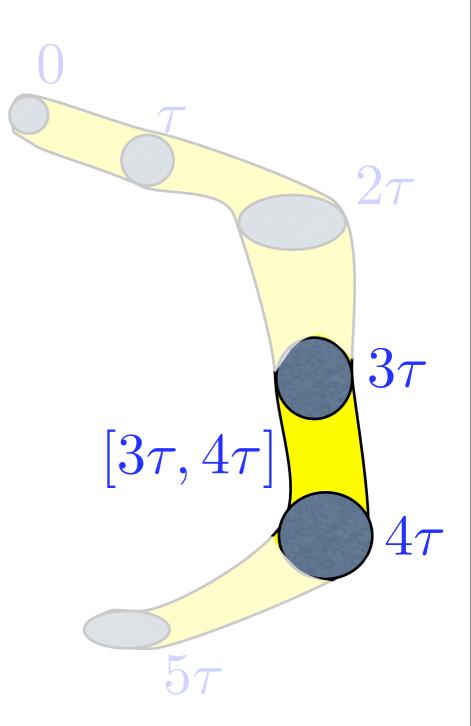
★ Instead:



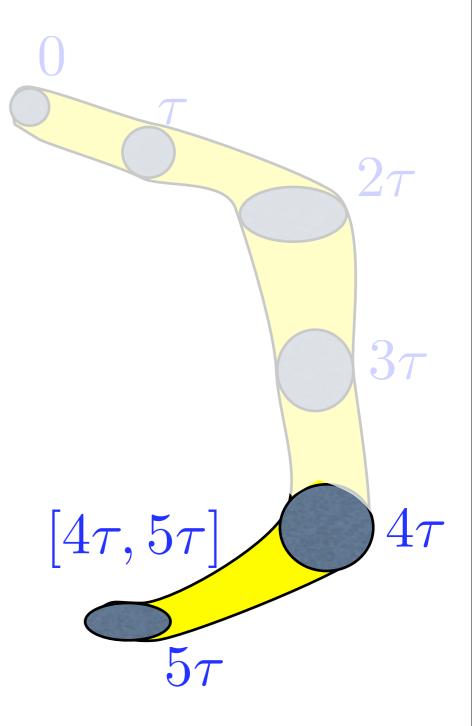
★ Instead:



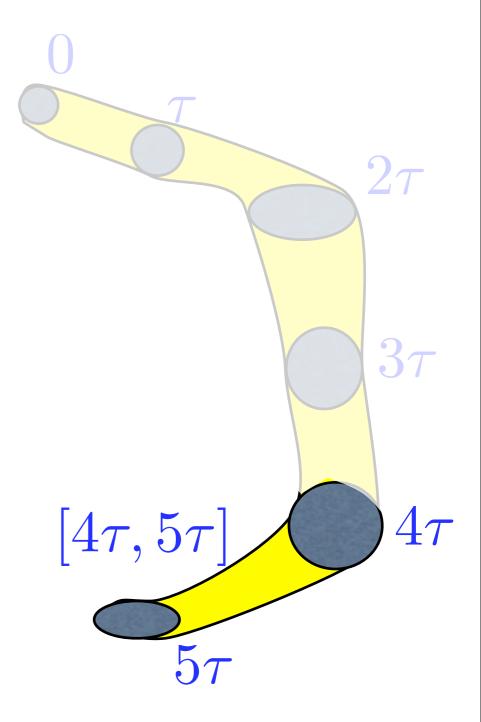
★ Instead:



★ Instead:



- ★ Solving a few smaller systems can be much easier than solving a single large system.
- ★ Control the error at each step to obtain a guaranteed final error.
- ★ Caching and reusing information from one step to the next may further reduce effort.



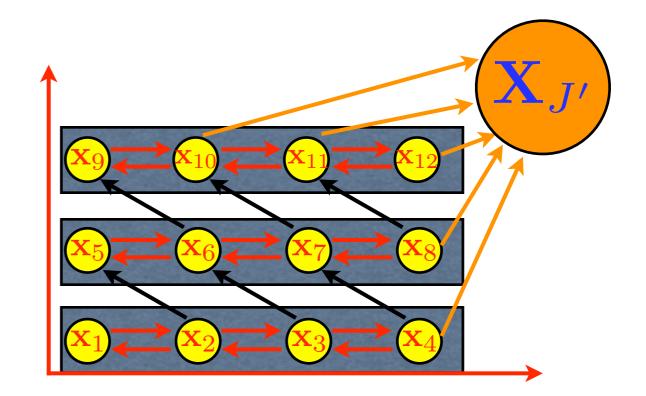
Outline

- Finite State Projection (FSP)
- Reductions to the FSP
 - ★ Aggregating unobservable states
 - ★ Time interval discretization
 - ★ Slow manifold projection
 Peles/Munsky/Khammash, JCP, 2006
 - ★ Coarse meshes for the CME.

Perturbation Theory and the FSP

- Some reactions occur faster and more frequently than others.
- This can result in a separation of time-scales in the CME.
 - Disadvantages: Often results in numerical stiffness and increased computational complexity.
 - Advantage: May be able to apply perturbation theory to reduce computational effort.

- Begin with a finite state (projected) Markov process.
- 2. Group states connected by frequent reactions.

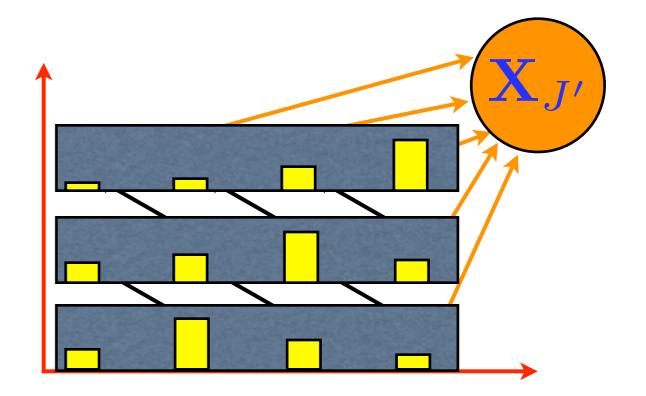


Red Arrows = Fast (Frequent) Reactions

Black Arrows = Slow (Rare) Reactions

Orange Arrows = (Rare) Transitions to Sink

- Begin with a finite state (projected) Markov process.
- 2. Group states connected by frequent reactions.
- 3. Find invariant distribution for each group.



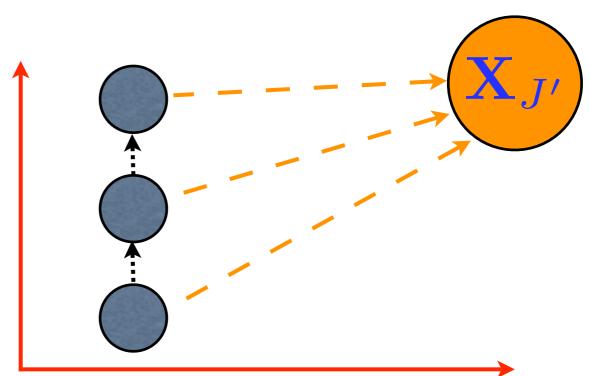
Red Arrows = Fast (Frequent) Reactions

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- Begin with a finite state (projected) Markov process.
- 2. Group states connected by frequent reactions.
- 3. Find invariant distribution for each group.
- 4. Average to find the rates of the slow reactions.

Reduced Markov Process

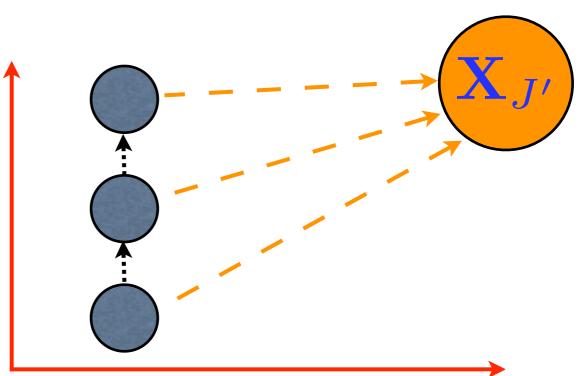


Dotted Black = Averaged Slow Reactions

Dashed Orange = Averaged Transitions to Sink

- Begin with a finite state (projected) Markov process.
- 2. Group states connected by frequent reactions.
- 3. Find invariant distribution for each group.
- 4. Average to find the rates of the slow reactions.

Reduced Markov Process



Dotted Black = Averaged Slow Reactions

Dashed Orange = Averaged Transitions to Sink

- 5. Solve for the solution on the slow-manifold.
- 6. Lift solution to original coordinate system.

Outline

- Finite State Projection (FSP)
- Reductions to the FSP
 - ★ Aggregating unobservable states
 - ★ Time interval discretization
 - ★ Slow manifold projection
 - ★ Coarse meshes for the CME

Munsky/Khammash, IEEE Trans. on Auto. Conrol, 2008

Coarse mesh approximation of the CME

- Precision requirements may change for different regions of the configurations space.
 - * Small populations require great precision.
 - * High populations require far less precision.
- By choosing a good coarse approximation of the CME, we can take advantage of this.

Coarse mesh approximation of the CME

Start with the full I-dimensional Markov lattice.

Choose a subset of mesh points.

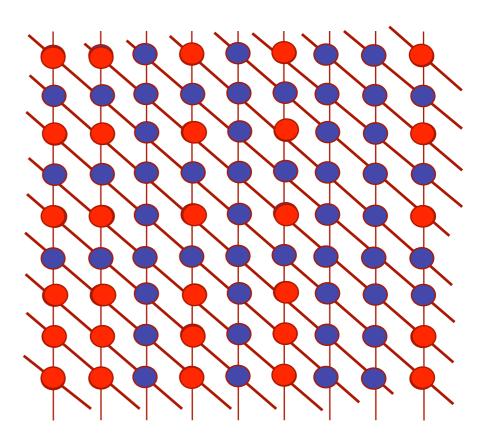
and specify an approximate relation for the probability of the removed points:
$$\mathbf{P} \approx \mathbf{\Phi} \mathbf{q}(t)$$

Solve the reduced system ODE: $\dot{\mathbf{q}} = \Phi^{-L} \mathbf{A} \Phi \mathbf{q}(t)$ and lift back to the original system coordinates:

$$\mathbf{P}(t) \approx \mathbf{\Phi} \exp(\mathbf{\Phi}^{-L} \mathbf{A} \mathbf{\Phi} t) \mathbf{\Phi}^{-L} \mathbf{P}(0)$$

Coarse Mesh: Multiple-species problems.

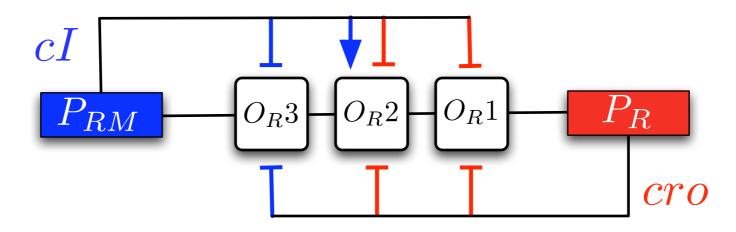
- I. Begin with original lattice.
- 2. Choose interpolation points.
- 3. Form interpolation (shape) function: $\mathbf{P}(t) \approx \mathbf{\Phi}\mathbf{q}(t)$
- 4. Project system to find reduced system of ODEs: $\dot{\mathbf{q}}(t) = \mathbf{\Phi}^{-L} \mathbf{A} \mathbf{\Phi} \mathbf{q}(t)$
- 5. Solve reduced system.
- 6. Lift back to original coordinates.



Outline

- Finite State Projection (FSP)
- **M** Reductions to the FSP
- Case Studies
 - ★ Lambda Phage.
 - ★ Heat Shock.

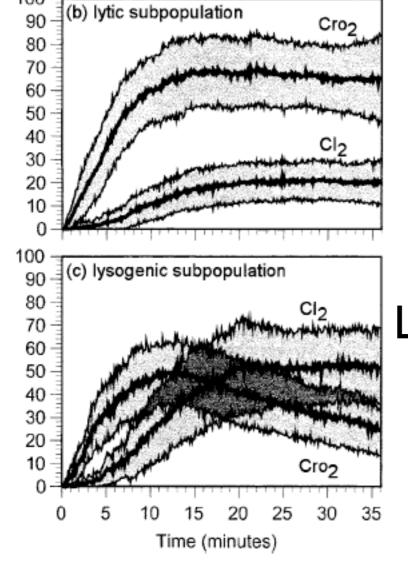
A toy model of phage lambda



- We consider only the core of the lambda switch.
- Two proteins, cI and cro.
- These activate and repress the P_R and P_{RM} promoters according to the model of Shea and Ackers, 1985.

The Phage Lambda Lysis-Lysogeny Decision

Arkin, Ross, McAdams, 1998. Full Model

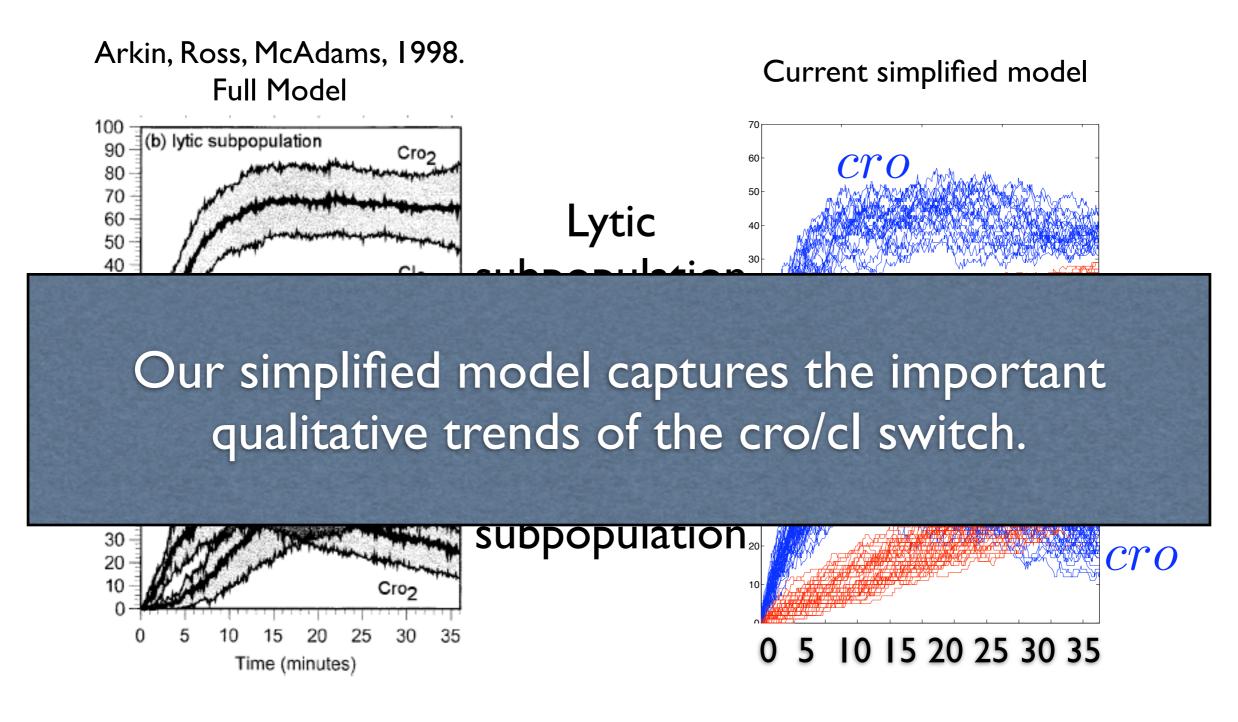


Lytic fate

- ★ Cro reaches a high level before CI is produced in much quantity.
- ★ Cro represses transcription of Cl.

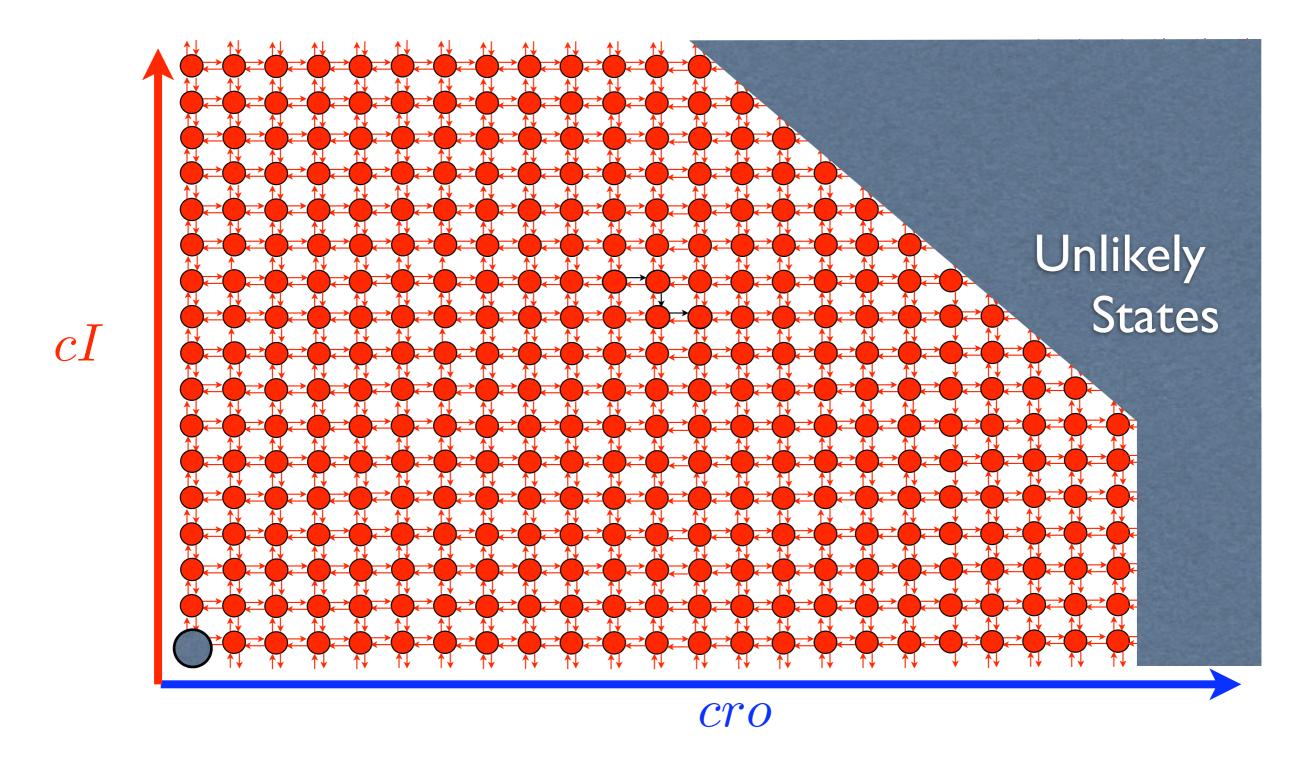
- Lysogenic fate
- ★ Cl increases a little earlier.
- ★ CI represses transcription of Cro.
- ★ CI is free to increase even further.

Relevance of Current Model

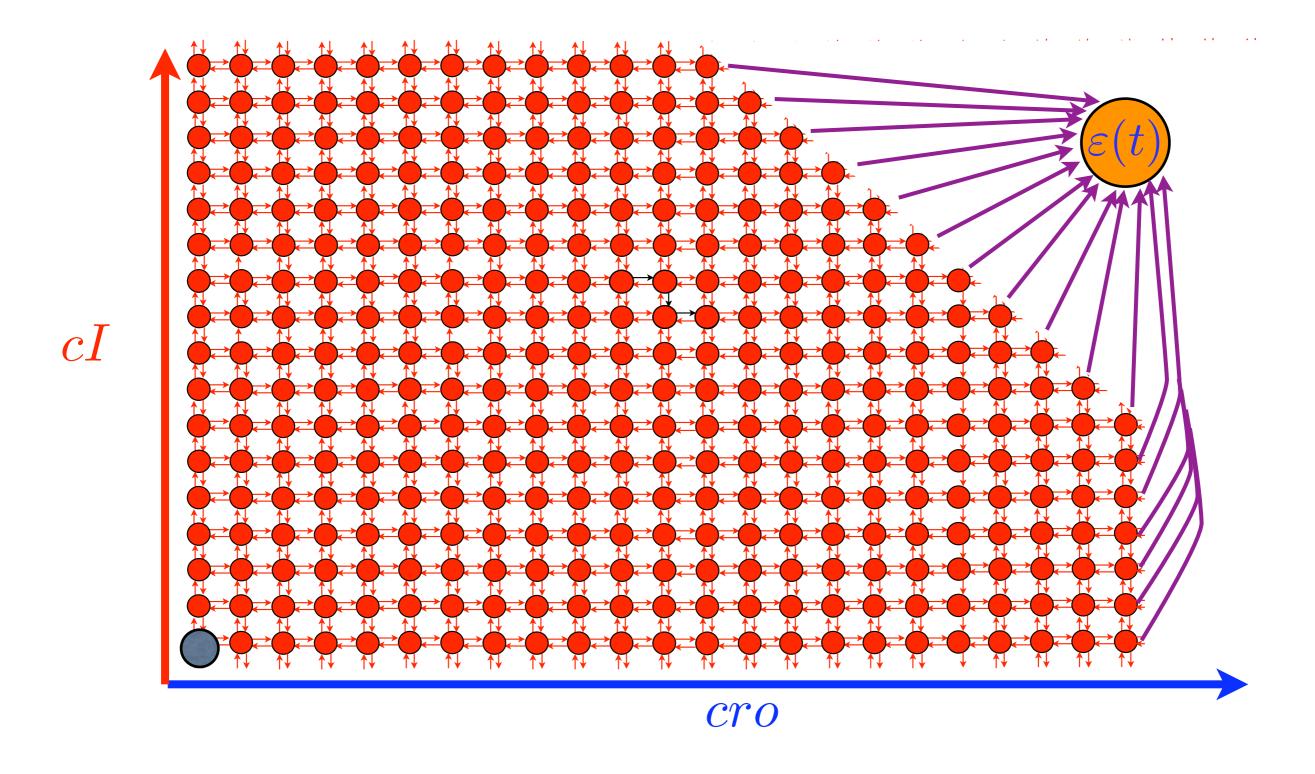


Computations done using Gillespie's SSA.

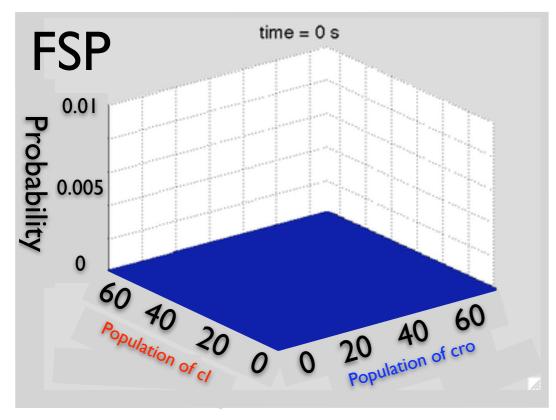
Applying the FSP to the Phage Lambda Switch

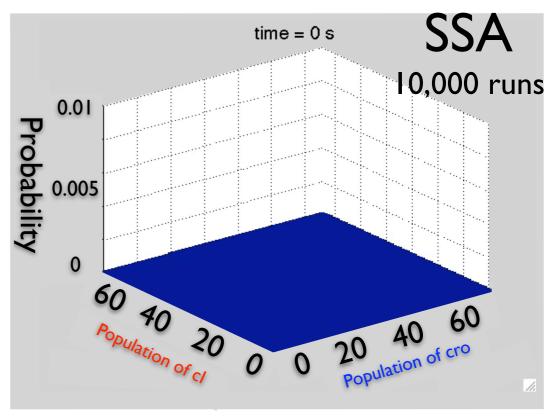


Applying the FSP to the Phage Lambda Switch



Efficiency and Accuracy of FSP Results





Method	# Simulations	Time (s)	$ig ig \mathbf{Error} ig _1$	
FSP	-a	163	$\leq 5.3 \times 10^{-3}$	Guaranteed

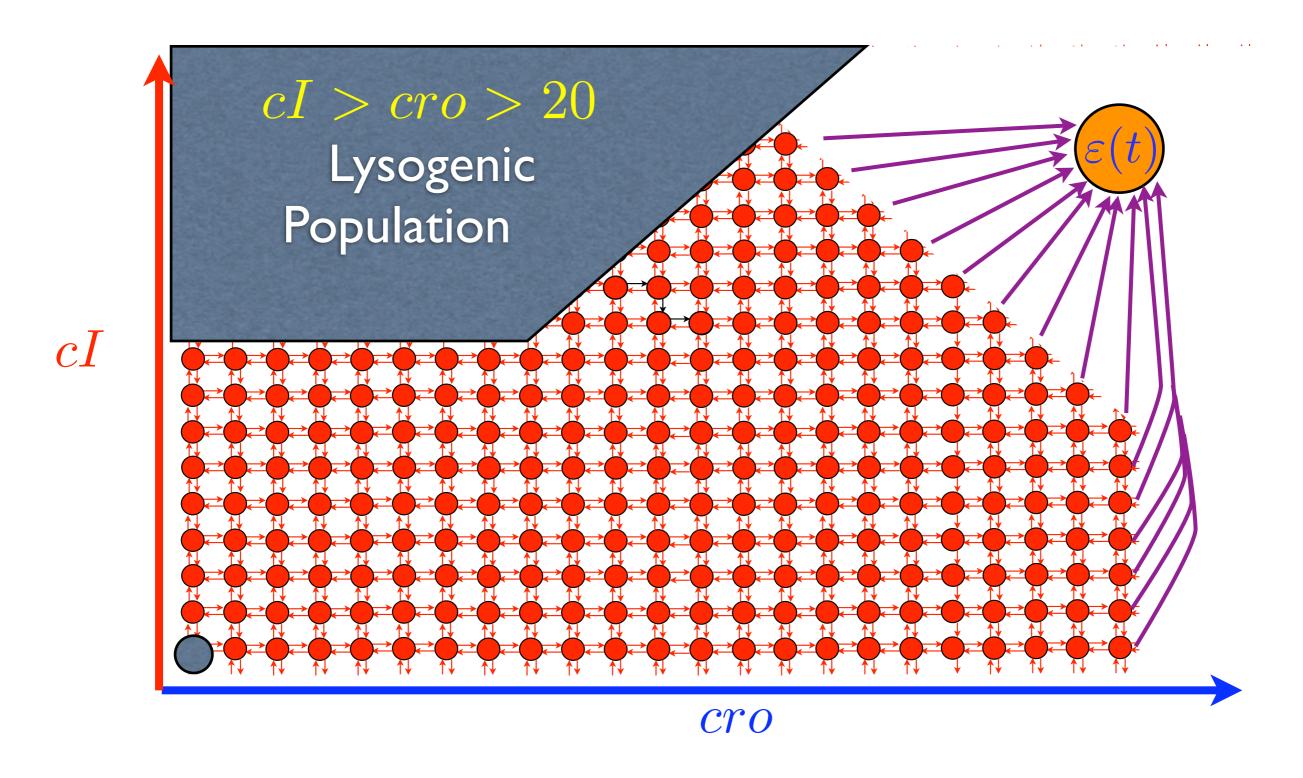
No Guarantees

^aThe FSP algorithm is run only once.

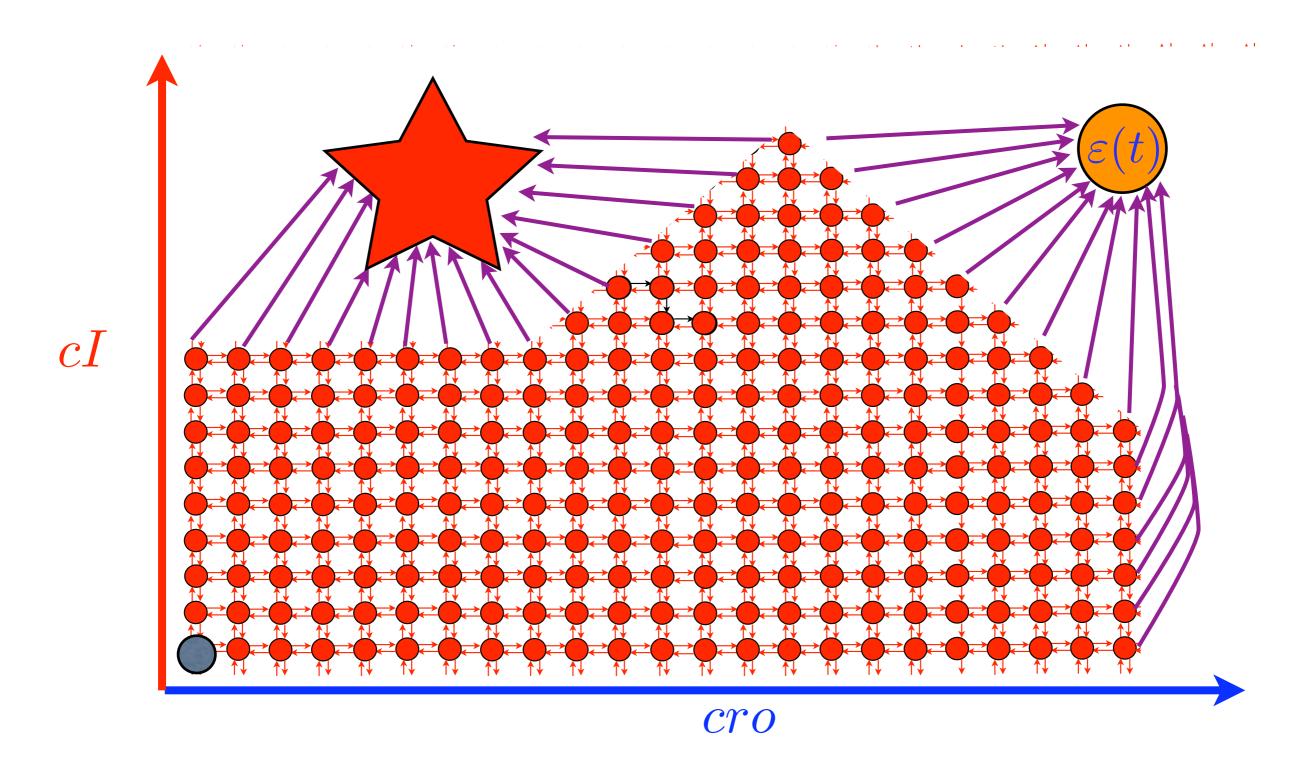
Additional information available with the FSP solution

- In many cases the FSP is faster and more accurate the Monte Carlo methods.
- Higher precision allows greater flexibility.
 - **★** Direct Computation of Switch Rates.

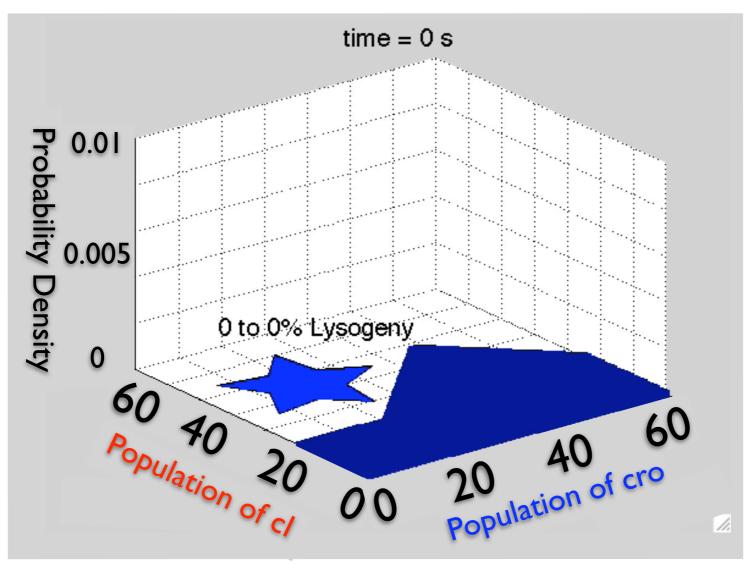
Using the FSP to Compute Switch Rates



Using the FSP to Compute Switch Rates



Using the FSP to Compute Switch Rates



Method	Time (s)	Relative Error	Guarantee?
FSP	$25.5 \mathrm{\ s}$	< 0.08 %	yes
10^4 SSA runs	440.0 s	$\approx 0.90 \%$	no

Additional information available with the FSP solution

- In many cases the FSP is faster and more accurate the Monte Carlo methods.
- Higher precision allows greater flexibility.
 - ★ Direct Computation of Switch Rates.
 - ★ Simultaneous consideration of many different initial conditions.

Comparing different initial conditions.

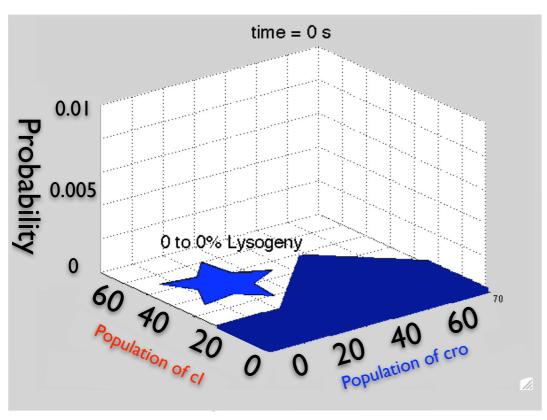
$$\mathcal{P}(t_0) \longrightarrow \tilde{\mathcal{P}}(t_0 + \tau)$$

- The FSP is an approximate map of distributions from one time to another.
- This map is valid for any initial distribution.
 - * Once computed, this map is cheap to apply again and again.
 - ★ The map automatically provides error bounds for any initial condition!

Comparing different initial conditions. (Increase in cro)

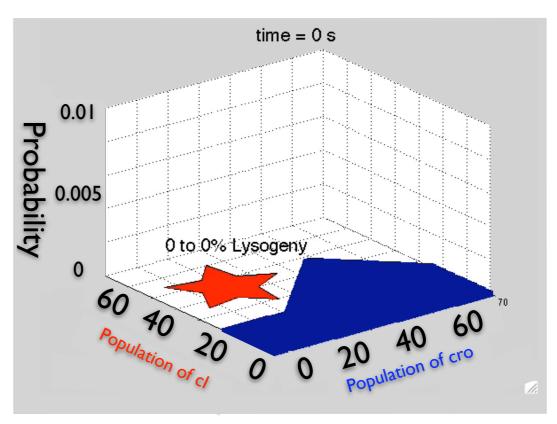
$$cI_0 = 0$$

$$cro_0 = 0$$



$$cI_0 = 0$$

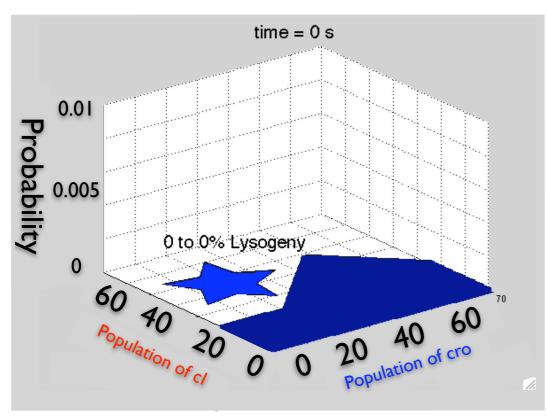
$$cro_0 = 5$$



Increasing the initial amount of cro yields a slight decrease in the lysogeny rate.

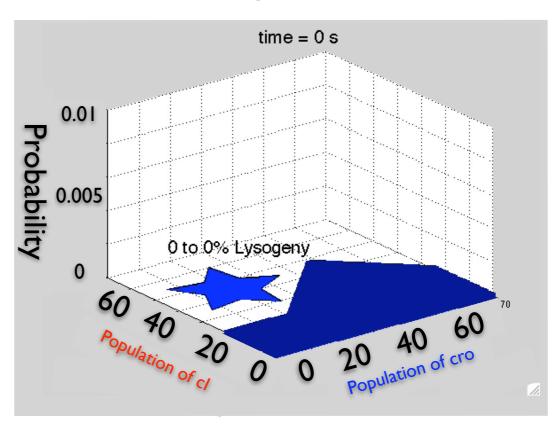
Comparing different initial conditions. (Increase in cI)

$$cI_0 = 0$$
$$cro_0 = 0$$



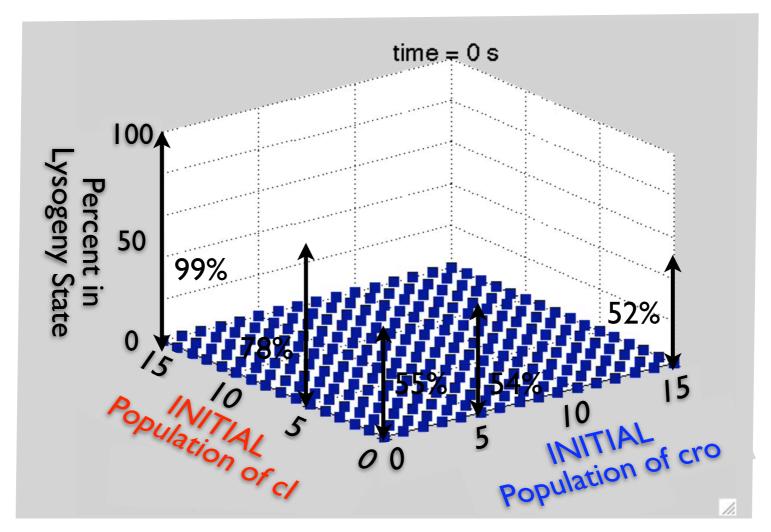
$$cI_0 = 5$$

$$cro_0 = 0$$



Increasing the initial amount of cI yields a significant increase in lysogeny rate.

Simultaneous comparison of an array of initial condition.)



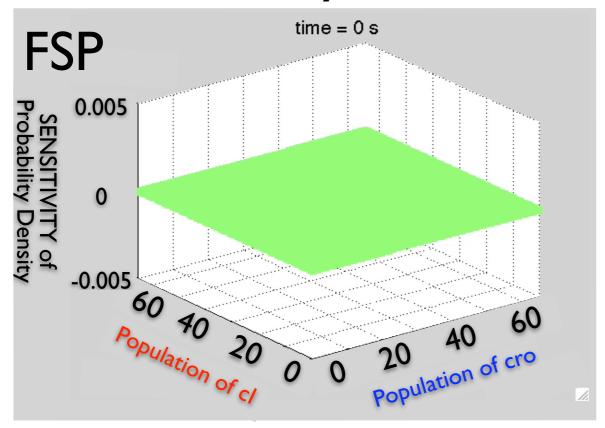
Method	Time (s)	# I.C.'s	$ig \ Error _1$	Guarantee?
FSP	66.9 s	2000	$<1\times10^{-4}$	yes
10^4 SSA runs	440.0 s	1	≈ 0.09	no
10^{13} SSA runs	$\approx 14,000 \text{ years!}$	2000	$\approx 1 \times 10^{-4}$	no

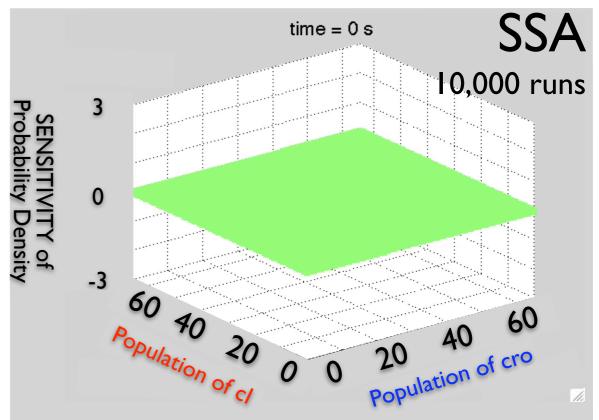
Additional information available with the FSP solution

- In many cases the FSP is both faster and more accurate than other available methods.
- Higher precision allows greater flexibility.
 - ★ Direct Computation of Switch Rates.
 - ★ Simultaneous consideration of many different initial conditions.
 - * Sensitivity to parameter changes.

Parametric Sensitivity of Probability Distributions.

Sensitivity to a small increase in cell Volume.





- ★ Sensitivity analysis requires a huge degree of accuracy.
- ★ Monte Carlo methods would require hundreds of millions of runs!!

Outline

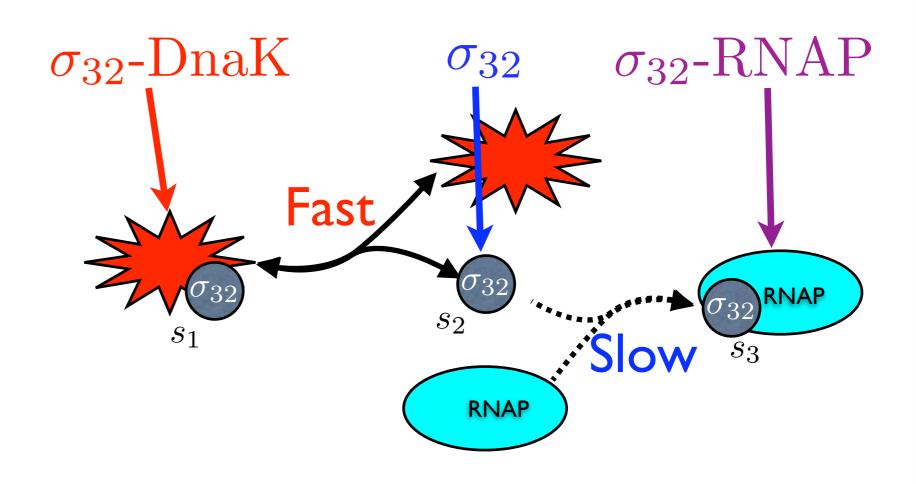
- Finite State Projection (FSP)
- **M** Reductions to the FSP
- Case Studies
 - ★ Lambda Phage.
 - ★ Heat Shock.

Toy Heat Shock Model in E. coli

3 forms for σ_{32} :

$$egin{array}{c} k_1 \ S_1 & \stackrel{}{\longleftarrow} S_2 \ k_2 \end{array}$$

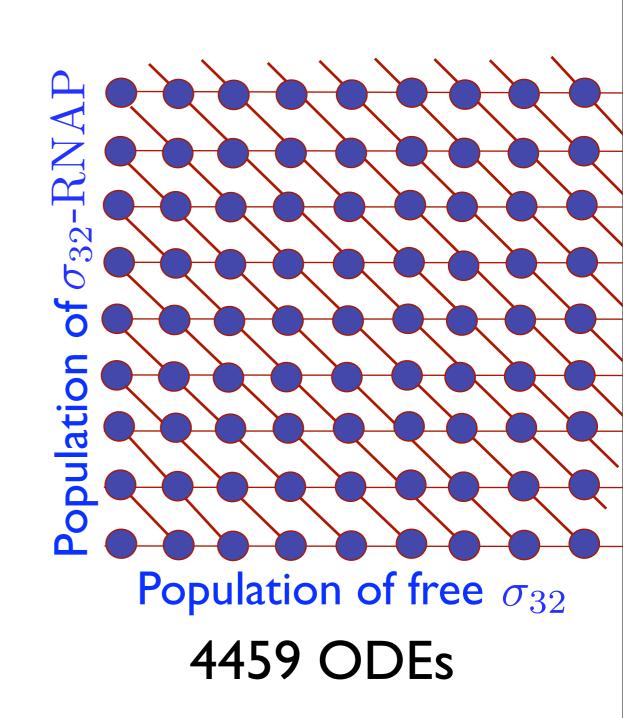
$$S_2 \stackrel{k_3}{\longrightarrow} S_3$$



El Samad et al, PNAS, vol. 102, No. 8, 2005

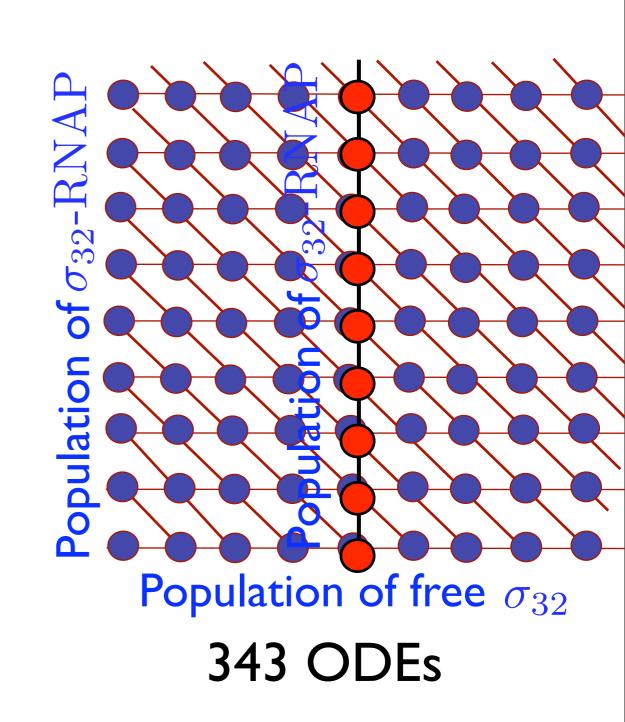
Five Different FSP Solution Schemes:

I. Full FSP



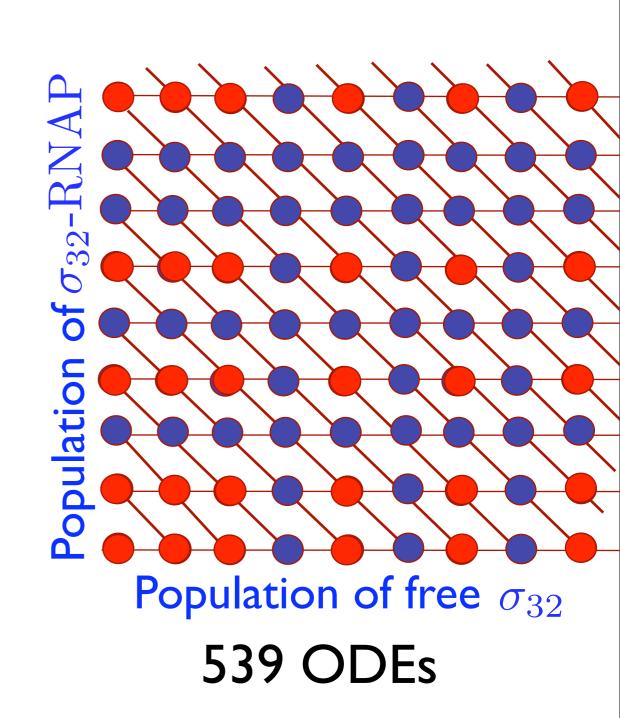
Five Different FSP Solution Schemes:

- I. Full FSP
- 2. Slow manifold (FSP-SM)



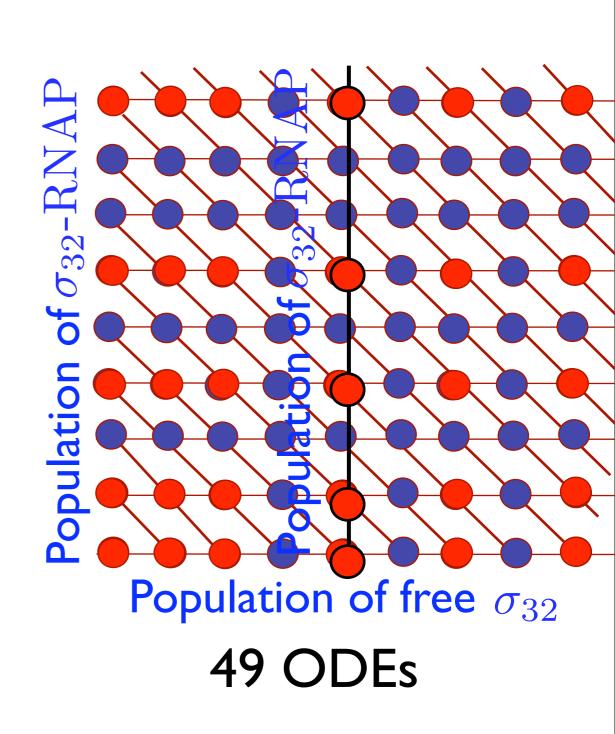
Five Different FSP Solution Schemes:

- I. Full FSP
- 2. Slow manifold (FSP-SM)
- 3. Interpolated (FSP-I)



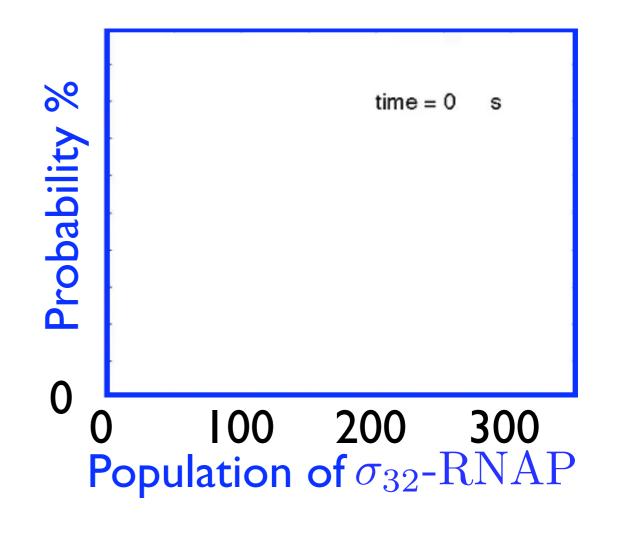
Five Different FSP Solution Schemes:

- I. Full FSP
- 2. Slow manifold (FSP-SM)
- 3. Interpolated (FSP-I)
- 4. Hybrid (FSP-SM/I)



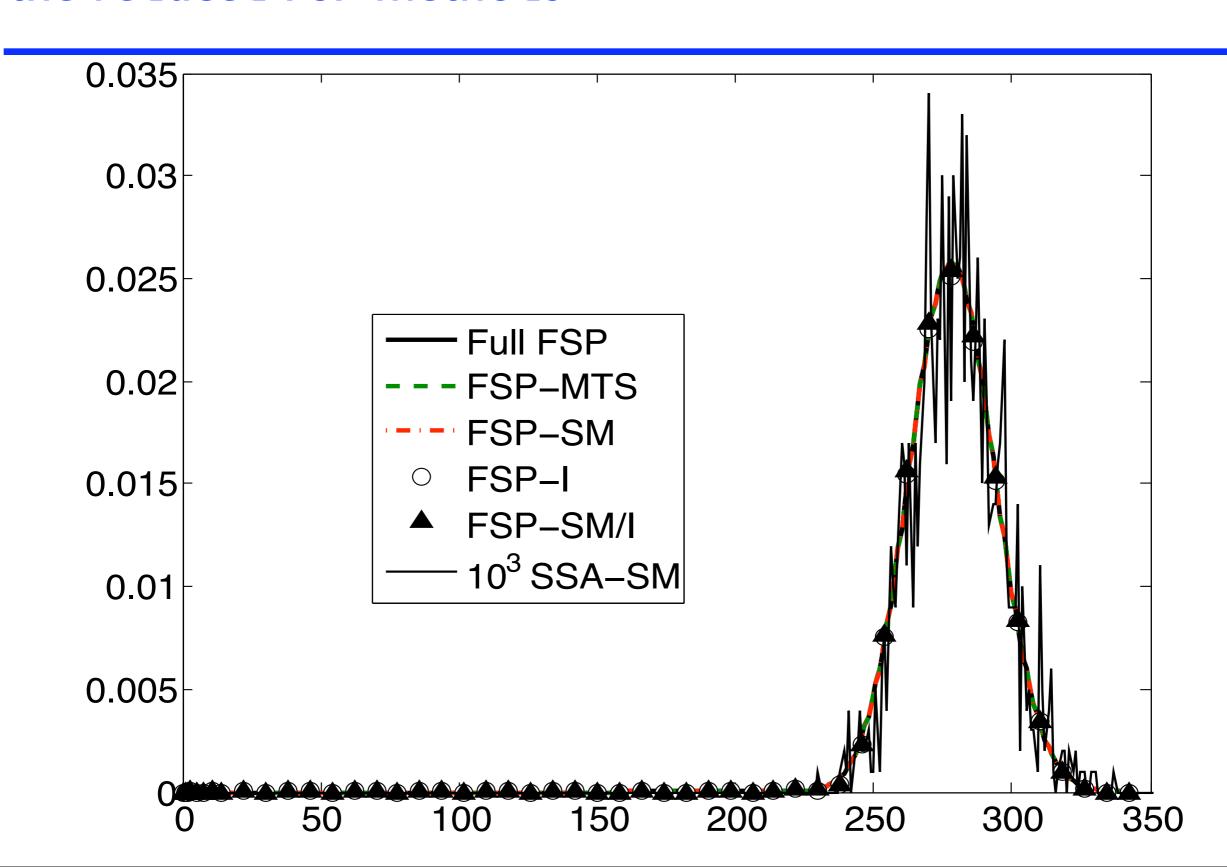
Five Different FSP Solution Schemes:

- I. Full FSP
- 2. Slow manifold (FSP-SM)
- 3. Interpolated (FSP-I)
- 4. Hybrid (FSP-SM/I)
- 5. Multiple time interval (FSP-MTI)



70 sets of 195 or fewer ODEs.

Efficiency and accuracy of the reduced FSP methods



Efficiency and accuracy of the reduced FSP methods

For final time $t_f = 300s$				
Method	Matrix Size	J_{solve}	J_{total}	∞-norm Error
FSP	4459	750s	750s	$< 3.0 \times 10^{-5}$
FSP-MTS	195^{1}	_	40.2s	$< 1.68 \times 10^{-4}$
FSP-SM	343	0.25s	0.94s	$\approx 5.1 \times 10^{-4}$
FSP-I	539	5.1s	6.1s	$\approx 7.7 \times 10^{-4}$
FSP-SM/I	49	0.04s	0.78s	$\approx 8.2 \times 10^{-4}$
10^4 SSA	Results would take more than 55 hours.			
10^3 SSA-SM	_	_	84.1s	≈ 0.0116
10^4 SSA-SM	_	_	925s	$\approx 3.4 \times 10^{-3}$
10^5 SSA-SM	_	_	9360s	$\approx 1.6 \times 10^{-3}$

The Reduced FSP approaches are much faster and more accurate than alternative approaches!

Conclusions

- Stochastic fluctuations or "noise" is present in the cell
 - Random nature of reactions
 - Quantization of reactants
 - Low copy numbers
- Fluctuations may be very important
 - Cell variability
 - Cell fate decisions
- Some tools are available
 - Monte Carlo simulations (SSA and variants)
 - Moment approximation methods
 - Linear noise approximation (Van Kampen)
 - Finite State Projection
- Many more are needed!

Conclusions

The Finite State Projection: a new tool for stochastic analysis of gene networks

Advantages:

- Accuracy: solutions with a guaranteed error bounds Particularly suitable for studying rare events
- Speed: solutions can be much faster than Monte Carlo methods especially when the system has large number of reactions/reaction firings
- Insight: Provides valuable information at little additional cost: Sensitivity/robustness to parameter changes
 Effect of changes in initial probabilities

Limitations

 Scalability: Not feasible when there are many species with broad distributions (over the time of interest [0, t])